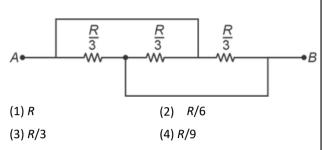


PHYSICS

SECTION - A Multiple Choice Questions: This section contains 20 multiple

choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct. **Choose the correct** answer:

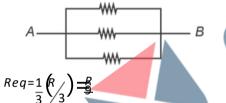
1. Find the equivalent resistance across A & B.



Answer (4)

2.

sol. Equivalent circuit can be redrawn as



A uniform wire of linear charge density is placed along *y*-axis. The locus of equipotential surface is

(1) x2 + y2 + z2 = constant

- (2) x2 + z2 = constant
- (3) xyz = constant
- (4) xy + yz + zx = constant

Answer (2)

Sol. Concentric cylinders are the equipotential surface.

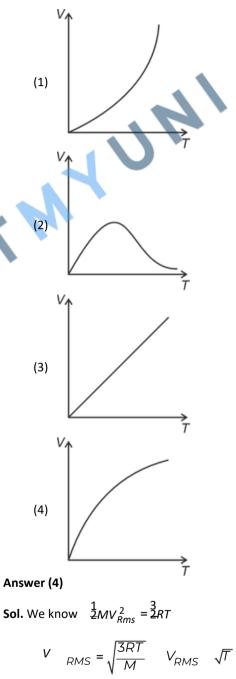
3. Which of following reaction is correct?

(Where symbols have their usual meanings)

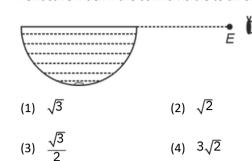
(1)
$$n \to p + e^{-} +$$
 (2) $n \to p + e^{+} +$
(3) $n \to p + e^{+} + -$ (4) $n \to p + e^{-} +$



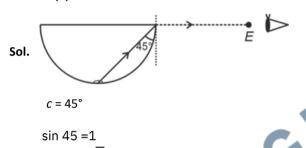
- Sol. In 2, 3 charge conservation is not holding and in neutron decay, antineutrin) os(released.
- 4. The graph of root mean square velocity *v/s* temperature is



A coin is placed at the bottom of a hemispherical container Sol. Let the resistance per unit length be , then filled with a liquid of refractive index . Find the least refractive index if the coin is visible to an observer at *E*.

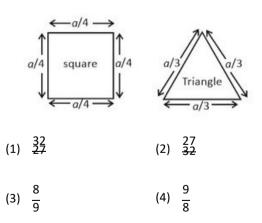


Answer (2)



= <u>sin</u> # §²

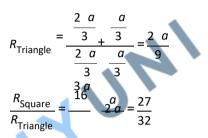
 In the given figure, the square and the triangle have same resistance per unit length. Find the ratio of their resistances about adjacent corners.





$$R_{\text{Square}} = \frac{\frac{3 \ a}{4} + \frac{a}{4}}{\frac{3 \ a}{4} + \frac{a}{4}} = \frac{3 \ a}{16}$$

For triangle,



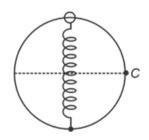
Assertion : Work done by central force is independent of path.

Reason: Potential energy is associated with every force. (1) Both Assertion and Reason are correct (2) Assertion is correct, Reason is incorrect (3) Assertion is incorrect, Reason is correct (4) Both Assertion and Reason are incorrect

Answer (4)

7.

- **Sol.** Not all central force/s are conservative so work done by central force might depend on path.
- 8. There is smooth ring of radius *R* in vertical plane. A spring of natural length *R* & elastic constant *K* is vertical along a diameter. The free end is connected to bead of mass *m* & when slightly disturbed it reaches point *C* with speed v where v is



(1)
$$\sqrt{\frac{KR2(\sqrt{2-1})}{2}mgR}$$

(2)
$$\sqrt{\frac{2KR^2(\sqrt{2-1})}{m}mR}$$

(3)
$$\sqrt{\frac{2KR2(\sqrt{2-1})}{m}mR}$$

(4)
$$\sqrt{\frac{KR2(\sqrt{2-1})}{m}mR}$$

Answer (2)

