## SECTION - A

1. $\int_{0}^{\infty} \frac{6}{\mathrm{e}^{3 \mathrm{x}}+6 \mathrm{e}^{2 \mathrm{x}}+11 \mathrm{e}^{\mathrm{x}}+6} \mathrm{dx}$
(1) $\log _{\mathrm{e}}\left(\frac{32}{27}\right)$
(2) $\log _{e}\left(\frac{256}{81}\right)$
(3) $\log _{\mathrm{e}}\left(\frac{512}{81}\right)$
(4) $\log _{e}\left(\frac{64}{27}\right)$

Sol. 1

$$
\begin{aligned}
& \mathrm{I}=\int_{0}^{\infty} \frac{6}{\left(e^{x}+1\right)\left(\mathrm{e}^{x}+2\right)\left(\mathrm{e}^{x}+3\right)} \mathrm{dx} \\
& \\
& =6 \int_{0}^{\infty}\left(\frac{\frac{1}{2}}{\mathrm{e}^{\mathrm{x}}+1}+\frac{-1}{\mathrm{e}^{x}+2}+\frac{\frac{1}{2}}{\mathrm{e}^{x}+3}\right) d x \\
& \\
& =3 \int_{0}^{\infty} \frac{\mathrm{e}^{-x}}{1+\mathrm{e}^{-\mathrm{x}}} \mathrm{dx}-6 \int_{0}^{\infty} \frac{\mathrm{e}^{-x} \mathrm{dx}}{1+2 \mathrm{e}^{-\mathrm{x}}}+3 \int_{0}^{\infty} \frac{\mathrm{e}^{-x}}{1+3 \mathrm{e}^{-x}} \mathrm{dx} \\
& =3\left[-\ln \left(1+\mathrm{e}^{-\mathrm{x}}\right)\right]_{0}^{\infty}+6 \frac{1}{2}\left[\ln \left(1+2 \mathrm{e}^{-\mathrm{x}}\right)\right]_{0}^{\infty} \\
& -\frac{3}{3}\left[\ln \left(1+3 \mathrm{e}^{-\mathrm{x}}\right)\right]_{0}^{\infty} \\
& \\
& =3 \ln 2-3 \ln 3+\ln 4 \\
& \\
& =3 \ln \frac{2}{3}+\ln 4 \\
& \\
& =\ln \frac{32}{27}
\end{aligned}
$$

2. Among
(S1) : $\lim _{n \rightarrow \infty} \frac{1}{n^{2}}(2+4+6+\ldots . .+2 n)=1$
(S2) : $\lim _{\mathrm{n} \rightarrow \infty} \frac{1}{\mathrm{n}^{16}}\left(1^{15}+2^{15}+3^{15}+\ldots . .+\mathrm{n}^{15}\right)=\frac{1}{16}$
(1) Only (S1) is true
(2) Both (S1) and (S2) are true
(3) Both (S1) and (S2) are false
(4) Only (S2) is true

## Sol. 2

$S_{1}: \lim _{n \rightarrow \infty} \frac{n(n+1)}{n^{2}}=1 \Rightarrow$ True
$S_{2}: \lim _{\mathrm{n} \rightarrow \infty} \frac{1}{\mathrm{n}^{16}}\left(\sum \mathrm{r}^{15}\right)=\lim _{\mathrm{n} \rightarrow \infty} \frac{1}{\mathrm{n}} \sum\left(\frac{\mathrm{r}}{\mathrm{n}}\right)^{15}$
$=\int_{0}^{1} \mathrm{x}^{15} \mathrm{dx}=\frac{1}{16} \Rightarrow$ True
3. The number of symmetric matrices of order 3 , with all the entries from the set $\{0,1,2,3,4,5,6,7,8,9\}$, is
(1) $10^{9}$
(2) $10^{6}$
(3) $9^{10}$
(4) $6^{10}$

Sol. 2
$A=\left[\begin{array}{lll}a & b & c \\ b & d & e \\ c & e & f\end{array}\right], a, b, c, d, e, f \in\{0,1,2, \ldots .9\}$, Number of matrices $=10^{6}$
4. Let $\vec{a}=\hat{i}+4 \hat{j}+2 \hat{k}, \vec{b}=3 \hat{i}-2 \hat{j}+7 \hat{k}$ and $\vec{c}=2 \hat{i}-\hat{j}+4 \hat{k}$. If a vector $\vec{d}$ satisfies $\vec{d} \times \vec{b}=\vec{c} \times \vec{b}$ and $\vec{d} . \vec{a}=24$, then $|\overrightarrow{\mathrm{d}}|^{2}$ is equal to -
(1) 323
(2) 423
(3) 413
(4) 313

Sol. 3

$$
\begin{aligned}
& \overrightarrow{\mathrm{d}} \times \overrightarrow{\mathrm{b}}=\overrightarrow{\mathrm{c}} \times \overrightarrow{\mathrm{b}} \\
& \Rightarrow(\overrightarrow{\mathrm{~d}}-\overrightarrow{\mathrm{c}}) \times \overrightarrow{\mathrm{b}}=0 \\
& \Rightarrow \overrightarrow{\mathrm{~d}}=\overrightarrow{\mathrm{c}}+\lambda \overrightarrow{\mathrm{b}}
\end{aligned}
$$

Also $\overrightarrow{\mathrm{d}} \overrightarrow{\mathrm{a}}=24$
$\Rightarrow(\overrightarrow{\mathrm{c}}+\lambda \overrightarrow{\mathrm{b}}) \cdot \overrightarrow{\mathrm{a}}=24$
$\lambda=\frac{24-\overrightarrow{\mathrm{a}} \cdot \overrightarrow{\mathrm{c}}}{\overrightarrow{\mathrm{b}} \cdot \overrightarrow{\mathrm{a}}}=\frac{24-6}{9}=2$
$\Rightarrow \overrightarrow{\mathrm{d}}=\overrightarrow{\mathrm{c}}+2(\overrightarrow{\mathrm{~b}})$
$=8 \hat{\mathbf{i}}-5 \hat{\mathrm{j}}+18 \hat{\mathrm{k}}$
$\Rightarrow|\overrightarrow{\mathrm{d}}|^{2}=64+25+324=413$
5. A coin is biased so that the head is 3 times as likely to occur as tail. This coin is tossed until a head or three tails occur. If X denotes the number of tosses of the coin, then the mean of X is-
(1) $\frac{21}{16}$
(2) $\frac{15}{16}$
(3) $\frac{81}{64}$
(4) $\frac{37}{16}$

Sol. 1
$\mathrm{P}(\mathrm{H})=\frac{3}{4}$
$\mathrm{P}(\mathrm{T})=\frac{1}{4}$

| $X$ | 1 | 2 | 3 |
| :--- | :--- | :--- | :--- |
| $P(X)$ | $\frac{3}{4}$ | $\frac{1}{4} \times \frac{3}{4}$ | $\left(\frac{1}{4}\right)^{3}+\left(\frac{1}{4}\right)^{2} \times \frac{3}{4}$ |

Mean $\bar{X}=\frac{3}{4}+\frac{3}{8}+3\left(\frac{1}{64}+\frac{3}{64}\right)$
$=\frac{3}{4}+\frac{3}{8}+\frac{3}{16}$
$=3\left(\frac{7}{16}\right)=\frac{21}{16}$
6. $\max _{0 \leq x \leq \pi}\left\{x-2 \sin x \cos x+\frac{1}{3} \sin 3 x\right\}=$
(1) 0
(2) $\pi$
(3) $\frac{5 \pi+2+3 \sqrt{3}}{6}$
(4) $\frac{\pi+2-3 \sqrt{3}}{6}$

Sol. 3
$f(x)=x-\sin 2 x+\frac{1}{3} \sin 3 x$
$\mathrm{f}^{\prime}(\mathrm{x})=1-2 \cos 2 \mathrm{x}+\cos 3 \mathrm{x}=0$
$\mathrm{x}=\frac{5 \pi}{6}, \frac{\pi}{6}$
$\therefore \mathrm{f}^{\prime \prime}(\mathrm{x})=4 \sin 2 \mathrm{x}-3 \sin 3 \mathrm{x}$
$\mathrm{f}^{\prime \prime}\left(\frac{5 \pi}{6}\right)<0$
$\Rightarrow\left(\frac{5 \pi}{6}\right)$ is point of maxima
$\mathrm{f}\left(\frac{5 \pi}{6}\right)=\frac{5 \pi}{6}+\frac{\sqrt{3}}{2}+\frac{1}{3}$
7. The set of all $\mathrm{a} \in \mathrm{R}$ for which the equation $\mathrm{x}|\mathrm{x}-1|+|\mathrm{x}+2|+\mathrm{a}=0$ has exactly one real root, is
(1) $(-\infty,-3)$
(2) $(-\infty, \infty)$
(3) $(-6, \infty)$
(4) $(-6,-3)$

## Sol. 2

$\mathrm{f}(\mathrm{x})=\mathrm{x}|\mathrm{x}-1|+|\mathrm{x}+2|$
$x|x-1|+|x+2|+a=0$
$\mathrm{x}|\mathrm{x}-1|+|\mathrm{x}+2|=-\mathrm{a}$


All values are increasing.
8. Let $P Q$ be a focal chord of the parabola $y^{2}=36 x$ of length 100 , making an acute angle with the positive $x$-axis. Let the ordinate of $P$ be positive and $M$ be the point on the line segment $P Q$ such that $P M: M Q=3: 1$. Then which of the following points does NOT lie on the line passing through M and perpendicular to the line PQ ?
(1) $(3,33)$
(2) $(6,29)$
(3) $(-6,45)$
(4) $(-3,43)$

## Sol. 4

$9\left(t+\frac{1}{t}\right)^{2}=100$
$\mathrm{t}=3$
$\Rightarrow \mathrm{P}(81,54) \& \mathrm{Q}(1,-6)$
M $(21,9)$
$\Rightarrow L$ is $(y-9)=\frac{-4}{3}(x-21)$
$3 y-27=-4 x+84$
$4 x+3 y=111$
9. For the system of linear equations
$2 x+4 y+2 a z=b$
$x+2 y+3 z=4$
$2 x-5 y+2 z=8$
which of the following is NOT correct?
(1) It has infinitely many solutions if $a=3, b=8$
(2) It has unique solution if $a=b=8$
(3) It has unique solution if $\mathrm{a}=\mathrm{b}=6$
(4) It has infinitely many solutions if $a=3, b=6$

## Sol. 4

$\Delta=\left|\begin{array}{ccc}2 & 4 & 2 a \\ 1 & 2 & 3 \\ 2 & -5 & 2\end{array}\right|=18(3-a)$
$\Delta_{\mathrm{x}}=\left|\begin{array}{ccc}\mathrm{b} & 4 & 2 \mathrm{a} \\ 4 & 2 & 3 \\ 8 & -5 & 2\end{array}\right|=(64+19 \mathrm{~b}-72 \mathrm{a})$
For unique solution $\Delta=0$
$\Rightarrow \mathrm{a} \neq 3$ and $\mathrm{b} \in \mathrm{R}$
For infinitely many solution :
$\Delta=\Delta_{x}=\Delta_{y}=\Delta_{z}=0$
$\Rightarrow a=3 \quad \because \Delta=0$
and $\mathrm{b}=8 \quad \because \Delta_{\mathrm{x}}=0$
10. Let $s_{1}, s_{2}, s_{3}, \ldots . . ., s_{10}$ respectively be the sum to 12 terms of 10 A.P. $s$ whose first terms are $1,2,3, \ldots, 10$ and the common differences are $1,3,5, \ldots \ldots \ldots ., 19$ respectively. Then $\sum_{\mathrm{i}=1}^{10} \mathrm{~s}_{\mathrm{i}}$ is equal to
(1) 7260
(2) 7380
(3) 7220
(4) 7360

