

## JEE-Main-28-06-2022-Shift-2 (Memory Based)

### Physics

**Question:** A body of mass 2 kg moving with speed 4 m/s encounters a region from  $x=0.5$  to  $x=1.5$  where  $F=-kx$ . Find the final velocity of the body. Given  $k=12$

**Options:**

- (a) 5 m/s
- (b) 2 m/s
- (c) 4 m/s
- (d) 6 m/s

**Answer: (b)**

**Solution:**

$$a = -\frac{kx}{m}$$

$$v \frac{dv}{dx} = -\frac{12x}{m}$$

$$v dv = -\frac{12x}{m} dx$$

$$\left[ \frac{v^2}{2} \right]_4^v = -\frac{12}{2} \left( \frac{x^2}{2} \right)_{0.5}^{1.5}$$

$$\frac{v^2}{2} - \frac{16}{2} = -6 \left( \frac{2.25 - 0.25}{2} \right)$$

$$v^2 - 16 = -12$$

$$v^2 = -12 + 16 = 4$$

$$v = 2 \text{ m/s}$$

**Question:** A ladder rest slantly with its base 3 m from the floor The wall is frictionless. Length of ladder is  $\sqrt{34}$  m Mass of ladder is 10kg. Find the ratio of reaction force by wall to reaction force by floor on ladder

**Options:**

- (a) 3/10
- (b) 9/10
- (c) 5/10
- (d) 7/10

**Answer: (a)**

**Solution:**

$$N_G = mg = 10 \times 10 = 100 \text{ N}$$

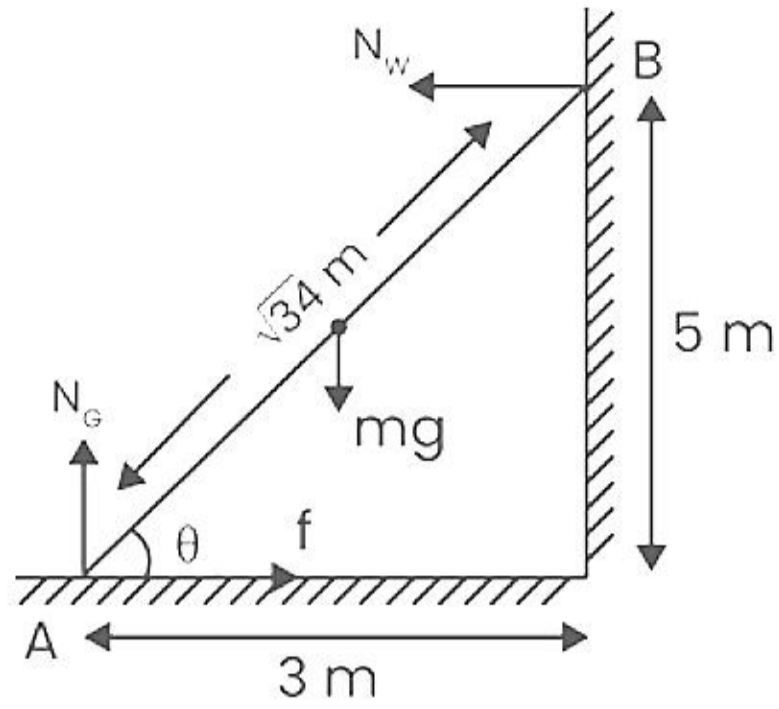
Taking moments about bottom point of ladder

$$N_w \times 5 = mg \times \frac{\sqrt{34}}{2} \cos \theta$$

$$N_w \times 5 = 100 \times \frac{\sqrt{34}}{2} \times \frac{3}{\sqrt{34}}$$

$$N_w = 30 \text{ N}$$

$$\frac{N_w}{N_G} = \frac{30}{100} = \frac{3}{10}$$



**Question:** If all the oxygen molecules dissociate into atoms and temperature is doubled then  $V_{rms}$  becomes \_\_\_ times the original