Sol. Loss in PE = gain in KE

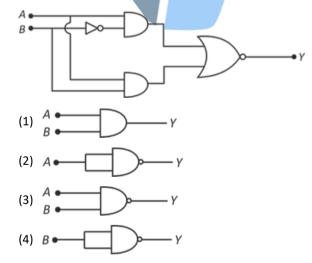
$$\frac{1}{2\kappa} \{2\} - 12k - (12) R2 + 12mgR = mv$$

$$\frac{1}{2\kappa R^{2}} 1 - 2 - 1 + 22 12m + vm2 gR = kR^{2}$$

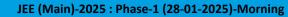
$$(2 - 1) + m 1mgv \neq 2$$

$$\sqrt{\frac{2\kappa R^{2}}{m}} = v$$

9. The equivalent logic gate for the circuit shown below is







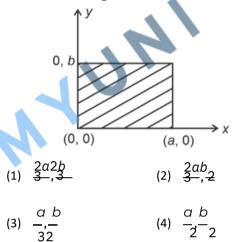
Sol. $Y = A\overline{B + AB}$

 $Y = \overline{A(B+B)}$

 $Y = \overline{A}$

10. Surace mass density varies as = $\frac{0x}{ab}$ for the given

plane sheet. Find the position of centre of mass for the distribution given



Answer (2)

2

Sol. As there is no variation of mass density in y direction. So

Now for *x* direction

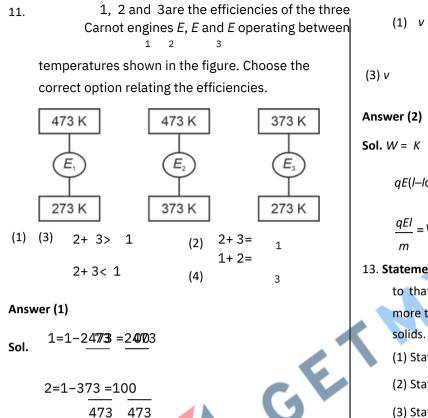
$$dm = \frac{0x \ b.dx}{ab}$$

$$xcm = \frac{a}{a} xdm = \frac{0}{a}^{a} x2dx$$

$$xcm = \frac{a}{0} dm = \frac{0}{a}^{a} x2dx$$

$$0 \ a3 \ 2$$

$$x_{cm} = \frac{3 \ a2}{3} = \frac{2}{3}a$$
So $r_{cm} = \frac{2}{3}a, \frac{b}{3}$



273

473

 $\frac{373}{373} = \frac{373}{373}$

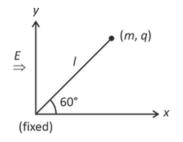
3=1-

1-=100 2

i.e., 2+3 1

100

3



(1)
$$v = \sqrt{\frac{2qEl}{m}}$$
 (2) $v = \sqrt{\frac{qEl}{m}}$
 $v = \sqrt{\frac{qEl}{2m}}$ (4) $v = 2\sqrt{\frac{qEl}{m}}$

 $qE(I-lcos60) = \frac{1}{2}mv2$

qEl = v2

13. Statement-I: Velocity of sound in solids is more compared to that in gases. Statement-II : Bulk modules of gas is more than that of

(1) Statement-I is correct statement-II is correct

(2) Statement-I is correct statement-II is incorrect

(3) Statement-I is incorrect statement-II is correct

(4) Statement-I is incorrect statement-II is incorrect

Answer (2)

Sol. Speed of sound in medium depends on elastic and inertia properly of medium.

for gas
$$v = \sqrt{\frac{B}{A}}$$

for solids = \sqrt{Y}

The elastic properly of solids happens to be many fold greater than that of elastic properly of gases. Bulk modulus of gas depends on the process,

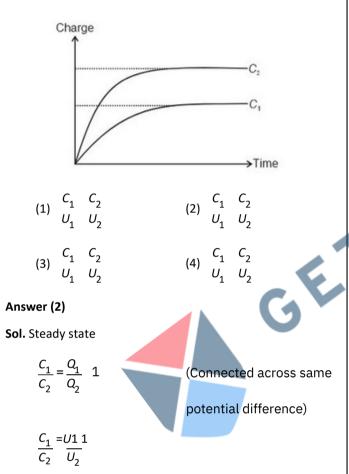
 $B = -\frac{P}{V/V}$, which varies between 0 to therefore in

general statement 2 is incorrect.

JEE (Main)-2025 : Phase-1 (28-01-2025)-Morning

 To capacitor C a1nd C are2 connected across same battery and store energies U 1and U re2spectively at

steady state. Choose the correct option by observing the graph of charge vs time shown below.