Sol. 1
$\mathrm{S}_{\mathrm{k}}=6(2 \mathrm{k}+(11)(2 \mathrm{k}-1))$
$S_{k}=6(2 k+22 k-11)$
$\mathrm{S}_{\mathrm{k}}=144 \mathrm{k}-66$
$\sum_{1}^{10} S_{k}=144 \sum_{k=1}^{10} k-66 \times 10$
$=144 \times \frac{10 \times 11}{2}-660$
$=7920-660$
$=7260$
11. For the differentiable function $\mathrm{f}: \mathrm{R}-\{0\} \rightarrow \mathrm{R}$, let $3 \mathrm{f}(\mathrm{x})+2 \mathrm{f}\left(\frac{1}{\mathrm{x}}\right)=\frac{1}{\mathrm{x}}-10$, then $\left|\mathrm{f}(3)+\mathrm{f}^{\prime}\left(\frac{1}{4}\right)\right|$ is equal to
(1) 13
(2) $\frac{29}{5}$
(3) $\frac{33}{5}$
(4) 7

Sol. 1

$$
\begin{aligned}
& {\left[3 f(x)+2 f\left(\frac{1}{x}\right)=\frac{1}{x}-10\right] \times 3} \\
& {\left[2 f(x)+3 f\left(\frac{1}{x}\right)=x-10\right] \times 2}
\end{aligned}
$$

$5 f(x)=\frac{3}{x}-2 x-10$
$\mathrm{f}(\mathrm{x})=\frac{1}{5}\left(\frac{3}{\mathrm{x}}-2 \mathrm{x}-10\right)$
$\mathrm{f}^{\prime}(\mathrm{x})=\frac{1}{5}\left(-\frac{3}{\mathrm{x}^{2}}-2\right)$
$\left|\mathrm{f}(3)+\mathrm{f}^{\prime}\left(\frac{1}{4}\right)\right|=\left|\frac{1}{5}(1-6-10)+\frac{1}{5}(-48-2)\right|$
$=|-3-10|=13$
12. The negation of the statement $((\mathrm{A} \wedge(\mathrm{B} \vee \mathrm{C})) \Rightarrow(\mathrm{A} \vee \mathrm{B})) \Rightarrow \mathrm{A}$ is
(1) equivalent to $\mathrm{B} \vee \sim \mathrm{C}$
(2) a fallacy
(3) equivalent to $\sim \mathrm{C}$
(4) equivalent to $\sim \mathrm{A}$

Sol. 4
$\mathrm{p}:((\mathrm{A} \wedge(\mathrm{B} \vee \mathrm{C})) \Rightarrow(\mathrm{A} \vee \mathrm{B})) \Rightarrow \mathrm{A}$
$[\sim(A \wedge(B \vee C)) \vee(A \vee B)] \Rightarrow A$
$[(A \wedge(B \vee C)) \wedge \sim(A \vee B)] \vee A$
$(f \vee A)=A$
$\sim p \equiv \sim A$
13. Let the tangent and normal at the point $(3 \sqrt{3}, 1)$ on the ellipse $\frac{x^{2}}{36}+\frac{y^{2}}{4}=1$ meet the $y$-axis at the points $A$ and $B$ respectively. Let the circle C be drawn taking AB as a diameter and the line $\mathrm{x}=2 \sqrt{5}$ intersect C at the points $P$ and $Q$. If the tangents at the points $P$ and $Q$ on the circle intersect at the point $(\alpha, \beta)$, then $\alpha^{2}-\beta^{2}$ is equal to
(1) $\frac{304}{5}$
(2) 60
(3) $\frac{314}{5}$
(4) 61

Sol. 1
Given ellipse $\frac{x^{2}}{36}+\frac{y^{2}}{4}=1$
$\frac{x}{4 \sqrt{3}}+\frac{y}{4}=1$
$y=4$

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\(\frac{x}{4}-\frac{4}{4 \sqrt{3}}=\frac{2}{\sqrt{3}}\)
\(y=-8\)
\(x^{2}+y^{2}+4 y-32=0\)
\(h x+k y+2(y+k)-32=0\)
\(\mathrm{k}=-2\)
\(h \mathrm{x}+2 \mathrm{k}-32=0\)
\(\mathrm{hx}=36\)
\(\alpha=\mathrm{h}=\frac{36}{2 \sqrt{5}}\)
\(\beta=\mathrm{k}=-2\)
\(\alpha^{2}-\beta^{2}=\frac{304}{5}\)
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14. The distance of the point $(-1,2,3)$ from the plane $\overrightarrow{\mathrm{r}} .(\hat{\mathrm{i}}-2 \hat{\mathrm{j}}+3 \hat{\mathrm{k}})=10$ parallel to the line of the shortest distance between the lines $\overrightarrow{\mathrm{r}}=(\hat{\mathrm{i}}-\hat{\mathrm{j}})+\lambda(2 \hat{\mathrm{i}}+\hat{\mathrm{k}})$ and $\overrightarrow{\mathrm{r}}=(2 \hat{\mathrm{i}}-\hat{\mathrm{j}})+\mu(\hat{\mathrm{i}}-\hat{\mathrm{j}}+\hat{\mathrm{k}})$ is
(1) $2 \sqrt{5}$
(2) $3 \sqrt{5}$
(3) $3 \sqrt{6}$
(4) $2 \sqrt{6}$

Sol. 4
Let $L_{1}: \overrightarrow{\mathrm{r}}=(\hat{\mathrm{i}}-\hat{\mathrm{j}})+\lambda(2 \hat{\mathrm{i}}+\hat{\mathrm{k}})$
$L_{2}: \overrightarrow{\mathrm{r}}=(2 \hat{\mathrm{i}}-\hat{\mathrm{j}})+\mu(\hat{\mathrm{i}}-\hat{\mathrm{j}}+\hat{\mathrm{k}})$
$\overrightarrow{\mathrm{n}}=\left|\begin{array}{ccc}\hat{\mathrm{i}} & \hat{\mathrm{j}} & \hat{\mathrm{k}} \\ 2 & 0 & 1 \\ 1 & -1 & 1\end{array}\right|$
$\overrightarrow{\mathrm{n}}=\hat{\mathrm{i}}-\hat{\mathrm{j}}-2 \hat{\mathrm{k}}$
$\mathrm{P}(-1,2,3)$


Equation of line along shortest distance of $L_{1}$ and $L_{2}$

$$
\begin{aligned}
& \frac{x+1}{1}=\frac{y-2}{-1}=\frac{z-3}{-2}=r \\
& \Rightarrow(x, y, z) \equiv(r-1,2-r, 3-2 r) \\
& \Rightarrow(r-1)-2(2-r)+3(3-2 r)=10 \\
& \Rightarrow r=-2 \\
& \Rightarrow Q(x, y, z) \equiv(-3,4,7) \\
& \Rightarrow P Q=\sqrt{4+4+16}=2 \sqrt{6}
\end{aligned}
$$

15. Let $\mathrm{B}=\left[\begin{array}{lll}1 & 3 & \alpha \\ 1 & 2 & 3 \\ \alpha & \alpha & 4\end{array}\right], \alpha>2$ be the adjoint of a matrix A and $|\mathrm{A}|=2$. then
$\left[\begin{array}{lll}\alpha & -2 \alpha & \alpha\end{array}\right] \mathrm{B}\left[\begin{array}{c}\alpha \\ -2 \alpha \\ \alpha\end{array}\right]$ is equal to
(1) 16
(2) 32
(3) 0
(4) -16

Sol. 4
Given, $B=\left[\begin{array}{lll}1 & 3 & \alpha \\ 1 & 2 & 3 \\ \alpha & \alpha & 4\end{array}\right]$
$|B|=4$
$1(8-3 \alpha)-3(4-3 \alpha)+\alpha(\alpha-2 \alpha)=4$
$-\alpha^{2}+6 \alpha-8=0$
$\alpha=2,4$
Given $\alpha>2$
So, $\alpha=2$ is rejected
$\left[\begin{array}{lll}4 & -8 & 4\end{array}\right]\left[\begin{array}{lll}1 & 3 & 4 \\ 1 & 2 & 3 \\ 4 & 4 & 4\end{array}\right]\left[\begin{array}{c}4 \\ -8 \\ 4\end{array}\right]=[-16]_{\mid \times 1}$
16. For $x \in R$, two real valued functions $f(x)$ and $g(x)$ are such that, $g(x)=\sqrt{x}+1$ and $\operatorname{fog}(x)=x+3-\sqrt{x}$. Then $f(0)$ is equal to
(1) 5
(2) 0
(3) -3
(4) 1

Sol. 1
$g(x)=\sqrt{x}+1$
$f o g(x)=x+3-\sqrt{x}$
$=(\sqrt{\mathrm{x}}+1)^{2}-3(\sqrt{\mathrm{x}}+1)+5$
$=g^{2}(x)-3 g(x)+5$
$\Rightarrow \mathrm{f}(\mathrm{x})=\mathrm{x}^{2}-3 \mathrm{x}+5$
$\therefore \mathrm{f}(0)=5$
But, if we consider the domain of the composite function $\operatorname{fog}(x)$ then in that case $f(0)$ will be not defined as $\mathrm{g}(\mathrm{x})$ cannot be equal to zero.
17. Let the equation of plane passing through the line of intersection of the planes $x+2 y+a z=2$ and $x-y+z=3$ be $5 x$ $-11 y+b z=6 a-1$. For $c \in Z$, if the distance of this plane from the point $(a,-c, c)$ is $\frac{2}{\sqrt{a}}$, then $\frac{a+b}{c}$ is equal to
(1) -4
(2) 2
(3) -2
(4) 4

## Sol. 1

$(x+2 y+a z-2)+\lambda(x-y+z-3)=0$
$\frac{1+\lambda}{5}=\frac{2-\lambda}{-11}=\frac{a+\lambda}{b}=\frac{2+3 \lambda}{6 a-1}$
$\lambda=-\frac{7}{2}, a=3, b=1$
$\frac{2}{\sqrt{\mathrm{a}}}=\left|\frac{5 a+11 c+b c-6 a+1}{\sqrt{25+121+1}}\right|$
$\mathrm{c}=-1$
$\therefore \frac{\mathrm{a}+\mathrm{b}}{\mathrm{c}}=\frac{3+1}{-1}=-4$
18. Fractional part of the number is $\frac{4^{2022}}{15}$ equal to
(1) $\frac{4}{15}$
(2) $\frac{8}{15}$
(3) $\frac{1}{15}$
(4) $\frac{14}{15}$

Sol. 3
$\left\{\frac{4^{2022}}{15}\right\}=\left\{\frac{2^{4044}}{15}\right\}=\left\{\frac{(1+15)^{1011}}{15}\right\}=\frac{1}{15}$
19. Let $y=y_{1}(x)$ and $y=y_{2}(x)$ be the solution curves of the differential equation $\frac{d y}{d x}=y+7$ with initial conditions $y_{1}(0)=0$ and $y_{2}(0)=1$ respectively. Then the curves $y=y_{1}(x)$ and $y=y_{2}(x)$ intersect at
(1) no point
(2) infinite number of points
(3) one point
(4) two points

Sol. 1
$\frac{d y}{d x}=y+7 \Rightarrow \frac{d y}{d x}-y=7$
I.F. $=e^{-x}$
$y e^{-x}=\int 7 e^{-x} d x$
$\Rightarrow y^{-x}=-7 e^{-x}+c$
$\Rightarrow \mathrm{y}=-7+\mathrm{ce}^{\mathrm{x}}$
$-7+7 \mathrm{e}^{\mathrm{x}}=-7+8 \mathrm{e}^{\mathrm{x}} \Rightarrow \mathrm{e}^{\mathrm{x}}=0$.
No solution
20. The area of the region enclosed by the curve $f(x)=\max \{\sin x, \cos x\},-\pi \leq x \leq \pi$ and the $x$-axis is
(1) $2 \sqrt{2}(\sqrt{2}+1)$
(2) $4(\sqrt{2})$
(3) 4
(4) $2(\sqrt{2}+1)$

Sol. 3


Area $=$
$\left|\int_{-\pi}^{\frac{-3 \pi}{4}} \sin x d x\right|+\left|\int_{\frac{-3 \pi}{4}}^{\frac{-\pi}{2}} \cos x d x\right|+\int_{\frac{-\pi}{2}}^{\frac{\pi}{4}} \cos x d x+\int_{\frac{\pi}{4}}^{\pi} \sin x d x=4$