**Options:**

- (a) 4
- (b) 3
- (c) 2
- (d) None of these

**Answer:** (c)

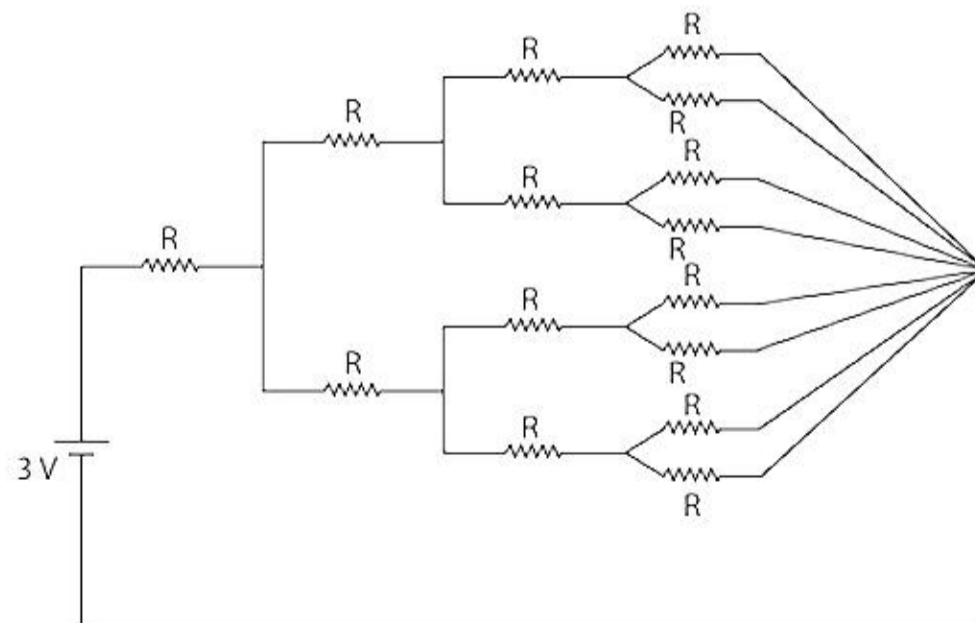
**Solution:**

$$V_{rms} = \sqrt{\frac{3RT}{M}}$$

$$(V_{rms})_0 = \sqrt{\frac{3R(2T)}{M/2}}$$

$$= 2\sqrt{\frac{3RT}{M}} = 2V_{rms}$$

**Question:**  $I = \frac{a}{5}$ ,  $R = 1\Omega$ , Find a.



**Options:**

- (a) 14
- (b) 8
- (c) 12
- (d) 2

**Answer: (b)**

**Solution:**

Starting from right most resistors

$$R \parallel R$$

$$R_A = \frac{R}{2} \Omega$$

$$R + \frac{R}{2} = \frac{3R}{2} \Omega$$

$$\frac{3R}{2} \parallel \frac{3R}{2}$$

$$R_B = \frac{3R}{4} \Omega$$

$$R + \frac{3R}{4} = \frac{7R}{4} \Omega$$

$$\frac{7R}{4} \parallel \frac{7R}{4}$$

$$R_C = \frac{7R}{8} \Omega$$

$$R_{eq} = R + \frac{7R}{8} = \frac{15R}{8} \Omega = \frac{15}{8} \Omega$$

$$I = \frac{V}{R} = \frac{3}{15} \times 8 = \frac{8}{5} A$$

$$\Rightarrow a = 8$$

**Question:** In YDSE slab of thickness  $t$  and RI 1.5 is inserted in front of one of the slits. As a result intensity at the central maxima remains the same. What is the minimum value of thickness required?

**Options:**

- (a)  $2\lambda$
- (b)  $4\lambda$
- (c)  $8\lambda$
- (d) None of these

**Answer: (a)**

**Solution:**

$$\text{Shift} = \frac{D}{d}(\mu - 1)t$$

$$= \frac{D\lambda}{d}$$

$$\Rightarrow t = \frac{\lambda}{\mu - 1} = 2\lambda$$

**Question:** If resistance of a resistor is  $2 \Omega$  at  $10^\circ\text{C}$  and it is  $3 \Omega$  at  $30^\circ\text{C}$  find the temperature coefficient of resistance

**Options:**

- (a)  $0.24 \times 10^{-2} / ^\circ\text{C}$

(b)  $4.4 \times 10^{-2} / ^\circ C$

(c)  $2.5 \times 10^{-2} / ^\circ C$

(d) None of these

**Answer: (c)**

**Solution:**

$$R_t = R_0 (1 + \alpha \Delta T)$$

$$3 = 2(1 + \alpha(30 - 10))$$

$$3 = 2 + 40\alpha$$

$$1 = 40\alpha$$

$$\alpha = \frac{1}{40} = 2.5 \times 10^{-2} / ^\circ C$$

**Question:** Particle moves along the straight line such that it moves  $1/3^{\text{rd}}$  distance with speed  $v_1$  the next  $1/3^{\text{rd}}$  distance with speed  $v_2$  and remaining  $1/3^{\text{rd}}$  distance with speed  $v_3$ . Then its average speed throughout motion is