15. Energy of photon of wavelength is E which is equal to kinetic energy of proton of mass *pm*. The ratio of de Broglie wavelengths of proton and photon is

(1)
$$\frac{1}{c}\sqrt{\frac{2E0}{m_{\rho}}}$$
 (2) $\frac{1}{c}\sqrt{\frac{E_{0}}{2mp}}$
(3) $\frac{2}{c}\sqrt{\frac{E0}{m_{\rho}}}$ (4) $\frac{1}{2c}\sqrt{\frac{E_{0}}{m_{\rho}}}$

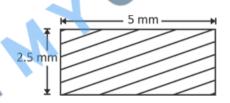


Sol.
$$\frac{1}{2} m_p \sqrt{2} = E0$$
 $p = \sqrt{2mpE0} = \frac{h}{\sqrt{2mpE0}}$

for photon
$$\frac{hc}{=}E0$$
 $ph=\frac{hc}{E_0}$

$$\frac{p}{ph} = \frac{h}{\sqrt{2mpE0}} \frac{E0}{hc} = \frac{1}{c} \sqrt{\frac{E_0}{2mp}}$$

16. The lengths of a rectangular sheet is measured from a screw gauge of pitch 0.75 mm and number of division on circular scale = 15. Find maximum possible error in measurement of area.



- (1) 0.225 mm2
- (2) 0.375 mm2
- (3) 0.75 mm2
- (4) 0.30 mm2

Answer (2)

Sol. Lest count of screw gauge

=0.75=0.05mm
15
Now, S = Ib

$$\frac{S}{-5} = \frac{1+b}{-7} + \frac{b}{-b}$$

 $\frac{S}{-5} = \frac{0.05}{-5} + \frac{9.95}{-5} = 100 + 100^{2}$

$$S = \frac{3}{100} \quad 52.5 = 0.375 \text{mm}^2$$

100

are two prisms of refractive indices of 1.54 & 1.72 respectively. If ray is not deviating after passing through two prisms, then find prism angle of second prism if prism angle of first prism is 4°

(1) 2°

(3) 4°

 $(2)_{3,5}^{\circ}$

Answer (2)

Sol. *S* = 1 *S* ,

 $(1.54 - 1)4^{\circ} = (1.72 - 1) A.$ A = 3°

18. The energy associated with a cylindrical region due to an EM wave $E = 100 \sin(kx - t)$ is U0. Find the equation of EM wave for which a cylinder of same length and half the diameter (as previous one) contains same energy U0.

(1) $200\sin(kx - t)$ (2) 25sin(t - kx) (3) $50\sin(kx - t)$ (4) 400sin(t Answer (1) **Sol.** For EM wave $E = 100 \sin(t - kx)$ Energy density $a = \int_{v}^{1} 0(E0)^2$ Initial volume $v = \mathbf{Q}/\mathbf{Q}$ Volume in next situation = $v = \frac{R2I}{\Delta} = \frac{V_0}{4}$ So, $\frac{u0=10(E)}{\frac{2}{2}0} \frac{0}{2} \frac{v0=10(E2)v0}{\frac{2}{2}1} \frac{1}{4}$ E2 0= ____ E1=2E0=200 V/m So, required equation will have 200 V/m of amplitude.

19.

20.

SECTION - B

Numerical Value Type Questions: This section contains 5 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.

21. The dimensions of Young's modules of elasticity per unit

length is MaLbTc then |a + b + c| is

Answer (3)

Sol.
$$\frac{Y}{L} = \frac{MLT^{-2}}{L^2} \stackrel{1}{L} = ML - 2T - 2$$

 $|1 - 2 - 2| = 3$

22. In a YDSE, the distance of the 10th bright fringe from the central maxima is 10 mm when light of wavelength used is 600 nm. Find the distance (in mm) of the 10th bright fringe from the central maxima if light of wavelength 660 nm is used instead.

Answer (11)

Sol. v = 10 d

 $\frac{y_2}{y_1} = \frac{2}{1}$

 $y_{2} = 2 y_{1}$

660 nm = 600 nm 10mm

= 11 mm

23.

24.

25.

CHEMISTRY

(i) > (ii) > (iii) > (iv) is the correct order of stability of SECTION - A Multiple Choice Questions: This section contains, carbocation. 20 multiple In the given reaction sequence: choice guestions. Each guestion has 4 choices (1), (2), (3) and 3. (4), out of which ONLY ONE is correct. CI Choose the correct answer : (i) KOH, 623 K, 300 atm (i) Br₂/KOH (R) (ii) CHCl₃/KOH (ii) CO₂ 1. What is the rate of reaction for releasing CO2(g) with (iii) H* aq. NaHCO3 among following? (iv) NH₃, Δ COOH COOH The compound R is OH NO. NO₂ CH3 NH₂ (1) (2)(2)(3)NO₂ (1) NH₂ (1) (1) > (2) > (3)(2)(3) > (2) > (1)OH OH (4)(2) > (3) > (1)(3)(1) > (3) > (2)NC CN Answer (3) (3) (4)Sol. pKa (Benzoic acid) : 4,27 pKa (o-Toluic acid) : $3.91 \rightarrow$ due to ortho effect. Answer (4) pKa (Picric acid) : 0.3 Sol. C OH Consider the following carbocations COOH 2. KOH, 623 K, 300 atm (i) CO2 (Ŧ) (ii) H (Ph).C € Ph₂CH CH3CH2-CH-CH2 NH_3, Δ OH OH (ii) OH (i) (iii) (iv) NH₂ - NH₂ NC The correct increasing order of stability of these CHCI3/KOH Br2/KOH carbocations is: (1) (i) < (ii) < (iii) < (iv) (3) (ii)(<2)(i(iii)v) < <(iv()iii<) (<i) (ii) \leq_1 (i) Which of the following pair have square pyramidal (4) (iv) < (iii) < (i) < (ii) shape? (2) SbF5, BrF5 (1) BrF5, XeOF4 Answer (2) Sol. Tropylium carbocation is most stable due to presence of (3) PCI5, XeOF4 (4) PCI5, SbF5 aromaticity.