## SECTION - B

21. The sum to 20 terms of the series $2.2^{2}-3^{2}+2.4^{2}-5^{2}+2.6^{2}-$ $\qquad$ is equal to $\qquad$ -.
Sol. 1310

$$
\begin{aligned}
& \left(2^{2}-3^{2}+4^{2}-5^{2}+20 \text { terms }\right)+\left(2^{2}+4^{2}+\ldots .+10 \text { terms }\right) \\
& -(2+3+4+5+\ldots \ldots+11)+4\left[1+2^{2}+\ldots . .10^{2}\right] \\
& -\left[\frac{21 \times 22}{2}-1\right]+4 \times \frac{10 \times 11 \times 21}{6} \\
& =1-231+14 \times 11 \times 10 \\
& =1540+1-231 \\
& =1310
\end{aligned}
$$

22. Let the mean of the data

| x | 1 | 3 | 5 | 7 | 9 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Frequency (f) | 4 | 24 | 28 | $\alpha$ | 8 |

be 5 . If m and $\sigma^{2}$ are respectively the mean deviation about the mean and the variance of the data, then $\frac{3 \alpha}{\mathrm{~m}+\sigma^{2}}$ is equal to $\qquad$ -.
Sol. 8
$5=\bar{x}=\frac{\sum x_{i} f_{i}}{\sum f_{i}}=\frac{4+72+140+7 \alpha+72}{64+\alpha}$
$\Rightarrow 320+5 \alpha=288+7 \alpha \Rightarrow 2 \alpha=32 \Rightarrow \alpha=16$
M.D. $(\overline{\mathrm{x}})=\frac{\sum \mathrm{f}_{\mathrm{i}}\left|\mathrm{x}_{\mathrm{i}}-\overline{\mathrm{x}}\right|}{\sum \mathrm{f}_{\mathrm{i}}}$ where $\sum \mathrm{f}_{\mathrm{i}}=64+16=80$
M.D. $(\overline{\mathrm{x}})=\frac{4 \times 4+24 \times 2+28 \times 0+16 \times 2+8 \times 4}{80}=\frac{8}{5}$

Variance $=\frac{\sum \mathrm{f}_{\mathrm{i}}\left(\mathrm{x}_{\mathrm{i}}-\overline{\mathrm{x}}\right)^{2}}{\sum \mathrm{f}_{\mathrm{i}}}$
$=\frac{4 \times 16+24 \times 4+0+16 \times 4+8 \times 16}{80}=\frac{352}{80}$
$\therefore \frac{3 \alpha}{m+\sigma^{2}}=\frac{3 \times 16}{\frac{128}{80}+\frac{352}{80}}=8$
23. Let $\alpha$ be the constant term in the binomial expansion of $\left(\sqrt{x}-\frac{6}{x^{\frac{3}{2}}}\right)^{n}, n \leq 15$. If the sum of the coefficients of the remaining terms in the expansion is 649 and the coefficient of $\mathrm{x}^{-\mathrm{n}}$ is $\lambda \alpha$, then $\lambda$ is equal to $\qquad$ .
Sol. 36
$\mathrm{T}_{\mathrm{k}+1}={ }^{\mathrm{n}} \mathrm{C}_{\mathrm{k}}(\mathrm{x})^{\frac{\mathrm{n}-\mathrm{k}}{2}}(-6)^{\mathrm{k}}(\mathrm{x})^{\frac{-3}{2} \mathrm{k}}$
$\frac{\mathrm{n}-\mathrm{k}}{2}-\frac{3}{2} \mathrm{k}=0$
$\mathrm{n}-4 k=0$
$(-5)^{\mathrm{n}}-\left({ }_{n} C_{\frac{\mathrm{n}}{4}}(-6)^{\frac{\mathrm{n}}{4}}\right)=649$
By observation $(625+24=649)$, we get $\mathrm{n}=4$
$\because \mathrm{n}=4 \& \mathrm{k}=1$
Required is coefficient of
$x^{-4}$ is $\left(\sqrt{4}-\frac{6}{x^{\frac{3}{2}}}\right)^{4}$
${ }^{4} C_{1}(-6)^{3}$
By calculating we will get $\lambda=36$
24. Let $\omega=\mathrm{z} \overline{\mathrm{z}}+\mathrm{k}_{1} \mathrm{z}+\mathrm{k}_{2} \mathrm{iz}+\lambda(1+\mathrm{i}), \mathrm{k}_{1}, \mathrm{k}_{2} \in \mathrm{R}$. Let $\operatorname{Re}(\omega)=0$ be the circle C of radius 1 in the first quadrant touching the line $y=1$ and the $y$-axis. If the curve $\operatorname{Im}(\omega)=0$ intersects $C$ at $A$ and $B$, then $30(A B)^{2}$ is equal to

Sol. 24
$\omega=\mathrm{z} \overline{\mathrm{z}}+\mathrm{k}_{1} \mathrm{z}+\mathrm{k}_{2} \mathrm{iz}+\lambda(1+\mathrm{i})$
$\operatorname{Re}(\omega)=x^{2}+y^{2}+k_{1} x-k_{2} y+\lambda=0$
Centre $\equiv\left(\frac{-\mathrm{k}_{1}}{2}, \frac{\mathrm{k}_{2}}{2}\right) \equiv(1,2)$
$\Rightarrow \mathrm{k}_{1}=-2, \mathrm{k}_{2}=4$
radius $=1 \Rightarrow \lambda=4$
$\operatorname{Im}=\mathrm{k}_{1} \mathrm{y}+\mathrm{k}_{2} \mathrm{x}+\lambda=0$
$\therefore 2 \mathrm{x}-\mathrm{y}+2=0$
$\mathrm{d}=\frac{2}{\sqrt{5}}$
$\frac{1^{2}}{4}=1-\frac{4}{5}=\frac{1}{5}$
$\therefore 301^{2}=24$
25. Let $\vec{a}=3 \hat{i}+\hat{j}-\hat{k}$ and $\vec{c}=2 \hat{i}-3 \hat{j}+3 \hat{k}$. If $\vec{b}$ is a vector such that $\vec{a}=\vec{b} \times \vec{c}$ and $|\vec{b}|^{2}=50$, then $\left|72-|\vec{b}+\vec{c}|^{2}\right|$ is equal to $\qquad$ .
Sol. 66

$$
|\vec{a}|=\sqrt{11},|\vec{c}|=\sqrt{22}
$$

$|\overrightarrow{\mathrm{a}}|=|\overrightarrow{\mathrm{b}} \times \overrightarrow{\mathrm{c}}|=|\overrightarrow{\mathrm{b}}||\overrightarrow{\mathrm{c}}| \sin \theta$
$\sqrt{11}=\sqrt{50} \sqrt{22} \sin \theta$
$\Rightarrow \sin \theta=\frac{1}{10}$
$|\overrightarrow{\mathrm{b}}+\overrightarrow{\mathrm{c}}|^{2}=|\vec{b}|^{2}+|\overrightarrow{\mathrm{c}}|^{2}+2 \overrightarrow{\mathrm{~b}} \cdot \overrightarrow{\mathrm{c}}$
$=|\overrightarrow{\mathrm{b}}|^{2}+|\vec{c}|^{2}+2|\overrightarrow{\mathrm{~b}}||\vec{c}| \cos \theta$
$=50+22+2 \times \sqrt{50} \times \sqrt{22} \times \frac{\sqrt{99}}{10}$
$=72+66$
$\left|72-|\vec{b}+\vec{c}|^{2}\right|=66$
26. Let $m_{1}$, and $m_{2}$ be the slopes of the tangents drawn from the point $P(4,1)$ to the hyperbola $H: \frac{y^{2}}{25}-\frac{x^{2}}{16}=1$. If Q is the point from which the tangents drawn to H have slopes $\left|\mathrm{m}_{1}\right|$ and $\left|\mathrm{m}_{2}\right|$ and they make positive intercepts $\alpha$ and $\beta$ on the $x$-axis, then $\frac{(\mathrm{PQ})^{2}}{\alpha \beta}$ is equal to $\qquad$ .

## Sol. 8

Equation of tangent to the hyperbola $\frac{y^{2}}{a^{2}}-\frac{x^{2}}{b^{2}}=1$
$y=m x \pm \sqrt{a^{2}-b^{2} m^{2}}$
passing through $(4,1)$
$1=4 \mathrm{~m} \pm \sqrt{25-16 \mathrm{~m}^{2}} \Rightarrow 4 \mathrm{~m}^{2}-\mathrm{m}-3=0$
$\Rightarrow \mathrm{m}=1, \frac{-3}{4}$
Equation of tangent with positive slopes $1 \& \frac{3}{4}$
$\left.\begin{array}{c}4 y=3 x-16 \\ y=x-3\end{array}\right\}$ with positive intercept on $x$-axis.
$\alpha=\frac{16}{3}, \beta=3$
Intersection points:
$\mathrm{Q}:(-4,-7)$
P: $(4,1)$
$P Q^{2}=128$
$\frac{\mathrm{PQ}^{2}}{\alpha \beta}=\frac{128}{16}=8$
27. Let the image of the point $\left(\frac{5}{3}, \frac{5}{3}, \frac{8}{3}\right)$ in the plane $x-2 y+z-2=0$ be $P$. If the distance of the point $Q(6,-2$, $\alpha), \alpha>0$, from P is 13 , then $\alpha$ is equal to
Sol. 15
Image of point $\left(\frac{5}{3}, \frac{5}{3}, \frac{8}{3}\right)$

$$
\begin{gathered}
\frac{x-\frac{5}{3}}{1}=\frac{y-\frac{5}{3}}{-2}=\frac{z-\frac{8}{3}}{1}=\frac{-2\left(1 \times \frac{5}{3}+(-2) \times \frac{8}{3}+1 \times \frac{8}{3}-2\right)}{1^{2}+2^{2}+1^{2}} \\
=\frac{1}{3}
\end{gathered}
$$

$\therefore x=2, y=1, z=3$
$13^{2}=(6-2)^{2}+(-2-1)^{2}+(\alpha-3)^{2}$
$\Rightarrow(\alpha-3)^{2}=144 \Rightarrow \alpha=15(\because \alpha>0)$
28. Let for $x \in R, S_{0}(x)=x, S_{k}(x)=C_{k} x+k \int_{0}^{x} S_{k-1}(t) d t$ where $C_{0}=1, C_{k}=1-\int_{0}^{1} S_{k-1}(x) d x, k=1,2,3, \ldots$ Then $S_{2}(3)+6 C_{3}$ is equal to $\qquad$ .

Sol. 18
Given, $S_{k}(x)=C_{k} x+k \int_{0}^{x} S_{k-1}(t) d t$
put $\mathrm{k}=2$ and $\mathrm{x}=3$
$\mathrm{S}_{2}(3)=\mathrm{C}_{2}(3)+2 \int_{0}^{3} \mathrm{~S}_{1}(\mathrm{t}) \mathrm{dt}$
Also, $S_{1}(x)=C_{1}(x)+\int_{0}^{x} S_{0}(t) d t$
$=C_{1} x+\frac{x^{2}}{2}$
$S_{2}(3)=3 C_{2}+2 \int_{0}^{3}\left(C_{1} t+\frac{t^{2}}{2}\right) d t$
$=3 C_{2}+9 C_{1}+9$
Also,
$C_{1}=1-\int_{0}^{1} \mathrm{~S}_{0}(x) d x=\frac{1}{2}$

$$
\begin{aligned}
& C_{2}=1-\int_{0}^{1} S_{1}(x) d x=0 \\
& C_{3}=1-\int_{0}^{1} S_{2}(x) d x \\
& =1-\int_{0}^{1}\left(C_{2} x+C_{1} x^{2}+\frac{x^{3}}{3}\right) d x=\frac{3}{4} \\
& S_{2}(x)=C_{2} x+2 \int_{0}^{x} S_{1}(t) d t \\
& =C_{2} x+C_{1} x^{2}+\frac{x^{3}}{3} \\
& \Rightarrow \mathrm{~S}_{2}(3)+6 \mathrm{C}_{3}=6 \mathrm{C}_{3}+3 \mathrm{C}_{2}+9 \mathrm{C}_{1}+9=18
\end{aligned}
$$

29. If $S=\left\{x \in R: \sin ^{-1}\left(\frac{x+1}{\sqrt{x^{2}+2 x+2}}\right)-\sin ^{-1}\left(\frac{x}{\sqrt{x^{2}+1}}\right)=\frac{\pi}{4}\right\}$, then is equal to $\qquad$ .