**Options:**

(a)  $\frac{v_1 v_2 + v_2 v_3 + v_3 v_1}{v_1 + v_2 + v_3}$

(b)  $\frac{v_1 v_2 v_3}{v_1 v_2 + v_2 v_3 + v_3 v_1}$

(c)  $\frac{v_1 + v_2 + v_3}{3}$

(d)  $\frac{3v_1 v_2 v_3}{v_1 v_2 + v_2 v_3 + v_3 v_1}$

**Answer: (d)**

**Solution:**

Average speed = total distance covered / total time taken

Let the total distance =  $3x$

Time taken to cover first one third ( $x$ ) =  $t_1 = \frac{x}{v_1}$

Time taken to cover second one third ( $x$ ) =  $t_2 = \frac{x}{v_2}$

Time taken to cover third one third ( $x$ ) =  $t_3 = \frac{x}{v_3}$

$$\text{Average speed} = \frac{3x}{\frac{x}{v_1} + \frac{x}{v_2} + \frac{x}{v_3}}$$

$$= \frac{3x}{x \left( \frac{v_3 v_2 + v_1 v_3 + v_1 v_2}{v_1 v_2 v_3} \right)}$$

$$= \frac{3v_1 v_2 v_3}{v_1 v_2 + v_2 v_3 + v_1 v_3}$$

**Question:** A water drop of radius  $1 \mu\text{m}$  in falls through air. Force of buoyancy and density of air is negligible. If the coefficient of viscosity of air is  $2.0 \times 10^{-5} \text{kgm}^{-1}\text{s}^{-1}$ . Find terminal velocity of water drop.

**Options:**

- (a)  $3.4 \times 10^{-4} \text{m} / \text{J}$
- (b)  $2.4 \times 10^{-4} \text{m} / \text{J}$
- (c)  $1.4 \times 10^{-4} \text{m} / \text{J}$
- (d)  $1.1 \times 10^{-4} \text{m} / \text{J}$

**Answer: (d)**

**Solution:**

$$V_T = \frac{2}{9} \frac{r^2 (\gamma - \rho) g}{\eta}$$

$$V_T = \frac{2}{9} \frac{r^2 \gamma g}{\eta}$$

$$V_T = \frac{2}{9} \times \frac{(1 \times 10^{-6})^2 \times 1 \times 10^3 \times 10}{2.0 \times 10^{-5}}$$

$$V_T = \frac{1}{9} \times 10^{-12+3+1+5} = 1.1 \times 10^{-4} \text{m} / \text{J}$$

**Question:** Two capacitors of capacities  $5 \mu\text{F}$  and  $10 \mu\text{F}$  connected and the switch is kept open. Initially potential on  $5 \mu\text{F}$  capacitor is  $30\text{V}$  and  $10 \mu\text{F}$  capacitor is uncharged. Find the charge on the  $10 \mu\text{F}$  capacitor once the switch is closed.

**Options:**

- (a)  $300 \mu\text{C}$
- (b)  $100 \mu\text{C}$
- (c)  $200 \mu\text{C}$
- (d)  $400 \mu\text{C}$

**Answer: (b)**

**Solution:**

$$V = \frac{C_1 V_1 + C_2 V_2}{C_1 + C_2}$$

$$V = \frac{5 \times 30}{5 + 10} = 10\text{V}$$

$$Q = 10 \times 10 = 100 \mu\text{C}$$

**Question:** Two bodies of equal mass has force of attraction  $F$ , then find the the force of attraction when one third of mass of one body is transferred to another

**Options:**

- (a)  $\frac{1}{9} F$
- (b)  $\frac{8}{9} F$
- (c)  $\frac{5}{9} F$

(d)  $\frac{7}{9}F$

**Answer: (b)**

**Solution:**

$$F_i = \frac{Gm^2}{r^2}$$
$$F_f = \frac{G\left(m - \frac{1}{3}m\right)\left(m + \frac{1}{3}m\right)}{r^2}$$
$$= G \frac{8m^2}{9r^2} = \frac{8}{9}F_i = \frac{8}{9}F$$

**Question:** A coil is placed in a time varying magnetic field. If the no. of turns are halved and the radius of wire is doubled. (Assume the coil to be short circuited) Then the power dissipated:

**Options:**

(a)  $4P_i$

(b)  $1P_i$

(c)  $7P_i$

(d)  $3P_i$

**Answer: (a)**

**Solution:**

$$N_i = n \quad N_f = \frac{n}{2}$$
$$r_i = r \quad r_f = 2r$$

Total length of wire =  $n(2\pi R)$  = where R is radius of loop finally if n becomes half, radius of loop has to double

$$\therefore \text{New emf} = -\left(\frac{n}{2}\right)\left(\pi(2R)^2\right)\frac{dB}{dt}$$

$$\text{New Power} = \frac{\left[-\frac{n}{2}\pi(4R^2)\frac{dB}{dt}\right]^2}{\rho \pi(2r)^2}$$

$$\text{Old power} = \frac{\left[-n\pi(R)^2\frac{dB}{dt}\right]^2}{\rho \pi r^2} \therefore P_f = 4P_i$$

**Question:**  $K_1$  and  $K_2$  are  $KE_{\max}$  of  $\lambda_1$  and  $\lambda_2$  Falling a metal If  $\lambda_1 = 3\lambda_2$  Find relation of  $K_1$  and  $K_2$

**Options:**

(a)  $3K_1 < K_2$

(b)  $4K_1 < K_2$

(c)  $5K_1 < K_2$

(d)  $2K_1 < K_2$

**Answer: (a)**

**Solution:**

Kinetic energy of the photoelectrons  $K = \frac{hc}{\lambda} - \phi$  where  $\phi$  is the work function of the metal

$\therefore$  For wavelength  $\lambda_1$   $K_1 = \frac{hc}{\lambda_1} - \phi \dots (1)$

For wavelength  $\lambda_2$ ,  $K_2 = \frac{hc}{\lambda_2} - \phi \dots (2)$

Given:  $\lambda_1 = 3\lambda_2$

$\therefore$  Equation (1) becomes  $K_1 = \frac{hc}{3\lambda_2} - \phi \dots (3)$

From (2) - (3), we get  $K_2 - K_1 = \frac{hc}{\lambda_2} - \frac{hc}{3\lambda_2}$

$$K_2 - K_1 = \frac{2}{3} \frac{hc}{\lambda_2} \Rightarrow \frac{hc}{\lambda_2} = \frac{3}{2} (K_2 - K_1)$$

Put this in (2),  $K_2 = \frac{3}{2} (K_2 - K_1) - \phi$

$$\Rightarrow K_2 - 3K_1 = 2\phi$$

As  $\phi > 0 \Rightarrow K_2 - 3K_1 > 0$

Thus  $K_1 < \frac{K_2}{3}$

**Question:** EM wave is moving in +x direction. If amplitude of electric field is  $E_0 = 60 \text{ N/C}$  which is oscillating in y direction, then find the equations of E and B

**Options:**

(a)  $E = 60 \sin(kx - \omega t) \hat{i}$   
 $B = 2 \times 10^{-7} \sin(k_x - \omega t) \hat{j}$

(b)  $E = 60 \sin(kx - \omega t) \hat{k}$   
 $B = 2 \times 10^{-7} \sin(k_x - \omega t) \hat{j}$

(c)  $E = 60 \sin(kx - \omega t) \hat{j}$   
 $B = 2 \times 10^{-7} \sin(k_x - \omega t) \hat{k}$

(d)  $E = 60 \sin(kx - \omega t) \hat{k}$   
 $B = 2 \times 10^{-7} \sin(k_x - \omega t) \hat{i}$

**Answer: (c)**

**Solution:**

$$E_0 = 60$$

$$\therefore B_0 = \frac{E_0}{C} = \frac{60}{3 \times 10^8} = 2 \times 10^{-7}$$

Since wave is moving in +x-dir

$$E = 60 \sin(kx - \omega t) \hat{j}$$

$$B = 2 \times 10^{-7} \sin(kx - \omega t) \hat{k}$$

**Question:** A solenoid is filled with material of susceptibility  $2 \times 10^{-7}$  Fractional change in field intensity compared to the case when air was present inside instead of material

**Options:**

(a)  $= 2 \times 10^{-5} \%$

(b)  $= 4 \times 10^{-5} \%$

(c)  $= 3 \times 10^{-5} \%$

(d)  $= 5 \times 10^{-5} \%$

**Answer: (a)**

**Solution:**

$$X_m = \frac{M}{H}$$

It is already fractional change in the magnetic induction due to the medium.