Answer (1)

Sol. $BrF_5 \Rightarrow sp^3d^2$

 $F^{3}d^{2}$ F F

Square pyramidal

$$XeOF_{4} \Rightarrow sp^{3}d^{2} \qquad F \swarrow Berger F$$

Square pyramidal

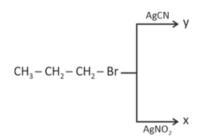
5. Which of the following set of quantum numbers have same energy?

(a) n = 2, l = 2, m = +1
(b) n = 2, l = 1, m = -1
(c) n = 3, l = 2, m = 0
(d) n = 3, l = 2, m = 1
(1) a, b
(2) b, c

- (3) c, d
- (4) a, c
- Answer (3)
- Sol.: (a) 2d = does not exist
 - (b) 2p
 - (c) 3d same energy
 - (d) 3d same energy

The value of n + l is same for (c) and (d) both represents 3d orbital

6. Consider the following reaction



The major product x and y respectively are

(1) CH3CH2CH2ONO &

CH3CH2CH2CN (2) CH3CH2CH2NO2 &

CH3CH2CH2CN (3) CH3CH2CH2NO2 &

CH3CH2CH2NC (4) CH3CH2CH2ONO

Answ&er C(3H)3 CH2CH2CN

Sol. CN- and NO-2 are ambidentate nucleophile but since

AgCN and AgNO2 are covalent compound, so only the nitrogen can donate electrons.

 Match the following column and choose the correct option.

.	Column-I		Column-II
(A)	H2O2→H2O+O2	(P)	Combustion reaction Disproportionation
(B)	NaH→Na+H2	(Q)	reaction Decomposition
(C)	CH4+02→ CO2+H2O	(R)	reaction Displacement
(D)	Fe+CuSO4→ FeSO4+Cu	(S)	reaction

(2) A-(R), B-(Q), C-(S), D-(P)

(3) A-(Q), B-(R), C-(P), D-(S)

(4) A-(R), B-(Q), C-(P), D-(S)

Answer (3)

Sol. H2O2 \rightarrow H2O + O2 Disproportionation reaction

 $NaH \rightarrow Na+H_2$: Decomposition reaction

CH4 +O2 →CO2 +H2O : Combustion reaction

Fe + CuSO4 → FeSO4 + Cu : Displacement reaction

 Among the following the incorrect order of atomic radius is

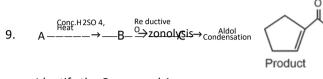
(1) B > Al > Mg > F

(3) Mg > Al > Be > O

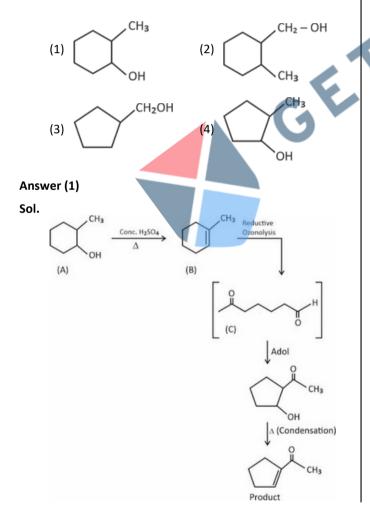
Answer (1)

Sol. As we move down the group size increases whereas on moving left to right in a period size decreases

(2) Alg BBeN NF> F



Identify the Compound A



10. $Et \ N \ Cl and Et \ CH \ CH \ (2)$

Statement-I: Compound (2) shows faster alkaline hydrolysis compared to (1).

Statement-II: Compound (1) shows substitution via neighbouring group participation.