Sol. 4

$$
\begin{aligned}
& \sin ^{-1}\left(\frac{(x+1)}{\sqrt{(x+1)^{2}+1}}\right)-\sin ^{-1}\left(\frac{x}{\sqrt{x^{2}+1}}\right)=\frac{\pi}{4} \\
& \because \frac{t}{\sqrt{t^{2}+1}} \in(-1,1) \\
& \sin ^{-1}\left(\frac{(\mathrm{x}+1)}{\sqrt{(\mathrm{x}+1)^{2}+1}}\right)=\sin ^{-1}\left(\frac{\mathrm{x}}{\sqrt{\mathrm{x}^{2}+1}}\right)+\frac{\pi}{4} \\
& \frac{(x+1)}{\sqrt{(x+1)^{2}+1}}=\left(\frac{1}{\sqrt{2}}\right) \cos \left(\sin ^{-1}\left(\frac{x}{\sqrt{x^{2}+1}}\right)\right)+\frac{1}{\sqrt{2}}\left(\frac{x}{\sqrt{x^{2}+1}}\right) \\
& =\frac{1}{\sqrt{2}}\left(\frac{1}{\sqrt{x^{2}+1}}+\frac{x}{\sqrt{x^{2}+1}}\right) \\
& \frac{(x+1)}{\sqrt{(x+1)^{2}+1}}=\frac{1}{\sqrt{2}}\left(\frac{1+x}{\sqrt{x^{2}+1}}\right)
\end{aligned}
$$

After solving this equation, we get

$$
\begin{gathered}
\mathrm{x}=-1 \text { or } \mathrm{x}=0 \\
S=\{-1,0\}
\end{gathered}
$$

$$
\begin{aligned}
& \sum_{x \in R}\left(\sin \left(\left(x^{2}+x+5\right) \frac{\pi}{2}\right)-\cos \left(\left(x^{2}+x+5\right) \pi\right)\right) \\
& =\left[\sin \left(\frac{5 \pi}{2}\right)-\cos (5 \pi)\right]+\left[\sin \left(\frac{5 \pi}{2}\right)-\cos (5 \pi)\right]=4
\end{aligned}
$$

30. The number of seven digit positive integers formed using the digits $1,2,3$ and 4 only and sum of the digits equal to 12 is $\qquad$ -.
Sol. 413
$x_{1}+x_{2}+x_{3}+x_{4}+x_{5}+x_{6}+x_{7}=12, x_{i} \in\{1,2,3,4\}$
No. of solutions $={ }^{5+7-1} C_{7-1}-\frac{7!}{6!}-\frac{7!}{5!}=413$

## SECTION - A

31. Different combination of 3 resistors of equal resistance $R$ are shown in the figures. The increasing order for
power dissipation is:
(A)

(C)

(B)

(D) $I$

(1) $\mathrm{P}_{\mathrm{C}}<\mathrm{P}_{\mathrm{B}}<\mathrm{P}_{\mathrm{A}}<\mathrm{P}_{\mathrm{D}}$
(2) $\mathrm{P}_{\mathrm{C}}<\mathrm{P}_{\mathrm{D}}<\mathrm{P}_{\mathrm{A}}<\mathrm{P}_{\mathrm{B}}$
(3) $\mathrm{P}_{\mathrm{B}}<\mathrm{P}_{\mathrm{C}}<\mathrm{P}_{\mathrm{D}}<\mathrm{P}_{\mathrm{A}}$
(4) $\mathrm{P}_{\mathrm{A}}<\mathrm{P}_{\mathrm{B}}<\mathrm{P}_{\mathrm{C}}<\mathrm{P}_{\mathrm{D}}$

Sol. (1)
Power dissipation, $\mathrm{P}=\mathrm{I}^{2} \mathrm{R}$
(A) $\mathrm{R}_{\mathrm{eq}}=\frac{\mathrm{R}}{2}+\mathrm{R}=\frac{3 \mathrm{R}}{2}$
(B) $\mathrm{R}_{\mathrm{eq}}=\frac{(2 \mathrm{R})(\mathrm{R})}{2 \mathrm{R}+\mathrm{R}}=\frac{2 \mathrm{R}}{3}$
(C) $\mathrm{R}_{\mathrm{eq}}=\frac{\mathrm{R}}{3}$
(D) $\mathrm{R}_{\mathrm{eq}}=3 \mathrm{R}$
$R_{D}>R_{A}>R_{B}>R_{C}$
Since, $\mathrm{P} \propto \mathrm{R}_{\text {eq }}$
$\mathrm{P}_{\mathrm{D}}>\mathrm{P}_{\mathrm{A}}>\mathrm{P}_{\mathrm{B}}>\mathrm{P}_{\mathrm{C}}$
32. For the following circuit and given inputs $A$ and $B$, chose the correct option for output ' $\mathrm{Y}^{\prime}$

(1)

(2)

(3)

1 (4)


Sol. (3)


Output, $y=\overline{\overline{\mathrm{A}} \cdot \mathrm{B}}=\overline{\overline{\mathrm{A}}}+\mathrm{B}$

$$
\mathrm{y}=\mathrm{A}+\overline{\mathrm{B}}
$$

$t_{1}$ to $t_{2}, \quad A=0, B=1, Y=0$
$\mathrm{t}_{2}$ to $\mathrm{t}_{3} \quad \mathrm{~A}=1, \mathrm{~B}=1, \mathrm{Y}=1$
$\mathrm{t}_{3}$ to $\mathrm{t}_{4} \quad \mathrm{~A}=0, \mathrm{~B}=0, \mathrm{Y}=1$
$\mathrm{t}_{4}$ to $\mathrm{t}_{5}, \quad \mathrm{~A}=1, \mathrm{~B}=1, \mathrm{Y}=1$
$\mathrm{t}_{5}$ to $\mathrm{t}_{6}, \quad \mathrm{~A}=1, \mathrm{~B}=0, \mathrm{Y}=1$
After $\mathrm{t}_{6}, \quad \mathrm{~A}=0, \mathrm{~B}=0, \mathrm{Y}=1$
33. A bullet of 10 g leaves the barrel of gun with a velocity of $600 \mathrm{~m} / \mathrm{s}$. If the barrel of gun is 50 cm long and mass of gun is 3 kg , then value of impulse supplied to the gun will be:
(1) 12 Ns
(2) 6 Ns
(3) 3 Ns
(4) 36 Ns

## Sol. (2)

Impulse, $|\overrightarrow{\mathrm{I}}|=|\Delta \overrightarrow{\mathrm{p}}|$

$$
\begin{aligned}
& =\mathrm{mV}-0 \\
& =\left(10 \times 10^{-3} \mathrm{~kg}\right)(600 \mathrm{~m} / \mathrm{s}) \\
\mathrm{I}= & 6 \mathrm{~N}-\mathrm{S}
\end{aligned}
$$

34. Which of the following Maxwell's equation is valid for time varying conditions but not valid for static conditions:
(1) $\oint \vec{D} \cdot \overrightarrow{d A}=Q$
(2) $\oint \overrightarrow{\mathrm{E}} \cdot \overrightarrow{\mathrm{dl}}=-\frac{\partial \phi_{\mathrm{B}}}{\partial \mathrm{t}}$
(3) $\oint \overrightarrow{\mathrm{E}} \cdot \overrightarrow{\mathrm{dl}}=0$
(4) $\oint \overrightarrow{\mathrm{B}} \cdot \overrightarrow{\mathrm{dl}}=\mu_{0} \mathrm{I}$

Sol. (2)
For static conditions

$$
\oint \overrightarrow{\mathrm{E}} \cdot \mathrm{~d} \overrightarrow{\mathrm{l}}=0
$$

For time varying condition,

$$
\oint \overrightarrow{\mathrm{E}} . \mathrm{d} \overrightarrow{\mathrm{l}}=-\frac{\partial \phi_{\mathrm{B}}}{\partial \mathrm{t}}
$$

35. Match List - I with List - II

| List - I <br> (Layer of atmosphere) | List - II <br> (Approximate height over earth's surface) |
| :--- | :--- |
| (A) F1 - Layer | (I) 10 km |
| (B) D - Layer | (II) $170-190 \mathrm{~km}$ |
| (C) Troposphere | (III) 100 km |
| (D) E - layer | (IV) $65-75 \mathrm{~km}$ |

Choose the correct answer from the options given below:
(1) A - II, B - I, C - IV, D - III
(2) A - II, B - IV, C - III, D - I
(3) A - II, B - IV, C - I, D - III
(4) A - III, B - IV, C - I, D - II

Sol. (3)
$\mathrm{F}_{1} \rightarrow$ Lower part of F layer of ionosphere $(170-190 \mathrm{Km})$
$\mathrm{D} \rightarrow$ Lowest layer of ionosphere $(65-75 \mathrm{Km})$
Troposphere $\rightarrow$ Lowest layer of atmosphere ( 10 Km )
$\mathrm{E} \rightarrow$ Middle part of ionosphere ( 100 Km )
36. The rms speed of oxygen molecule in a vessel at particular temperature is $\left(1+\frac{5}{x}\right)^{\frac{1}{2}} v$, where $v$ is the average speed of the molecule. The value of x will be: (Take $\pi=\frac{22}{7}$ )
(1) 28
(2) 27
(3) 8
(4) 4

Sol. (1)
$\mathrm{V}_{\mathrm{rms}}=\sqrt{\frac{3 R T}{\mathrm{M}}}$
$\mathrm{V}_{\mathrm{avg}}=v=\sqrt{\frac{8 \mathrm{RT}}{\pi \mathrm{M}}}$
$\mathrm{V}_{\mathrm{rms}}=\sqrt{\frac{3 \pi}{8}} v$
$\mathrm{V}_{\mathrm{rms}}=\sqrt{\frac{3}{8} \times \frac{22}{7}} v=\left(\frac{33}{28}\right)^{1 / 2} v$
$\mathrm{V}_{\mathrm{rms}}=\left(1+\frac{5}{28}\right)^{1 / 2} v$
$\mathrm{x}=28$
37. The ratio of powers of two motors is $\frac{3 \sqrt{x}}{\sqrt{x}+1}$, that are capable of raising 300 kg water in 5 minutes and 50 kg water in 2 minutes respectively from a well of 100 m deep. The value of $x$ will be
(1) 16
(2) 2
(3) 4
(4) 2.4

Sol. (1)
$\mathrm{P}=\frac{\text { Work }}{\text { Time }}$
$\mathrm{P}_{1}=\frac{\mathrm{mgh}}{\mathrm{t}_{1}}=\frac{(300) \mathrm{g}(100)}{5}$
$P_{2}=\frac{(50) g(100)}{2}$
$\frac{\mathrm{P}_{1}}{\mathrm{P}_{2}}=\frac{600}{250}=\frac{12}{5}=\frac{3 \times 4}{4+1}$
$\frac{\mathrm{P}_{1}}{\mathrm{P}_{2}}=\frac{3 \sqrt{16}}{\sqrt{16}+1}$
$\mathrm{x}=16$
38. Two trains 'A' and 'B' of length ' $l$ ' and ' $4 l$ ' are travelling into a tunnel of length 'L' in parallel tracks from opposite directions with velocities $108 \mathrm{~km} / \mathrm{h}$ and $72 \mathrm{~km} / \mathrm{h}$, respectively. If train 'A' takes 35 s less time than train 'B' to cross the tunnel then, length 'L' of tunnel is:
(Given $\mathrm{L}=60 l$ )
(1) 2700 m
(2) 1800 m
(3) 1200 m
(4) 900 m

Sol. (2)

L

L

$$
\begin{aligned}
& \mathrm{V}_{\mathrm{A}}=108 \times \frac{5}{18}=30 \mathrm{~m} / \mathrm{s} \\
& \mathrm{~V}_{\mathrm{B}}=72 \times \frac{5}{18}=20 \mathrm{~m} / \mathrm{s} \\
& \mathrm{~T}_{\mathrm{A}}=\frac{\ell+\mathrm{L}}{30}, \mathrm{~T}_{\mathrm{B}}=\frac{4 \ell+\mathrm{L}}{20} \\
& \mathrm{~T}_{\mathrm{A}}=\mathrm{T}_{\mathrm{B}}-35 \\
& \frac{\ell+\mathrm{L}}{30}=\frac{4 \ell+\mathrm{L}}{20}-35
\end{aligned}
$$