$\therefore$  % age change

$$= X_m \times 100$$

$$= 2 \times 10^{-7} \times 100$$

$$= 2 \times 10^{-5} \%$$

**Question:** Half life of radioactive material is 200 days. Find percent of substance remaining in 83 days

**Options:**

(a) 65%

(b) 55%

(c) 44%

(d) 75%

**Answer: (d)**

**Solution:**

$$N = N_0 \left( \frac{1}{2} \right)^{t/T}$$

$T =$  half life

$t =$  time elapsed

$$\therefore N = N_0 \left( \frac{1}{2} \right)^{\frac{83}{200}}$$

$$\frac{N}{N_0} \times 100 = \left( \frac{1}{2} \right)^{\frac{83}{200}} \times 100$$

$$\approx 75\%$$



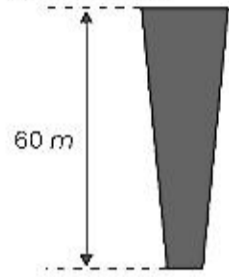
**Question:** Water falls at a rate of 600 kg/s from a height of 60 m as shown. How many bulbs of capacity 100 W each will glow from the energy produced at the bottom of the fall. Assume full conversion of energy of falling water and all bulbs glowing at 100 W each.

**Options:**

- (a) 25
- (b) 50
- (c) 3600
- (d) 1000

**Answer:** (c)

**Solution:**



Potential energy loss per second

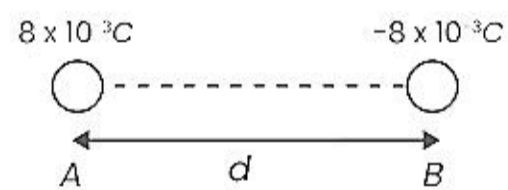
$$= 600(10)(60) \text{ J/s}$$

$$= 36 \times 10^4 \text{ J/s} = 36 \times 10^4 \text{ W}$$

Each bulb consumes 100 W,

$\therefore$  Total no of bulbs which can glow is 3600.

**Question:** Two opposite charges are placed at a distance  $d$  as shown. Electric field strength at mid point is  $6.4 \times 10^4 \text{ N/C}$ . Then the value of  $d$  is



**Options:**

- (a)  $20\sqrt{10}$
- (b)  $50\sqrt{10}$
- (c)  $30\sqrt{10}$
- (d)  $60\sqrt{10}$

**Answer:** (c)

**Solution:**

$$E = \frac{2kQ}{(d/2)^2}$$

$$6.4 \times 10^4 = \frac{2 \times 9 \times 10^9 \times 8 \times 10^{-3}}{d^2 / 4}$$

$$\therefore d^2 = 9000$$

$$d = 30\sqrt{10}$$

**Question:** In series RLC circuit voltage across capacitance and inductance is twice that of resistance. If  $R = 50\Omega$ ,  $V = 220\text{V}$ ,  $f = 50\text{Hz}$ . If  $L = 1/k\pi$  then value of  $k$  is (in m H)

**Options:**

(a)  $11^{-2}$

(b)  $10^{-2}$

(c)  $12^{-2}$

(d)  $15^{-2}$

**Answer: (b)**

**Solution:**

$$V_L = 2V_R$$

$$\omega L = 2R$$

$$L = \frac{2R}{\omega} = \frac{2R}{2\pi f} = \frac{2(s)}{2\pi(50)}$$

$$= \frac{1}{10\pi} = \frac{1}{10\pi} \times \frac{10^{-3}}{10^{-3}}$$

$$\therefore K = 10^{-2}$$

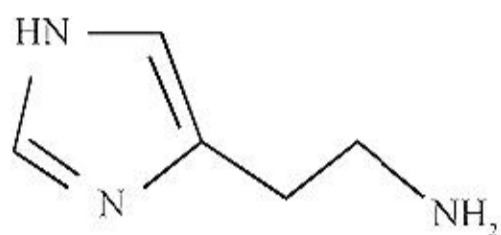
## JEE-Main-28-06-2022-Shift-2 (Memory Based)

### Chemistry

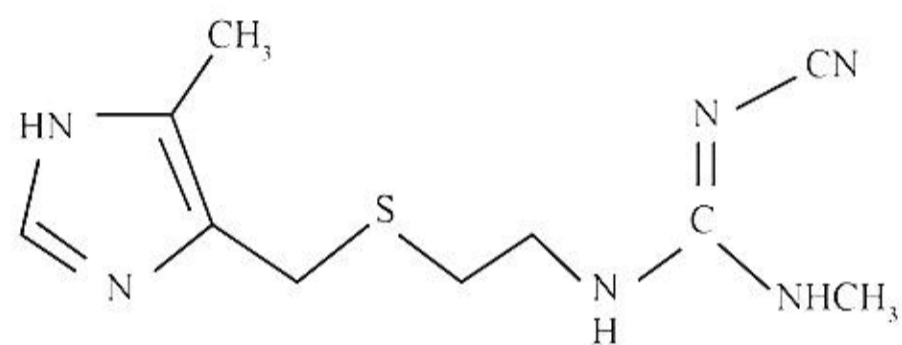
**Question:** Which of the following is the structure of Tagamet?