- (1) Statement-I is correct and statement-II is incorrect
- (2) Statement-I is incorrect and statement-II is correct
- (3) Statement-I and statement-II both are correct
- (4) Statement-I and statement-II both are incorrect

Answer (2)



11. Which of the following has same energy in absence of electric and magnetic field for hydrogen atom?

ОН

(1) 2s, 3p	(2) 3s, 2p
(3) 2s, 2p	(4) 3s, 4f

Answer (3)

Sol. For hydrogen atom in absence of electric and magnetic field, energy only depends on principal quantum number(n). For same the value of n, energy will be same Hence 2s and 2p have same energy.

12. Which of the following reaction(s)/test(s) can be used to distinguish acetaldehyde and acetone?

(A) Iodoform Test	(B) Cannizzaro
(C) Aldol Condensation	(D) Fehling's Test
(E) Tollen's Test	(F) Clemmensen's Reduction

(1) (D), (E) Only
 (2) (A), (B), (C), (F) only
 (3) (B), (C), (F) only
 (4) (B), (C), (D), (E) only

Answer (1)

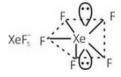
Sol. CH3CHO will react with Fehling's solution and Tollen's reagent while CH3COCH3 will not. lodoform, Aldol and Clemmensen's reduction will be shown by both

- 13. Which of the following give violet colour in Borax bead
 - test?
 - (1) Cr3+
 - (2) Mn2+
 - (3) Co3+
 - (4) Fe2+

Answer (2)

- **Sol.** Colour of Borax bead in oxidising flame is violet for Mn2+.
- 14. Which of the following compounds have the same number of lone pair on central atom as CIF3.
 - (1) XeF-5
 - (2) XeF2
 - (3) BrF5
 - (4) I₃
- Answer (1)
- Sol. $CIF_3 \Rightarrow$

No. of lone pairs on central atom = 2



No. of lone pairs on central atom = 2 XeF2 no. of lone pairs on central atom = 3 I_3^- no. of lone pairs on central atom = 3 BrF5 no. of lone pair on central atom-1 15. **Statement 1:** For titration of oxalic acid using KMnO4, warming of acid solution is required whereas in case of Ferrous Ammonium Sulphate, it is done at room temperature.

Statement 2: Fe2+ converts to Fe3+ during titration.

- (1) Both Statement 1 and statement 2 are correct
- (2) Statement 1 is correct and statement 2 is incorrect
- (3) Statement 1 is incorrect and statement 2 is correct
- (4) Both Statement 1 and statement 2 both are

incorrect

Answer (1)

Sol. In case of oxalic acid warming of solution (50°C-60°C) is done as rate of reaction is very slow at room temperature.

Fe2++ MnO4- + 8H→+Mn2++240++ 5Fe3+

- 16.
- 17.
- 18.
- 19.
- 20.

SECTION - B Numerical Value Type Questions: This section

contains 5

Numerical based questions. The answer to each question should be rounded-off to the nearest integer. 21. fH of H(g) is

222 kJ/mol, fH of O(g) is 250 kJ/mol, fH

of H2O(g) is -248 kJ/mol. What is the value of Bond Energy of O – H bond in H2O in kJ/mol?

Answer (471)

sol.
$$H2(g)+O21(g) \rightarrow H2O(g)$$

 $2 \qquad Hf = (B.E. of react ant (B.E. of product))$

-248 = 444 + 250 - (2 BO.E - .H)2B.E. _{о–н} =942 2B.EO.-H=471kJ/mole

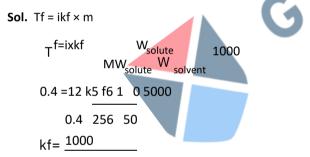
22. 70% by mass solution of HNO3 is taken having density 1.41 g/mL. Calculate the molarity of solution

Answer (16)

M_10xd Sol. M_0 10 70 1.41 = 63 = 15.66 M

23. 1g of a non-electrolyte solute (MW = 256 g/mol) is dissolved in 50g of solvent, freezing point of solution is lowered by 0.40 K. Calculate the Molal depression constant of solvent. E

Answer (5)



k f= 5.12 K kg mo-l 1

24. A compound contains 14.4% carbon, 1.2% hydrogen and 84.4% chlorine, calculate empirical formula mass of compound.

(Molar mass of C = 12, H = 1, Cl = 35.5]

Answer (84)

Sol. Let mass of compound = 100g

Mole Molar ratio

1

H = 1.2g
$$1 = 1.2$$
 1
Cl = 84.4g $\frac{84.4}{35.5} = 2.4$ 2
E.F. = CHCl2
E.F. mass = 84

25. How many of the following ions have same value of spin only magnetic moment? Ni2+, V2+, Ti2+, Sc3+, Ti3+

Answer (2)

Sol. The ions having same number of unpaired electrons

having same value of spin only magnetic moment.