Given, $L=60 \ell$

$$
\begin{aligned}
& \frac{61 \ell}{30}=\frac{64 \ell}{20}-35 \\
& \frac{192 \ell-122 \ell}{60}=35
\end{aligned}
$$

$$
70 \ell=60 \times 35
$$

$$
\ell=30 \mathrm{~m}
$$

$$
\mathrm{L}=60 \ell=1800 \mathrm{~m}
$$

39. Two bodies are having kinetic energies in the ratio $16: 9$. If they have same linear momentum, the ratio of their masses respectively is:
(1) $16: 9$
(2) $4: 3$
(3) $9: 16$
(4) $3: 4$

## Sol. (3)

Kinetic energy, $K E=\frac{P^{2}}{2 m}$

$$
\begin{aligned}
& \frac{\mathrm{k}_{1}}{\mathrm{k}_{2}}=\frac{\mathrm{m}_{2}}{\mathrm{~m}_{1}} \\
& \frac{16}{9}=\frac{\mathrm{m}_{2}}{\mathrm{~m}_{1}} \\
& \frac{\mathrm{~m}_{1}}{\mathrm{~m}_{2}}=\frac{9}{16}
\end{aligned}
$$

40. The figure shows a liquid of given density flowing steadily in horizontal tube of varying cross - section. Cross sectional areas at $A$ is $1.5 \mathrm{~cm}^{2}$, and $B$ is $25 \mathrm{~mm}^{2}$, if the speed of liquid at $B$ is $60 \mathrm{~cm} / \mathrm{s}$ then $\left(P_{A}-P_{B}\right)$ is:
(Given $P_{A}$ and $P_{B}$ are liquid pressures at $A$ and $B$ points)
Density $\rho=1000 \mathrm{~kg} \mathrm{~m}^{-3}$
$A$ and $B$ are on the axis of tube

(1) 175 Pa
(2) 36 Pa
(3) 27 Pa
(4) 135 Pa

## Sol. (1)

By equation of continuity,

$$
\begin{aligned}
& \mathrm{A}_{1} \mathrm{~V}_{1}=\mathrm{A}_{2} \mathrm{~V}_{2} \\
& \left(1.5 \times 10^{-4}\right) \mathrm{V}_{\mathrm{A}}=\left(25 \times 10^{-6}\right) 60 \mathrm{~cm} / \mathrm{s} \\
& \mathrm{~V}_{\mathrm{A}}=10 \mathrm{~cm} / \mathrm{s}
\end{aligned}
$$

By Bernoulli's theorem,

$$
\begin{aligned}
& P_{A}+\frac{1}{2} \rho V_{A}^{2}=P_{B}+\frac{1}{2} \rho V_{B}^{2} \\
& P_{A}-P_{B}=\frac{\rho}{2}\left(V_{B}^{2}-V_{A}^{2}\right) \\
& P_{A}-P_{B}=\frac{1000}{2}\left(60^{2}-10^{2}\right) \times 10^{-4} \\
& P_{A}-P_{B}=175 \mathrm{~Pa}
\end{aligned}
$$

41. ${ }_{92}^{238} \mathrm{~A} \rightarrow{ }_{90}^{234} \mathrm{~B}+{ }_{2}^{4} \mathrm{D}+\mathrm{Q}$

In the given nuclear reaction, the approximate amount of energy released will be:
[Given, mass of ${ }_{92}^{238} \mathrm{~A}=238.05079 \times 931.5 \mathrm{MeV} / \mathrm{c}^{2}$,

$$
\begin{aligned}
& \text { mass of }{ }_{90}^{234} \mathrm{~B}=234.04363 \times 931.5 \mathrm{MeV} / \mathrm{c}^{2} \\
& \text { mass of } \left.{ }_{2}^{4} \mathrm{D}=4.00260 \times 931.5 \mathrm{MeV} / \mathrm{c}^{2}\right]
\end{aligned}
$$

(1) 4.25 MeV
(2) 5.9 MeV
(3) 3.82 MeV
(4) 2.12 MeV

Sol. (1)
$\mathrm{Q}=\Delta \mathrm{mC} \mathrm{C}^{2}$
$\mathrm{Q}=(238.05079-234.04363-4.00260) \times 931.5 \mathrm{MeV}$
$\mathrm{Q}=0.00456 \times 931.5 \mathrm{MeV}$
$\mathrm{Q}=4.25 \mathrm{MeV}$
42. A disc is rolling without slipping on a surface. The radius of the disc is R . At $\mathrm{t}=0$, the top most point on the disc is A as shown in figure. When the disc completes half of its rotation, the displacement of point A from its initial position is

(1) $2 R \sqrt{\left(1+4 \pi^{2}\right)}$
(2) $R \sqrt{\left(\pi^{2}+4\right)}$
(3) 2 R
(4) $\mathrm{R} \sqrt{\left(\pi^{2}+1\right)}$

Sol. (2)


Displacement $=A^{\prime} \mathrm{A}$

$$
\begin{aligned}
& \mathrm{A}^{\prime} \mathrm{A}=\sqrt{(\pi \mathrm{R})^{2}+(2 \mathrm{R})^{2}} \\
& \mathrm{~A}^{\prime} \mathrm{A}=\mathrm{R} \sqrt{\pi^{2}+4}
\end{aligned}
$$

43. Which graph represents the difference between total energy and potential energy of a particle executing SHM vs it's distance from mean position?
(1)

(2)

(3)

(4)


Sol. (2)
Total energy in SHM = E
$E=K+U$
$E-U=K$
$\mathrm{E}-\mathrm{U}=\mathrm{K}$
$\mathrm{E}-\mathrm{U}=\frac{1}{2} \mathrm{~m} \omega^{2}\left(\mathrm{~A}^{2}-\mathrm{x}^{2}\right)$

44. Two charges each of magnitude 0.01 C and separated by a distance of 0.4 mm constitute an electric dipole. If the dipole is placed in an uniform electric field ' $\vec{E}$ ' of 10 dyne/C making $30^{\circ}$ angle with $\vec{E}$, the magnitude of torque acting on dipole is
(1) $1.5 \times 10^{-9} \mathrm{Nm}$
(2) $2.0 \times 10^{-10} \mathrm{Nm}$
(3) $1.0 \times 10^{-8} \mathrm{Nm}$
(4) $4.0 \times 10^{-10} \mathrm{Nm}$

## Sol. (2)

Dipole moment, $\mathrm{P}=\mathrm{qd}$

$$
\begin{aligned}
& \mathrm{P}=0.01 \times 0.4 \times 10^{-3} \\
& \mathrm{P}=4 \times 10^{-6} \mathrm{C}-\mathrm{m}
\end{aligned}
$$

Torque, $\tau=\mathrm{pE} \sin \theta$

$$
\begin{aligned}
& \tau=4 \times 10^{-6} \times\left(10 \times 10^{-5}\right) \times \sin 30^{\circ} \\
& \tau=4 \times 10^{-10} \mathrm{~N}-\mathrm{m}
\end{aligned}
$$

45. Under isothermal condition, the pressure of a gas is given by $P=a V^{-3}$, where $a$ is a constant and $V$ is the volume of the gas. The bulk modulus at constant temperature is equal to
(1) $\frac{P}{2}$
(2) 2 P
(3) P
(4) $3 P$

Sol. (4)

$$
\mathrm{P}=\mathrm{aV}^{-3}
$$

$\frac{d P}{d V}=-3 \mathrm{aV}^{-4}$
Bulk modulus, $B=-V \frac{d P}{d V}$
$B=-V\left(\frac{-3 a}{V^{4}}\right)$
$B=3 \frac{\mathrm{a}}{\mathrm{V}^{3}}=3 \mathrm{P}$
46. A planet having mass 9 Me and radius 4 Re , where Me and Re are mass and radius of earth respectively, has escape velocity in km/s given by:
(Given escape velocity on earth $\mathrm{V}_{\mathrm{e}}=11.2 \times 10^{3} \mathrm{~m} / \mathrm{s}$ )
(1) 11.2
(2) 67.2
(3) 33.6
(4) 16.8

## Sol. (4)

Escape velocity, $v_{e}=\sqrt{\frac{2 G M}{R}}$
$\mathrm{V}_{\mathrm{p}}=\sqrt{\frac{2 \mathrm{G}\left(9 \mathrm{~m}_{\mathrm{e}}\right)}{4 \mathrm{R}_{\mathrm{e}}}}=\frac{3}{2}\left(\mathrm{~V}_{\mathrm{e}}\right)_{\text {earth }}$
$\mathrm{v}_{\mathrm{p}}=\frac{3}{2} \times 11.2 \mathrm{~km} / \mathrm{s}$
$\mathrm{v}_{\mathrm{p}}=16.8 \mathrm{~km} / \mathrm{s}$
47. A body of mass $(5 \pm 0.5) \mathrm{kg}$ is moving with a velocity of $(20 \pm 0.4) \mathrm{m} / \mathrm{s}$. Its kinetic energy will be
(1) $(1000 \pm 140) \mathrm{J}$
(2) $(500 \pm 140) \mathrm{J}$
(3) $(500 \pm 0.14) \mathrm{J}$
(4) $(1000 \pm 0.14) \mathrm{J}$

## Sol. (1)

Kinetic energy, $K E=\frac{1}{2} \mathrm{mv}^{2}$

$$
\mathrm{KE}=\frac{1}{2} \times 5 \times 20^{2}
$$

$$
\mathrm{KE}=1000 \mathrm{~J}
$$

$\frac{\Delta \mathrm{K}}{\mathrm{K}}=\frac{\Delta \mathrm{m}}{\mathrm{m}}+\frac{2 \Delta \mathrm{v}}{\mathrm{v}}$
$\frac{\Delta \mathrm{K}}{1000}=\frac{0.5}{5}+2 \times \frac{0.4}{20}$
$\Delta \mathrm{K}=1000(0.1+0.04)$
$\Delta \mathrm{K}=1000 \times 0.14$
$\Delta \mathrm{K}=140 \mathrm{~J}$
$K E=(1000 \pm 140) \mathrm{J}$
48. The difference between threshold wavelengths for two metal surfaces $A$ and $B$ having work function $\phi_{A}=9 \mathrm{eV}$ and $\phi_{\mathrm{B}}=4.5 \mathrm{eV}$ in nm is:
\{Given, hc $=1242 \mathrm{eV} \mathrm{nm}\}$
(1) 276
(2) 264
(3) 540
(4) 138

Sol. (4)
$\phi=\frac{\mathrm{hc}}{\lambda}$
$\lambda_{\mathrm{A}}=\frac{1242}{9}=138 \mathrm{~nm}$
$\lambda_{\text {B }}=\frac{1242}{4.5}=276 \mathrm{~nm}$
$\lambda_{\mathrm{B}}-\lambda_{\mathrm{A}}=276-138=138 \mathrm{~nm}$
49. The source of time varying magnetic field may be
(A) A permanent magnet
(B) An electric field changing linearly with time
(C) Direct current
(D) A decelerating charge particle
(E) An antenna fed with a digital signal

Choose the correct answer from the options given below:
(1) (B) and (D) only
(2) (C) and (E) only
(3) (D) only
(4) (A) only

Sol. (3)
Accelerated charge particle produces EMW which has time varying E and B.
If $E$ is linear function of time then $B$ will be constant.
50. A vessel of depth ' $d$ ' is half filled with oil of refractive index $n_{1}$ and the other half is filled with water of refractive index $n_{2}$. The apparent depth of this vessel when viewed from above will be -
(1) $\frac{d\left(n_{1}+n_{2}\right)}{2 n_{1} n_{2}}$
(2) $\frac{\mathrm{d}_{1} \mathrm{n}_{2}}{\left(\mathrm{n}_{1}+\mathrm{n}_{2}\right)}$
(3) $\frac{\mathrm{dn}_{1} \mathrm{n}_{2}}{2\left(\mathrm{n}_{1}+\mathrm{n}_{2}\right)}$
(4) $\frac{2 \mathrm{~d}\left(\mathrm{n}_{1}+\mathrm{n}_{2}\right)}{\mathrm{n}_{1} \mathrm{n}_{2}}$

Sol. (1)

$\left(\mathrm{d}_{\text {app }}\right)_{1}=\frac{\mathrm{d}}{2\left(\frac{\mathrm{n}_{1}}{\mathrm{n}_{2}}\right)}=\frac{\mathrm{n}_{2} \mathrm{~d}}{2 \mathrm{n}_{1}}$
$\left(\mathrm{d}_{\text {app }}\right)_{2}=\frac{\left(\mathrm{d}_{\text {app }}\right)_{1}+\frac{\mathrm{d}}{2}}{\mathrm{n}_{2}}$
$=\frac{\left(\frac{\mathrm{n}_{2}}{\mathrm{n}_{1}}+1\right) \frac{\mathrm{d}}{2}}{\mathrm{n}_{2}}$
$\left(\mathrm{d}_{\mathrm{app}}\right)_{2}=\frac{\left(\mathrm{n}_{1}+\mathrm{n}_{2}\right) \mathrm{d}}{2 \mathrm{n}_{1} \mathrm{n}_{2}}$

## SECTION - B

51. When a resistance of $5 \Omega$ is shunted with a moving coil galvanometer, it shows a full scale deflection for a current of 250 mA , however when $1050 \Omega$ resistance is connected with it in series, it gives full scale deflection for 25 volt. The resistance of galvanometer is $\qquad$ $\Omega$.