**Options:**

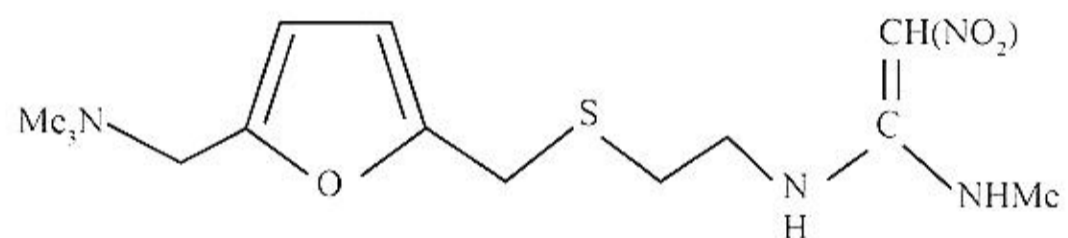
(a)



(b)



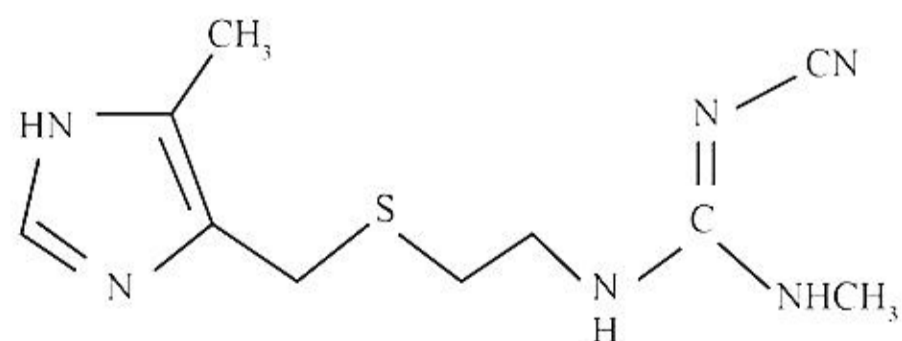
(c)



(d) None of these

**Answer:** (b)

**Solution:**



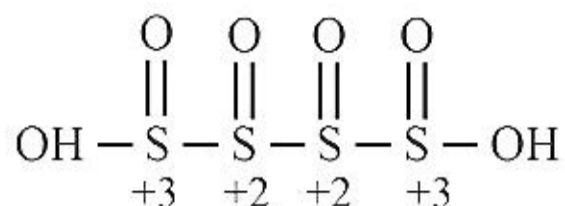
**Question:** In which of oxyacids of sulphur both sulphur have different oxidation state.

**Options:**

- (a)  $\text{H}_2\text{S}_4\text{O}_6$
- (b)  $\text{H}_2\text{S}_2\text{O}_8$
- (c)  $\text{H}_2\text{S}_2\text{O}_7$
- (d) All of these

**Answer:** (a)

**Solution:**



**Question:** Match the following.

Column-I	Column-II
(i) Positively charged	(A) Gel
(ii) Negatively charged	(B) Starch
(iii) Macromolecular starch	(C) CuS
(iv) Cheese	(D) $\text{Fe}_2\text{CO}_3 \cdot x \text{H}_2\text{O}$

**Options:**

- (a) (i)  $\rightarrow$  (D); (ii)  $\rightarrow$  (C); (iii)  $\rightarrow$  (B); (iv)  $\rightarrow$  (A)
- (b) (i)  $\rightarrow$  (B); (ii)  $\rightarrow$  (C); (iii)  $\rightarrow$  (A); (iv)  $\rightarrow$  (D)
- (c) (i)  $\rightarrow$  (C); (ii)  $\rightarrow$  (B); (iii)  $\rightarrow$  (D); (iv)  $\rightarrow$  (A)
- (d) (i)  $\rightarrow$  (D); (ii)  $\rightarrow$  (A); (iii)  $\rightarrow$  (D); (iv)  $\rightarrow$  (B)

**Answer:** (a)

**Solution:**

- (i) Positively charged  $\Rightarrow \text{Fe}_2\text{CO}_3 \cdot x \text{H}_2\text{O}$
- (ii) Negatively charged  $\Rightarrow \text{CuS}$
- (iii) Macromolecular starch  $\Rightarrow$  Starch
- (iv) Cheese  $\Rightarrow$  Gel