 $=n(\eta+2)$ BM (n = Number of unpaired electron)

Ni2+ 3d8		1, 1, 1, 1, 1	
n = 2			
V ²⁺	3d3	1 1 1	
n = 3			
Ti2⁺	3d2	111	
n = 2			
Sc3⁺	3d0		
n = 0			
Ti3⁺	3d1	1	
n = 1			

Ni2+ and Ti²⁺ have same value of spin only magnetic moment.

MATHEMATICS

SECTION - A Multiple Choice Questions: This section contains

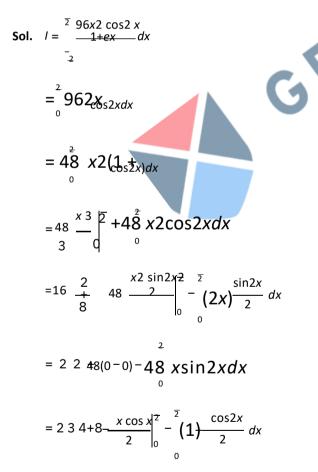
20 multiple

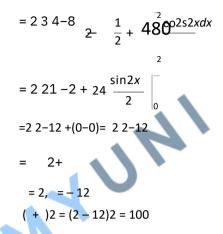
choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

Choose the correct answer :

1.	If $l = \frac{2}{1 + ex} \frac{96x2\cos 2x}{1 + ex} dx =$	(2+) ^{then} (+)2 is
	$\frac{1}{2}$	
	(1) 100	(2) 144
	(3) 169	(4) 196







Number of ways to form 5 digit numbers greater than 50000 with the use of digits 0, 1, 2, 3, 4, 5, 6, 7 such that sum of first and last digit is not more than 8, is equal to

	(1) 5119	(2) 5120	
	(3) 4607	(4) 4608	
Ansv	wer (3)		
Sol. a <u>b c d e</u>		a 5	
		a+e 8	
		not all <i>b, c, d, e</i> are zero	
		if <i>a</i> = 5.	
	(i) <i>a</i> = 5		
	5 <u>e</u>	<i>e</i> {3, 2, 1, 0}	
	834–1 ways		
	(ii) <i>a</i> = 6		
	6 <u>e</u>	<i>e</i> {2, 1, 0}	
	83 3 ways		
(iii) <i>a</i> = 7			
	7 <u>e</u>	<i>e</i> {1, 0}	
	83 2 ways		
	Total ways		
	83(2 +3 + 4) - 1 = 460	08 – 1 = 4607	

3. If the image of the point <i>P</i>	(4, 4, 3) in the		line
$\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{1}$ is Q()	,). Then (+	+) is
equal to			
(1) 7			
(2) $\frac{31}{3}$			
11			
(3) 3			
(4) 8			
Answer (2)			
Sol. <i>P</i> (4, 4, 3)			
$\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{1} =$			
Any pint of line $R(2 + 1, -1)$	+2, +1)		
$\overline{PR}: \mathbf{I}(-) + (\mathbf{I})^{\hat{P}}(\mathbf{I}) + (\mathbf{I})^{\hat{P}}(\mathbf{I})^{\hat{P}}(\mathbf{I})$	2 <i>k</i> ^		
<i>PR</i> 2,1,1 =0			
2(2 - 3) + (- 2) + 2(- 2)	= 0		
6 = 10		G	
$=\frac{5}{3}$			
5			
$R \frac{13}{3}, \frac{11}{3}, \frac{8}{3}$			
Now, Q(,,)			
$\frac{+4}{2} = \frac{13}{3}, \frac{+4}{2} = \frac{11}{3}, -\frac{1}{2}$	$\frac{+3}{2} = \frac{8}{3}$		
$=\frac{14}{3}, =\frac{10}{3}, =\frac{7}{3}$			
$+ + = \frac{14+10+7}{3} = \frac{33}{3}$	1		
4. If $tf(t)dt = x^{2}f(x)$ and $f(2)$) = 3, then <i>f</i> (6)	equals to	
(1) 1 ⁰	(2) 6		
(3) 3	(4) 2		
Answor (1)			

Answer (1)

tf(t)dt=x2f(x)Sol. Λ Differentiating both sides w.r.t 'x' xf(x) = x2f(x) + 2xf(x)x2dy _____+xy=0 $\frac{dy}{y} = \frac{-dx}{x}$ $\ln y + \ln x = \ln c$ yx = cAs f(2) = 36 = c 🧹 yx = 6Put *x* = 6 y(6) = 6*y* = 1 Option (1) is correct Let R be a relation such that $R = \{(x, y) : x, y (x + Z \text{ and } x + Z)\}$ 5. y) is even}, then the relation R is (1) Reflexive and symmetric but not transitive (2) Reflexive and transitive but not symmetric (3) Transitive only (4) Equivalence relation Answer (4) Sol. for reflexive If (x, x) ZR : *x* + *x* + 2*x* R is relexive For symmetric If (x, y) Rx + y = even*y* + *x* = even 6(*y*, *x*) *R R* is symmetric