Sol. (50)
For ammeter,

$I_{g}(G)=\left(250-I_{g}\right) 5$
$I_{g}=\frac{1250}{5+G} \mathrm{~mA}$
For voltmeter,

$V=I_{g} R$
$25=\mathrm{I}_{\mathrm{g}}(\mathrm{G}+1050)$
From equation (1),
$25=\frac{1250 \times 10^{-3}}{\mathrm{G}+5}(\mathrm{G}+1050)$
$20(\mathrm{G}+5)=\mathrm{G}+1050$
$19 \mathrm{G}=1050-100$
$\mathrm{G}=\frac{950}{19}=50 \Omega$
52. The radius of $2^{\text {nd }}$ orbit of $\mathrm{He}^{+}$of Bohr's model is $r_{1}$ and that of fourth orbit of $\mathrm{Be}^{3+}$ is represented as $r_{2}$. Now the ratio $\frac{r_{2}}{r_{1}}$ is $x: 1$. The value of $x$ is $\qquad$
Sol. (2)
$\mathrm{r} \propto \frac{\mathrm{n}^{2}}{\mathrm{Z}}$
$\frac{\mathrm{r}_{2}}{\mathrm{r}_{1}}=\left(\frac{\mathrm{n}_{2}}{\mathrm{n}_{1}}\right)^{2} \times \frac{\mathrm{z}_{1}}{\mathrm{z}_{2}}$
$\frac{\mathrm{r}_{2}}{\mathrm{r}_{1}}=\left(\frac{4}{2}\right)^{2} \times \frac{2}{4}$
$\underline{\mathrm{r}_{2}}=2$
$\mathrm{r}_{1}$
$\mathrm{x}=2$
53. A solid sphere is rolling on a horizontal plane without slipping. If the ratio of angular momentum about axis of rotation of the sphere to the total energy of moving sphere is $\pi: 22$ the, the value of its angular speed will be rad/s.
Sol. (4)
Angular momentum,
$\mathrm{L}=\mathrm{I} \omega$
$\mathrm{L}=\frac{2}{5} \mathrm{MR}^{2} \omega$
Energy $=\frac{1}{2} \mathrm{MV}^{2}+\frac{1}{2} \mathrm{I} \omega^{2}$
$\mathrm{E}=\frac{1}{2} \mathrm{M}(\omega \mathrm{R})^{2}+\frac{1}{2}\left(\frac{2}{5} \mathrm{MR}^{2}\right) \omega^{2}$
$=\frac{7}{10} \mathrm{M} \omega^{2} \mathrm{R}^{2}$
$\frac{L}{E}=\frac{4}{7 \omega}=\frac{\pi}{22}$
$\omega=\frac{88}{7 \pi}=\frac{88}{7 \times \frac{22}{7}}=4 \mathrm{rad} / \mathrm{s}$
54. A fish rising vertically upward with a uniform velocity of $8 \mathrm{~ms}^{-1}$, observes that a bird is diving vertically downward towards the fish with the velocity of $12 \mathrm{~ms}^{-1}$. If the refractive index of water is $\frac{4}{3}$, then the actual velocity of the diving bird to pick the fish, will be $\qquad$ $\mathrm{ms}^{-1}$.

## Sol. (3)


$\mathrm{d}_{\text {app }}=\mathrm{d}_{1}+\mu \mathrm{d}$
$\mathrm{v}_{\text {app }}=\mathrm{v}_{1}+\mu \mathrm{v}$
$12=8+\frac{4}{3} v$
$4=\frac{4}{3} \mathrm{v}$
$\mathrm{v}=3 \mathrm{~m} / \mathrm{s}$
55. The elastic potential energy stored in a steel wire of length 20 m stretched through 2 cm is 80 J . The cross sectional area of the wire is $\qquad$ $\mathrm{mm}^{2}$.
(Given, $\mathrm{y}=2.0 \times 10^{11} \mathrm{Nm}^{-2}$ )

## Sol. (40)

Energy, $\mathrm{U}=\frac{1}{2} \mathrm{kx}^{2}$
$80=\frac{1}{2} \mathrm{k}\left(2 \times 10^{-2}\right)^{2}$
$\mathrm{k}=\frac{160}{4 \times 10^{-4}}$
$\mathrm{k}=4 \times 10^{5} \mathrm{~N} / \mathrm{m}$
$\frac{\mathrm{yA}}{\ell}=4 \times 10^{5}$
$A=\frac{4 \times 10^{5} \times 20}{2 \times 10^{11}}$
$\mathrm{A}=40 \times 10^{-6} \mathrm{~m}^{2}$
$\mathrm{A}=40 \mathrm{~mm}^{2}$
56. From the given transfer characteristic of a transistor in CE configuration, the value of power gain of this configuration is $10^{x}$, for $R_{B}=10 \mathrm{k} \Omega, \mathrm{R}_{\mathrm{C}}=1 \mathrm{k} \Omega$. The value of x is $\qquad$


Sol. (3)
Current gain, $\beta=\frac{\Delta \mathrm{I}_{\mathrm{C}}}{\Delta \mathrm{I}_{\mathrm{B}}}$
$\beta=\frac{10 \mathrm{~mA}}{100 \mu \mathrm{~A}}$
$\beta=100$
Power gain $\beta^{2} \frac{R_{C}}{R_{B}}$
$=10^{4} \times \frac{1}{10}$
$=10^{3}$
So, $x=3$
57. In the given figure, an inductor and a resistor are connected in series with a batter of emf $E$ volt. $\frac{E^{a}}{2 b} \mathrm{~J} / \mathrm{s}$ represents the maximum rate at which the energy is stored in the magnetic field (inductor). The numerical value of $\frac{b}{a}$ will be $\qquad$


Sol. (25)
$\mathrm{U}=\frac{1}{2} \mathrm{LI}^{2}$
$\mathrm{I}=\mathrm{I}_{0}\left(1-\mathrm{e}^{-\mathrm{t} / \tau}\right)$
Rate of energy, $\mathrm{P}=\frac{\mathrm{dU}}{\mathrm{dt}}$
$\mathrm{P}=\mathrm{LI} \frac{\mathrm{dI}}{\mathrm{dt}}$
$\frac{d P}{d t}=L\left(I \frac{\mathrm{~d}^{2} \mathrm{I}}{\mathrm{dt}^{2}}+\left(\frac{\mathrm{dI}}{\mathrm{dt}}\right)^{2}\right)$
For maximum rate, $\frac{\mathrm{dP}}{\mathrm{dt}}=0$
$\mathrm{I} \frac{\mathrm{d}^{2} \mathrm{I}}{\mathrm{dt}^{2}}=-\left(\frac{\mathrm{dI}}{\mathrm{dt}}\right)^{2}$
$\mathrm{I}=\mathrm{I}_{0}\left(1-\mathrm{e}^{\mathrm{t} / \tau}\right)$
$\frac{\mathrm{dI}}{\mathrm{dt}}=\frac{\mathrm{I}_{0}}{\tau} \mathrm{e}^{-\mathrm{t} / \tau}$
$\frac{\mathrm{d}^{2} \mathrm{I}}{\mathrm{dt}^{2}}=-\frac{\mathrm{I}_{0}}{\tau^{2}} \mathrm{e}^{-\mathrm{t} / \tau}$
By equation (1),
$I_{0}\left(1-e^{-t / \tau}\right) \times \frac{I_{0}}{\tau^{2}} e^{-t / \tau}=\frac{-I_{0}^{2}}{\tau^{2}} e^{-2 t / \tau}$
Let $\mathrm{e}^{-t / \tau}=\mathrm{x}$
$x-x^{2}=x^{2}$
$\mathrm{x}=\frac{1}{2}$
Maximum power,
$\mathrm{P}=\mathrm{LI} \frac{\mathrm{dI}}{\mathrm{dt}}$
$\mathrm{P}=\mathrm{LI}_{0}\left(1-\frac{1}{2}\right)\left(\frac{\mathrm{I}_{0}}{\tau} \times \frac{1}{2}\right)$
$\mathrm{P}=\frac{\mathrm{LI} I_{0}^{2}}{4 \times \frac{\mathrm{L}}{\mathrm{R}}}=\frac{\mathrm{I}_{0}^{2} \mathrm{R}}{4}$
$\mathrm{P}=\frac{\mathrm{E}^{2}}{4 \mathrm{R}}$
$\mathrm{a}=2,2 \mathrm{~b}=4 \mathrm{R}$
$\mathrm{b}=2 \mathrm{R}=50$
$\frac{\mathrm{b}}{\mathrm{a}}=25$
58. A potential $V_{o}$ is applied across a uniform wire of resistance $R$. The power dissipation is $P_{1}$. The wire is then cut into two equal halves and a potential of $\mathrm{V}_{\mathrm{o}}$ is applied across the length of each half. The total power dissipation across two wires is $P_{2}$. The ratio $P_{2}: P_{1}$ is $\sqrt{x}: 1$. The value of $x$ is $\qquad$
Sol. (16)


$$
\mathrm{P}_{1}=\frac{\mathrm{v}_{0}^{2}}{\mathrm{R}}
$$


$P_{2}=\frac{\mathrm{v}_{0}^{2}}{\left(\frac{\mathrm{R}}{2}\right)}+\frac{\mathrm{v}_{0}^{2}}{\left(\frac{\mathrm{R}}{2}\right)}$
$\mathrm{P}_{2}=4 \mathrm{P}_{1}$
$\frac{\mathrm{P}_{2}}{\mathrm{P}_{1}}=\frac{4}{1}=\frac{\sqrt{\mathrm{x}}}{1}$
$x=16$
59. At a given point of time the value of displacement of a simple harmonic oscillator is given as $\mathrm{y}=\mathrm{A} \cos \left(30^{\circ}\right)$. If amplitude is 40 cm and kinetic energy at that time is 200 J , the value of force constant is $1.0 \times 10^{\mathrm{x}} \mathrm{Nm}^{-1}$. The value of $x$ is $\qquad$
Sol. (4)
$v=\omega \sqrt{\mathrm{A}^{2}-\mathrm{x}^{2}}$
$y=A \times \frac{\sqrt{3}}{2}$
$v=\omega \sqrt{\mathrm{A}^{2}-\frac{3 \mathrm{~A}^{2}}{4}}=\frac{\omega \mathrm{A}}{2}$
Given, $\mathrm{KE}=200 \mathrm{~J}$
$\frac{1}{2} \mathrm{~m} \frac{\omega^{2} \mathrm{~A}^{2}}{4}=200$
$K A^{2}=1600 \quad\left(K=m \omega^{2}\right)$
$K=\frac{1600}{\left(40 \times 10^{-2}\right)^{2}}$
$\mathrm{K}=10^{4} \mathrm{~N} / \mathrm{m}$
$\mathrm{x}=4$
60. A thin infinite sheet charge and an infinite line charge of respective charge densities $+\sigma$ and $+\lambda$ are placed parallel at 5 m distance from each other. Points ' P ' and ' Q ' are at $\frac{3}{\pi} \mathrm{~m}$ and $\frac{4}{\pi} \mathrm{~m}$ perpendicular distances from line charge towards sheet charge, respectively. ' $\mathrm{E}_{\mathrm{P}}$ ' and ${ }^{\prime} \mathrm{E}_{\mathrm{Q}}$ ' are the magnitudes of resultant electric field intensities at point 'P' and 'Q' respectively. If $\frac{E_{P}}{E_{Q}}=\frac{4}{a}$ for $2|\sigma|=|\lambda|$, then the value of $a$ is $\qquad$
Sol. (6)


$$
\mathrm{E}_{\mathrm{P}}=\frac{2 \mathrm{~K} \lambda}{\mathrm{r}}-\frac{\sigma}{2 \varepsilon_{0}}
$$

$$
\mathrm{E}_{\mathrm{P}}=\frac{\sigma}{2 \varepsilon_{0}}-\frac{\lambda}{2 \pi \varepsilon_{0}\left(\frac{3}{\pi}\right)}
$$

$$
\mathrm{E}_{\mathrm{P}}=\frac{2 \sigma}{2 \varepsilon_{0}}-\frac{2 \sigma}{6 \varepsilon_{0}}=\frac{\sigma}{6 \varepsilon_{0}}
$$