**Question:** A compound has 8% H, 70% C, 16% N, Molecular Mass is 160. Find the formula of compound.

**Options:**

- (a)  $\text{C}_{12}\text{H}_{16}\text{N}_2$
- (b)  $\text{C}_{12}\text{H}_{18}\text{N}_2$

(c)  $C_{11}H_{16}N$

(d)  $C_{12}H_{15}N$

**Answer:** (a)

**Solution:**

Compound contain 8% H, 70% C and 16% N

$$\text{No. of moles of C} = \frac{70}{12} = 5.8 \approx 6$$

$$\text{No. of moles of H} = \frac{8}{1} = 8$$

$$\text{No. of moles of N} = \frac{16}{14} = 1 : 1 \approx 1$$

Mole ratio C : H : N = 6 : 8 : 1

Empirical formula =  $C_6H_8N$

Molecular mass = 160

Empirical formula mass =  $12 \times 6 + 2 \times 1 + 14 = 94$

$n = 2$

Formula of compound =  $(C_6H_8N)_2 = C_{12}H_{16}N_2$

**Question:** What is correct about photochemical smog?

**Options:**

(a) It is reducing in nature

(b) It occurs in humid conditions

(c) It is formed due to the action of sunlight on Hydrocarbons

(d) All of these

**Answer:** (c)

**Solution:** Photochemical smog results from the action of sunlight on hydrocarbons.

**Question:** Which of the following is basic oxide?

**Options:**

(a) CaO

(b)  $SiO_2$

(c)  $\text{Al}_2\text{O}_3$

(d) NO

**Answer:** (a)

**Solution:** CaO - basic oxide

$\text{SiO}_2$  - acidic oxide,

$\text{Al}_2\text{O}_3$  - Amphoteric oxide,

NO - neutral oxide

**Question:** An ideal gas is stored in a vessel of volume 416 ml at temperature 300 K and Pressure 1.5 atm. What is the mass of gas? (Molecular mass of gas 100g/mol)

**Options:**

(a) 3.32 g

(b) 2.53 g

(c) 3.01 g

(d) 1.92 g

**Answer:** (b)

**Solution:**

$$PV = \frac{w}{M}RT$$

$$\therefore w = \frac{PVM}{RT} = \frac{1.5 \times 0.416 \times 100}{0.0821 \times 300} = 2.53 \text{ g}$$

**Question:** 2.5 g of protein taken and made 500 ml of solution. Osmotic pressure of solution is  $5.03 \times 10^{-3}$  at 300 K. Find the no. of glycine unit.

**Options:**

(a)  $1.9 \times 10^{16}$  units

(b)  $2.8 \times 10^{15}$  units

(c)  $1.2 \times 10^{15}$  units

(d)  $2.2 \times 10^{10}$  units

**Answer:** (c)

**Solution:**

$$\text{Osmotic pressure } (\pi) = \left( \frac{n_2}{V} \right) RT$$

$$\pi V = \frac{w_2 RT}{M_2}$$

$$M_2 = \frac{w_2 RT}{\pi} = \frac{2.5 \times 0.0821 \times 300}{5.03 \times 10^{-8}} = 12.2 \times 10^8 \text{ g}$$

$$\text{No. of glycine units} = \frac{2.5 \times 6.023 \times 10^{23}}{12.2 \times 10^8} = 1.2 \times 10^{15} \text{ units}$$

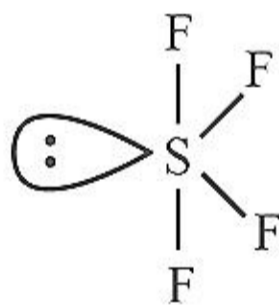
**Question:** In SF<sub>4</sub> what is the bond angle?

**Options:**

- (a) 90°, 120°
- (b) 90°, 117°
- (c) 89°, 120°
- (d) 89°, 117°

**Answer:** (d)

**Solution:**



**Question:** The volume of 0.01 M KMnO<sub>4</sub> solution which can oxidize 20 ml of 0.05 M Mohr salt solution in acidic medium is

**Options:**

- (a) 10 ml
- (b) 20 ml
- (c) 30 ml
- (d) 40 ml

**Answer:** (b)

**Solution:**  $M_1 V_1 Z_1 = M_2 V_2 Z_2$

$$0.01 \times V_1 \times 5 = 0.05 \times 20 \times 1$$

$$V_1 = 20 \text{ ml}$$

**Question:** In the buffer solution, having  $\text{pH} = 4$  and  $\text{pK}_a = 1.3 \times 10^{-5}$ , find the ratio of salt/acid is