If (x, y)Rx + y = even(y, z)Ry + z = even

$$x + 2y + Z = \text{even} \quad x + 2z = \text{even} - 2y = \text{ven} \qquad x = 2\sqrt{2} - 1 \qquad x = (2\overline{2} - 1)$$

$$x + \overline{2} = \text{even} \qquad x = 2\sqrt{2} - 1 \qquad x = (2\overline{2} - 1)$$

$$x = (2\overline{2$$

bad oranges. The variance of X is 51 49 (1) (2) 162 268 63 91 (4) (3) 206 108 Answer (2) Х 0 1 2 $\frac{{}^{7}C_{2}}{{}^{9}C_{2}} = \frac{{}^{7}C_{1} \cdot {}^{2}C_{1}}{{}^{9}C_{2}} = \frac{{}^{2}C_{2}}{{}^{9}C_{2}}$ Sol. P(X)Variance = $0^2 \cdot \frac{{}^7C_2}{{}^9C_2} + 1^2 \cdot \frac{{}^{7C_2}C_1}{{}^9C_2} + 2^2 \cdot \frac{{}^2C_2}{{}^9C_2}$ $-\frac{0.6}{\frac{2}{9}C_{2}}+\frac{1.0.6}{\frac{1}{9}C_{2}}+\frac{2.6}{\frac{2}{9}C_{2}}^{2}$ 11 $=\frac{7}{18}+\frac{4}{36}-1\frac{7}{8}+\frac{2}{36}2$ $=\frac{49}{162}$ 10. Let $f(x) = \frac{2x}{2x+\sqrt{2x+\sqrt{2x+\sqrt{k}}}}$ then 81^{f} is equal to Ans Sol (2) 4¹ (2) 4² (1) --(3) $\frac{81}{412}$ (4) 81 Answer (1) $\frac{f(x)=\frac{2^{x}}{2x+2^{1/2}}=\frac{2x}{2x+\sqrt{2}}$ Sol. $f(1-x) = \frac{21-x}{21-x+2^2} = \frac{\frac{2}{2x}}{\frac{2}{2x}+2^2_{1/2}} = \frac{2}{2+\sqrt{2}2x}$ $=\frac{\sqrt{2}}{2^{x}+\sqrt{2}}$

There are 2 bad oranges mixed with 7 good oranges and

2 oranges are drawn at random. Let X be the number of

9.

$$f(x) + f(1-x) = \frac{\sqrt{2} + 2^x}{\sqrt{2} + 2^x} = 1$$

$$\int_{k=1}^{81} \frac{k}{82} + \int_{82}^{2} \int \frac{1}{82} + \dots$$

$$\dots + \int \frac{40}{82} + \int \frac{1}{82} + \int \frac{42}{82}$$

$$+ \frac{1}{82} + \int \frac{1}{82} + \int \frac{42}{82} + \int \frac{41}{82}$$

$$= \int \frac{1}{82 + f} \frac{81}{82} + \int \frac{1}{82} + \int \frac{41}{82} + \int \frac{41}{82} + \dots$$

$$= \int \frac{40}{82 + 42} + \int \frac{41}{82} + \int \frac{41}{82} + \int \frac{1}{82} + \int$$

 $= 3a_{100} - 99$

Using
$$\frac{3}{2} \frac{100}{2} = \frac{a 100 - 1}{2}$$

 $a^{100 = 2} 3 \frac{2}{3} \frac{100}{1 + 1}$
12. Let k1 and k2 be two randomly selected natural numbers.
The probability that $\binom{1}{2}$ k is non-zero is (where $i = \sqrt{-1}$)
(1) $\frac{1}{2}$ (2) $\frac{3}{4}$
(3) $\frac{1}{4}$ (4) $\frac{1}{6}$
Answer (2)
Sol. $(i)k1 + (i)2k$ is non zero
 $k_1 : 41 + r1, r1 (0, 1, 2, 3)$
 $k_2 : 42 + r2, r2 (0, 1, 2, 3)$
The pairs to get zero will be
 $(1, -1), (i, -1)$
 $(i)(1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (1, -1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(1, (-1) pair$
 $(r1, r2) \cdot ((2, 0), (0, 2))$
 $(i)(2)$
 $k = \frac{1 - \frac{1}{16} = \frac{12}{16} = \frac{3}{4}$
Ty $ABC, A(4sin, 4cos) J, B(-2cos, 0) and C(2, 2sin) . If$
locus of centroid is $(3x - 2)2 + (3y)2 =$, then is
 $(1) 20$
 $(2) 4$
 $(3) 16$
 $(3) 12$
Answer (1)
Answer (1)