Similarly, $\mathrm{E}_{\mathrm{Q}}=\frac{\sigma}{2 \varepsilon_{0}}-\frac{2 \sigma}{2 \pi \varepsilon_{0}\left(\frac{4}{\pi}\right)}=\frac{\sigma}{4 \varepsilon_{0}}$

$$
\frac{\mathrm{E}_{\mathrm{P}}}{\mathrm{E}_{\mathrm{Q}}}=\frac{4}{6}
$$

$$
a=6
$$

## SECTION - A

61. Given below are two statements :

Statement I: Permutit process is more efficient compared to the synthetic resin method for the softening of water.
Statement II: Synthetic resin method results in the formation of soluble sodium salts.
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Both the Statements I and II are correct
(2) Statement I is incorrect but Statement II is correct
(3) Statement I is correct but Statement II is incorrect
(4) Both the Statements I and II are incorrect

Sol. 4
Nowadays hard water is softened by using synthetic ion exchangers. This method is more efficient than zeolite process/Permutit process
62. Which one of the following is most likely a mismatch?
(1) Zinc - Liquation
(2) Copper - Electrolysis
(3) Titanium - van Arkel Method
(4) Nickel - Mond process

Sol. 1
Zinc is refined by distillation method, which is used for metals having low boiling point.
63. The energy of an electron in the first Bohr orbit of hydrogen atom is $-2.18 \times 10^{-18} \mathrm{~J}$. Its energy in the third Bohr orbit is $\qquad$ _.
(1) $\frac{1}{27}$ of this value
(2) $\frac{1}{9}$ th of this value
(3) One third of this value
(4) Three times of this value

Sol. 2
$E_{1,1}=-2.18 \times 10^{-18} \mathrm{~J}$
$\mathrm{E}_{3,1}=\mathrm{E}_{1,1} \times \frac{1^{2}}{3^{2}}$
$\mathrm{E}_{3,1}=\frac{1}{9} \times \mathrm{E}_{1,1}$
64.


In the above reaction, left hand side and right hand side rings are named as ' A ' and ' B ' respectively. They undergo ring expansion. The correct statement for this process is:
(1) Finally both rings will become six membered each.
(2) Ring expansion can go upto seven membered rings
(3) Finally both rings will become five membered each.
(4) Only A will become 6 membered.

Sol. 1




65. Match The following

| Column-A | Column-B |
| :--- | :--- |
| a) Nylon 6 | I. Natural Rubber |
| b) Vulcanized Rubber | II. Cross Linked |
| c) cis-1, 4-polyisoprene | III. Caprolactam |
| d) Polychloroprene | IV. Neoprene |

Choose the correct answer from options given below:
(1) $\mathrm{a} \rightarrow \mathrm{II}, \mathrm{a} \rightarrow$ III, $\mathrm{c} \rightarrow \mathrm{IV}, \mathrm{d} \rightarrow \mathrm{I}$
(2) $\mathrm{a} \rightarrow \mathrm{IV}, \mathrm{b} \rightarrow$ III, $\mathrm{c} \rightarrow$ II, $d \rightarrow$ I
(3) $\mathrm{a} \rightarrow \mathrm{III}, \mathrm{b} \rightarrow$ II, $\mathrm{c} \rightarrow \mathrm{I}, \mathrm{d} \rightarrow$ IV
(4) $\mathrm{a} \rightarrow \mathrm{III}, \mathrm{b} \rightarrow \mathrm{IV}, \mathrm{c} \rightarrow \mathrm{I}, \mathrm{d} \rightarrow \mathrm{II}$

## Sol. 3

Nylon-6 - Caprolactum (Monomer)
Natural rubber- Isoprene (Monomer)
Vulcanized rubber - Sulphur containing rubber
Neoprene- Chloroprene (Monomer)
66. What happens when a lyophilic sol is added to a lyophobic sol?
(1) Film of lyophobic sol is formed over lyophilic sol.
(2) Lyophilic sol is dispersed in lyophobic sol.
(3) Lyophobic sol is coagulated
(4) Film of lyophilic sol is formed over lyophobic sol.

Sol. 4
Protective film of lyophilic sol is formed over lyophobic sol.
Which protects it from coagulation.
67. In the reaction given below

(1)

(2)

(3)

(4)


Sol. 3

68. In the following reaction ' $X$ ' is

(1) $\mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{4} \mathrm{CH}_{2} \mathrm{Cl}$

(3) $\mathrm{Cl}-\mathrm{CH}_{2}-\left(\mathrm{CH}_{2}\right)_{4}-\mathrm{CH}_{2}-\mathrm{Cl}$
(2) $\mathrm{H}_{2} \mathrm{C}$

(4) $\mathrm{CH}_{3} \mathrm{CH}-\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{3}$ $\mathrm{CH}_{3}$

Sol. 4
n -alkanes on heating in this presence of anhydrous $\mathrm{AlCl}_{3}$ and hydrogen chloride gas isomerise to branched chain alkanes. The major product has one methyl side chain.

69. 2-Methyl propyl bromide reacts with $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{O}^{-}$and gives ' A ' whereas on reaction with $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$ it gives ' B '. The mechanism followed in these reactions and the products ' A ' and ' B ' respectively are :
(1) $\mathrm{S}_{\mathrm{N}} 1, \mathrm{~A}=$ tert-butyl ethyl ether; $\mathrm{S}_{\mathrm{N}} 1, \mathrm{~B}=2$-butyl ethyl ether
(2) $\mathrm{S}_{\mathrm{N}} 2, \mathrm{~A}=2$-butyl ethyl ether; $\mathrm{S}_{\mathrm{N}} 2, \mathrm{~B}=$ iso-butyl ethyl ether
(3) $\mathrm{S}_{\mathrm{N}} 2, A=$ iso-butyl ethyl ether; $\mathrm{S}_{\mathrm{N}} 1, B=$ tert-butyl ethyl ether
(4) $\mathrm{S}_{\mathrm{N}} 1, A=$ tert-butyl ethyl ether; $\mathrm{S}_{\mathrm{N}} 2, \mathrm{~B}=$ iso-butyl ethyl ether

Sol. 3
(i)

$\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{O}^{-}$is strong nucleophile.
(ii)

$\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$ is weak nucleophile.
70. In the reaction given below

' A ' is
(1)

(2)

(3)

(4)


Sol. 2

71. $\mathrm{D}-(+)$ Glyceraldehyde $\xrightarrow[\substack{\text { ii) } \mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+} \\ \text {iii) } \mathrm{HNO}_{3}}]{\text { i } \mathrm{HCN}}$

The products formed in the above reaction are
(1) Two optically active products
(2) One optically inactive and one meso product.
(3) One optically active and one meso product
(4) Two optically inactive products

Sol. 3

$D(+)$
Glyceraldehyde

72. $\mathrm{CIF}_{5}$ at room temperature is a:
(1) Colourless liquid with square pyramidal geometry
(2) Colourless gas with trigonal bipyramidal geometry
(3) Colourless gas with square pyramidal geometry
(4) Colourless liquid with trigonal bipyramidal geometry

Sol. 1

$\mathrm{ClF}_{5}$ is colourless liquid.
73. The pair of lanthanides in which both elements have high third - ionization energy is:
(1) $\mathrm{Dy}, \mathrm{Gd}$
(2) $\mathrm{Eu}, \mathrm{Gd}$
(3) $\mathrm{Lu}, \mathrm{Yb}$
(4) $\mathrm{Eu}, \mathrm{Yb}$

Sol. 4

74. The mismatched combinations are
A. Chlorophyll - Co
B. Water hardness - EDTA
C. Photography $-\left[\mathrm{Ag}(\mathrm{CN})_{2}\right]^{-}$
D. Wilkinson catalyst - $\left[\left(\mathrm{Ph}_{3} \mathrm{P}\right)_{3} \mathrm{RhCl}\right]$
E. Chelating ligand - D-Penicillamine

Choose the correct answer from the options given below :
(1) A and C Only
(2) D and E Only
(3) A and E Only
(4) A, C, and E Only

Sol. 1
Mg is present in chlorophyll and in black and white photography the developed film is fixed by washing with hypo solution which dissolves the undecomposed AgBr to form a complex ion $\left[\mathrm{Ag}\left(\mathrm{S}_{2} \mathrm{O}_{3}\right)_{2}\right]^{3-}$
75. Which of the following statements are not correct ?
A. The electron gain enthalpy of F is more negative than that of Cl .
B. Ionization enthalpy decreases in a group of periodic table.
C. The electronegativity of an atom depends upon the atoms bonded to it.
D. $\mathrm{Al}_{2} \mathrm{O}_{3}$ and NO are examples of amphoteric oxides.

Choose the most appropriate answer from the options given below :
(1) A, C and D Only
(2) B and D Only
(3) A, B and D Only
(4) A, B, C and D

## Sol. 1

Electronegativity of an element depends on the atom with which it is attached.
$\mathrm{NO}=$ neutral oxide
$\mathrm{Al}_{2} \mathrm{O}_{3}=$ amphoteric oxide
76. The radical which mainly causes ozone depletion in the presence of UV radiations is :
(1) $\mathrm{NO}^{\bullet}$
(2) $\dot{\mathrm{O}} \mathrm{H}$
(3) $\mathrm{CH}_{3}^{\bullet}$
(4) $\mathrm{Cl}^{\bullet}$

Sol. 4
$\mathrm{O}_{2}(\mathrm{~g}) \xrightarrow{\mathrm{UV}} \mathrm{O}(\mathrm{g})+\mathrm{O}(\mathrm{g})$
$\mathrm{O}_{2}(\mathrm{~g})+\mathrm{O}(\mathrm{g}) \longrightarrow \mathrm{O}_{3}(\mathrm{~g})$
$\mathrm{CF}_{2} \mathrm{Cl}_{2}(\mathrm{~g}) \xrightarrow{\mathrm{uv}} \dot{\mathrm{Cl}}(\mathrm{g})+\dot{\mathrm{C}} \mathrm{F}_{2} \mathrm{Cl}(\mathrm{g})$
$\dot{\mathrm{Cl}}(\mathrm{g})+\mathrm{O}_{3}(\mathrm{~g}) \longrightarrow \mathrm{Cl} \dot{\mathrm{O}}(\mathrm{g})+\mathrm{O}_{2}(\mathrm{~g})$
$\mathrm{Cl} \dot{\mathrm{O}}(\mathrm{g})+\mathrm{O}(\mathrm{g}) \longrightarrow \dot{\mathrm{Cl}}(\mathrm{g})+\mathrm{O}_{2}(\mathrm{~g})$
77. In which of the following processes, the bond order increases and paramagnetic character changes to diamagnetic one?
(1) $\mathrm{O}_{2} \rightarrow \mathrm{O}_{2}^{+}$
(2) $\mathrm{O}_{2} \rightarrow \mathrm{O}_{2}^{2-}$
(3) $\mathrm{NO} \rightarrow \mathrm{NO}^{+}$
(4) $\mathrm{N}_{2} \rightarrow \mathrm{~N}_{2}^{+}$

Sol. 3
NO is paramagnetic with $\mathrm{BO}=2.5, \mathrm{NO}^{+}$is diamagnetic with $\mathrm{BO}=3$
78. The incorrect statement from the following for borazine is:
(1) It is a cyclic compound.
(2) It has electronic delocalization.
(3) It can react with water.
(4) It contains banana bonds.