$$= \frac{(biai)}{a_{\overline{a}}} 2$$

$$S_{i+1} = Si(1-ei2)$$

$$Si+1 = Si(1-ei2)$$

$$Si+1 = Si(1-ei2)$$

$$S_{\overline{i}}=6, S = 2=6 = \frac{4}{9}S^{3}=6 = \frac{4}{9}z^{2}$$

$$K \cong T = \frac{6}{1-4} = -\frac{54}{5}$$

$$\frac{5}{K} \frac{5i}{2} = \frac{5}{5} = 54$$
15. Let $z1 = \sqrt{3+2} \sqrt{2i}$ and $\sqrt{3}|z1| = |z2|$ and $arg(z2) = arg(z1) + 6$, then the area of triangle wi vertices $z1, z2$ and origin is (in sq. units)
(1) $\frac{11\sqrt{3}}{4}$
(3) $\frac{11}{4}$
(4) $\frac{2\sqrt{2}}{3}$
(3) $\frac{11}{4}$
(4) $\frac{2\sqrt{2}}{3}$
(3) $\frac{11}{4}$
(4) $\frac{2\sqrt{2}}{3}$
(5).
$$Area of OAB = \frac{1}{2} \sqrt{11} \sqrt{\frac{33\sin 6}{-1}}$$

$$= \frac{11\sqrt{3}}{4}$$
 square units

2*x, x* 0 16. Let $f(x) = \min(1 x + \frac{1}{2}, 1 + 2[x]), 0x_2$ 5,x2 If is the number of points of discontinuity and is the number of points of non-differentiability, then (+) is equal to (where [.] denote greatest integer function) (1) 6 (2) 5 (3) 4 (4) 8 Answer (1) 2x, x 0 **Sol.** $f(x) = \min(+1_x + [x], 1 + 2[x]), Q_2$ x 2 5 $1 + x + [x] = 1 + \{x\} + 2[x]$ Since {*x*} 0 *x R* 1 + x + x + x = 1 + 2[x]0 2*x,x x* 0 $f(x) = \pm 2[x], 0$ 5 x 2 5-3 1 2 1 Number of discontinuity = 3 = 3 Number of point of non-differentiability = 3 = 3 **17.** If $=1 + \binom{6}{n-1} 3^{n-1} 12C2n-1$, then distance of point $(12, \sqrt{3})$ rom the line x- $\sqrt{3}y+100=0$ is, 109 (1) — 2 (2) 55 (3) 54 Answ(24)(109

Sol. =1+
$$\int_{r=1}^{6} \sqrt{9^2} \int_{1}^{r_1} 12Q_{n-1}$$

=1+ $\int_{\sqrt{3}i}^{-1} \sqrt{9^2} \int_{1}^{r_1} 12Q_{n-1}$
=1+ $\int_{\sqrt{3}i}^{-1} \int_{r=1}^{r_1} (\sqrt{3}i) 2r + \frac{4}{22r-1}$
Let $\sqrt{3}i \times x$
=1+ $\frac{1}{\sqrt{3}i} \int_{r=1}^{1} \frac{1}{2}C2r - x12r - 1$
=1+ $\frac{1}{\sqrt{3}i} \int_{r=1}^{1} \frac{1}{2}C2r - x12r - 1$
=1+ $\frac{1}{\sqrt{3}i} \int_{r=1}^{1} \frac{1}{2}C2r - x12r - 1$
Let $(1+x)12 \sum (2x^{11} + \frac{12}{2}C3 \times 3 + 1^2C_5 \times 5^{5} + \frac{1}{2}C_{11} \times 1^{11}}{12} + \frac{1}{2}C_{2}^{2} \times \frac{1}{2} + \frac{12}{2}C_{11} \times 1^{12} + \frac{12}{2}C_{11} \times 1^{12} + \frac{12}{2}C_{12} \times \frac{1}{2}C_{12} \times \frac{$

...(1)

40

Tr r <u>₹</u>0

$$= 100 \frac{40}{2} [2a+39d] - \frac{19}{2} [2a+18d]$$

$$= 100 \frac{40}{2} 41d \frac{19}{2} 20d$$

$$= 100 \frac{40}{2} \frac{41}{2025} - \frac{1920}{2025}$$

$$= 126$$

$$22.$$

$$23.$$

$$24.$$

$$25.$$