Sol. 4
Borazine is $\mathrm{B}_{3} \mathrm{~N}_{3} \mathrm{H}_{6}$

$\mathrm{B}_{3} \mathrm{~N}_{3} \mathrm{H}_{6}+9 \mathrm{H}_{2} \mathrm{O} \rightarrow 3 \mathrm{NH}_{3}+3 \mathrm{H}_{3} \mathrm{BO}_{3}+3 \mathrm{H}_{2}$
79. Among the following compounds, the one which shows highest dipole moment is
(1)

(2)

(3)

(4)


Sol. 2
Among the given compounds, the following compound has the highest dipole moment because both the +ve and -ve ends acquire aromaticity.

80. $\mathrm{Be}(\mathrm{OH})_{2}$ reacts with $\mathrm{Sr}(\mathrm{OH})_{2}$ to yield an ionic salt. Choose the incorrect option related to this reaction from the following :
(1) Be is tetrahedrally coordinated in the ionic salt.
(2) The reaction is an example of acid - base neutralization reaction.
(3) The element Be is present in the cationic part of the ionic salt.
(4) Both Sr and Be elements are present in the ionic salt.

Sol. 3
$\mathrm{Be}(\mathrm{OH})_{2}$ is amphoteric in nature.
$\mathrm{Sr}(\mathrm{OH})_{2}$ is basic in nature.
These two undergo acid - base reaction to form a salt.
$\mathrm{Be}(\mathrm{OH})_{2}+\mathrm{Sr}(\mathrm{OH})_{2} \rightarrow \underset{\text { (salt) }}{\mathrm{Sr}\left[\mathrm{Be}(\mathrm{OH})_{4}\right]}$
(salt)

## SECTION - B

81. Solution of 12 g of non-electrolyte (A) prepared by dissolving it in 1000 mL of water exerts the same osmotic pressure as that of 0.05 M glucose solution at the same temperature. The empirical formula of A is $\mathrm{CH}_{2} \mathrm{O}$. The molecular mass of A is $\qquad$ g. (Nearest integer)

Sol. 240
$\pi_{\mathrm{A}}=\pi_{\text {glucose }}$
$\mathrm{C}_{\mathrm{A}} \mathrm{RT}=\mathrm{CRT}$
$\frac{12 / \mathrm{M}_{\mathrm{A}}}{1}=0.05$
$M_{A}($ Molar mass of $A)=\frac{12}{0.05}=\frac{1200}{5}=240 \mathrm{gm}$
82. $\mathrm{KMnO}_{4}$ is titrated with ferrous ammonium sulphate hexahydrate in presence of dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$. Number of water molecules produced for 2 molecules of $\mathrm{KMnO}_{4}$ is $\qquad$ _.
Sol. 68
$2 \mathrm{KMnO}_{4}+8 \mathrm{H}_{2} \mathrm{SO}_{4}+10 \mathrm{FeSO}_{4} \cdot\left(\mathrm{NH}_{4}\right)_{2} \mathrm{SO}_{4} \cdot 6 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{K}_{2} \mathrm{SO}_{4}+2 \mathrm{MnSO}_{4}+5 \mathrm{Fe}_{2}\left(\mathrm{SO}_{4}\right)_{3}+10\left(\mathrm{NH}_{4}\right)_{2} \mathrm{SO}_{4}+68 \mathrm{H}_{2} \mathrm{O}$ On the basis of above equation, 68 molecules of water will be produced from 2 molecules of $\mathrm{KMnO}_{4}$.
83. 20 mL of calcium hydroxide was consumed when it was reacted with 10 mL of unknown solution of $\mathrm{H}_{2} \mathrm{SO}_{4}$. Also 20 mL standard solution of 0.5 MHCl containing 2 drops of phenolphthalein was titrated with calcium hydroxide, the mixture showed pink colour when burette displayed the value of 35.5 mL whereas the burette showed 25.5 mL initially. The concentration of $\mathrm{H}_{2} \mathrm{SO}_{4}$ is $\qquad$ M.(Nearest integer)

## Sol. 1

miliequivalent of $\mathrm{Ca}(\mathrm{OH})_{2}=$ miliequivalent of $\mathrm{H}_{2} \mathrm{SO}_{4}$
$\mathrm{M}_{1} \times 2 \times 20 \quad=\quad \mathrm{M}_{2} \times 2 \times 10$
$2 \mathrm{M}_{1}=\mathrm{M}_{2}$
miliequivalent of $\mathrm{HCl}=$ miliequivalent of $\mathrm{Ca}(\mathrm{OH})_{2}$
$20 \times 0.5=10 \times \mathrm{M}_{1} \times 2$
$\mathrm{M}_{1}=0.5 \mathrm{M}$
Concentration of $\mathrm{H}_{2} \mathrm{SO}_{4}=\mathrm{M}_{2}=2 \mathrm{M}_{1}$

$$
\begin{aligned}
& =2 \times 0.5 \\
& =1 \mathrm{M}
\end{aligned}
$$

84. $\mathrm{t}_{87.5}$ is the time required for the reaction to undergo $87.5 \%$ completion and $\mathrm{t}_{50}$ is the time required for the reaction to undergo $50 \%$ completion. The relation between $\mathrm{t}_{87.5}$ and $\mathrm{t}_{50}$ for a first order reaction is $\qquad$ $\mathrm{t}_{87.5}$ $=x \times t_{50}$. The value of $x$ is $\qquad$ . (Nearest integer)
Sol. 3

85. A certain quantity of real gas occupies a volume of $0.15 \mathrm{dm}^{3}$ at 100 atm and 500 K when its compressibility factor is 1.07 . Its volume at 300 atm and 300 K (When its compressibility factor is 1.4 ) is $\times 10^{-4} \mathrm{dm}^{3}$. (Nearest integer)

Sol. 392

$$
\begin{aligned}
& \mathrm{Z}=\frac{\mathrm{PV}}{\mathrm{nRT}} \\
& \frac{\mathrm{Z}_{1}}{\mathrm{Z}_{2}}=\left(\frac{\mathrm{P}_{1} \mathrm{~V}_{1}}{\mathrm{nRT}}\right) \times\left(\frac{\mathrm{nRT}_{1}}{\mathrm{P}_{2} \mathrm{~V}_{2}}\right) \\
& \frac{1.07}{1.4}=\left(\frac{100 \times 0.15}{500}\right)\left(\frac{300}{300 \times \mathrm{V}_{2}}\right) \\
& \mathrm{V}_{2}=\frac{0.03 \times 1.4}{1.07}=0.03925 \\
&=392 \times 10^{-4} \mathrm{dm}^{3}
\end{aligned}
$$

86. A metal surface of $100 \mathrm{~cm}^{2}$ area has to be coated with nickel layer of thickness 0.001 mm . A current of 2 A was passed through a solution of $\mathrm{Ni}\left(\mathrm{NO}_{3}\right)_{2}$ for ' $x$ ' seconds to coat the desired layer. The value of x is $\qquad$ _.
(Nearest integer)
( $\rho_{\mathrm{Ni}}\left(\right.$ density of Nickel) is $10 \mathrm{~g} \mathrm{~mL}^{-1}$, Molar mass of Nickel is $60 \mathrm{~g} \mathrm{~mol}^{-1} \mathrm{~F}=96500 \mathrm{C} \mathrm{mol}^{-1}$ )
Sol. 161
Volume of nickel required $=100 \times 0.001 \times 10^{-3} \times 100$

$$
=0.01 \mathrm{~cm}^{3}
$$

Mass of Nickel required $=0.01 \times 10$

$$
=0.1 \mathrm{gm}
$$

Moles $=\frac{0.1}{60}=\frac{1}{600} \mathrm{~mol}$
$\mathrm{Ni}^{2+}+2 \mathrm{e}^{-} \rightarrow \mathrm{Ni}(\mathrm{s})$
for coating of 1 mol Ni , charge required $=2 \times 96500 \mathrm{C}$
for coating of $\frac{1}{600} \mathrm{~mol}$, charge required $=2 \times 96500 \times \frac{1}{600} \mathrm{C}$

$$
=\frac{965}{3} C
$$

$I=\frac{q}{t}$
$\mathrm{t}=\frac{965 / 3}{2}=160.83 \mathrm{sec} \approx 161$
87. 25.0 mL of $0.050 \mathrm{MBa}\left(\mathrm{NO}_{3}\right)_{2}$ is mixed with 25.0 mL of 0.020 M NaF . $\mathrm{K}_{\mathrm{sp}}$ of $\mathrm{BaF}_{2}$ is $0.5 \times 10^{-6}$ at 298 K . The ratio of $\left[\mathrm{Ba}^{2+}\right]\left[\mathrm{F}^{-}\right]^{2}$ and $\mathrm{K}_{\mathrm{sp}}$ is $\qquad$ (Nearest integer)
Sol. 5
$\left[\mathrm{Ba}^{+2}\right]=\frac{25 \times 0.05}{50}=0.025 \mathrm{M}$
$\left[\mathrm{F}^{-}\right]=\frac{25 \times 0.02}{50}=0.01 \mathrm{M}$
$\left[\mathrm{Ba}^{+2}\right]\left[\mathrm{F}^{-}\right]^{2}=25 \times 10^{-7}$
$\mathrm{K}_{\mathrm{sp}}=5 \times 10^{-7}$ (given)
Ratio $=\frac{\left[\mathrm{Ba}^{+2}\right]\left[\mathrm{F}^{-}\right]^{2}}{\mathrm{~K}_{\text {sp }}}=5$
88. $\quad \mathrm{A}_{2}+\mathrm{B}_{2} \rightarrow 2 \mathrm{AB} . \Delta \mathrm{H}_{\mathrm{f}}^{0}=-200 \mathrm{~kJ} \mathrm{~mol}^{-1}$ new line $\mathrm{AB}, \mathrm{A}_{2}$ and $\mathrm{B}_{2}$ are diatomic molecules. If the bond enthalpies of $A_{2}, B_{2}$ and $A B$ are in the ratio $1: 0.5: 1$, then the bond enthalpy of $A_{2}$ is $\qquad$ $\mathrm{kJ} \mathrm{mol}^{-1}$. (Nearest integer)

## Sol. 800

$\mathrm{A}_{2}+\mathrm{B}_{2} \rightarrow 2 \mathrm{AB} \quad \Delta \mathrm{H}_{\mathrm{f}}^{\circ}=-200 \mathrm{~kJ} / \mathrm{mol}$
Bond enthalpy of $\mathrm{A}_{2}=\mathrm{x}$
Bond enthalpy of $\mathrm{B}_{2}=0.5 \mathrm{x}$
Bond enthalpy of $\mathrm{AB}=\mathrm{x}$
$\Delta \mathrm{H}_{\mathrm{f}}^{\circ}=\mathrm{x}+0.5 \mathrm{x}-2 \mathrm{x}=-2(200)$
$-0.5 x=-400$
$\mathrm{x}=\frac{400}{0.5}=800 \mathrm{~kJ} / \mathrm{mol}$
Bond enthalpy of $\mathrm{A}_{2}=\mathrm{x}=800 \mathrm{~kJ} / \mathrm{mol}$
89. An organic compound gives 0.220 g of $\mathrm{CO}_{2}$ and 0.126 g of $\mathrm{H}_{2} \mathrm{O}$ on complete combustion. If the $\%$ of carbon is 24 then the \% of hydrogen is $\qquad$ $\times 10^{-1}$. ( Nearest integer)

## Sol. 56

$\%$ of carbon $=\frac{\frac{0.220}{44} \times 12}{x} \times 100$
( $x=$ mass of organic compound $)$
$24=\frac{6}{x}$
$\mathrm{x}=0.25 \mathrm{gm}$
$\%$ of $\mathrm{H}=\frac{\frac{0.126}{18} \times 2 \times 1}{0.25} \times 100$
$=5.6=56 \times 10^{-1}$
90. For the given reaction


The total number of possible products formed by tertiary carbocation of A is
Sol. 5

(A)

(B)