**Options:**

(a)  $10^{-0.8}$

(b) 0.1

(c)  $10^{0.8}$

(d)  $10^{-2.1}$

**Answer:** (a)

**Solution:**  $\text{pK}_a = -\log(1.3 \times 10^{-5})$

$$= 5 - \log 1.3$$

$$= 4.85$$

$$\text{pH} = \text{pK}_a + \log \frac{[\text{salt}]}{[\text{acid}]}$$

$$4 = 4.85 - \log \frac{[\text{salt}]}{[\text{acid}]}$$

$$-0.85 = \log \frac{[\text{salt}]}{[\text{acid}]}$$

$$\frac{\text{salt}}{\text{acid}} = 10^{-0.85}$$

**Question:** Nitration of aniline with  $\text{HNO}_3 + \text{H}_2\text{SO}_4$  gives

**Options:**

(a) p-nitroaniline

(b) m-nitroaniline

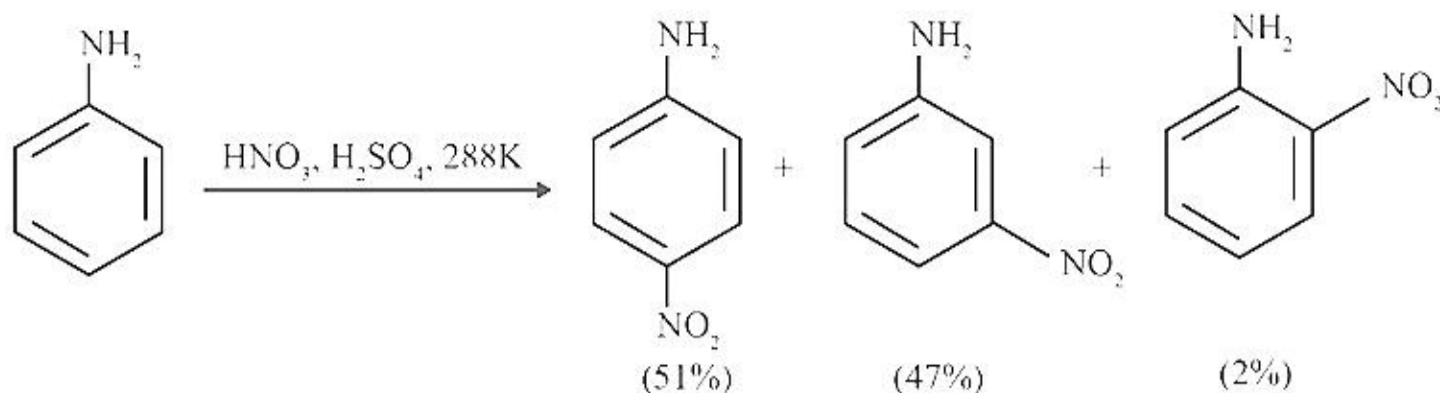
(c) o-nitroaniline

(d) All of these

**Answer:** (d)

**Solution:**





**Question: Assertion:** Natural form of rubber is cis 1,4 polyisoprene

**Reason:** There are weak vander waals forces giving its coiled structure

**Options:**

- (a) Both assertion and reason are true, reason is correct explanation of assertion.
- (b) Both assertion and reason are true, but reason is not a correct explanation of assertion.
- (c) Assertion is true, but reason is false
- (d) Assertion is false, but reason is true

**Answer:** (b)

**Solution:** Both assertion and reason are true, but reason is not correct explanation of A

Natural rubber may be considered as a linear polymer of isoprene (2-methyl-1, 3-butadiene) and is also called as cis-1, 4-polyisoprene.

The cis-polyisoprene molecule consists of various chains held together by weak van der Waals interactions and has a coiled structure. Thus. It can be stretched like a spring and exhibits elastic properties.

**Question:** In extraction of copper FeO and FeSiO<sub>3</sub> are respectively

**Options:**

- (a) slag, gangue
- (b) gangue, slag
- (c) both are slag
- (d) both are gangue

**Answer:** (b)

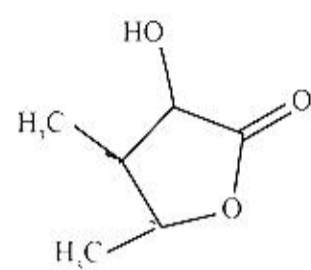
**Solution:** FeO is gangue and SiO<sub>2</sub> is flux to form slag FeSiO<sub>3</sub>.



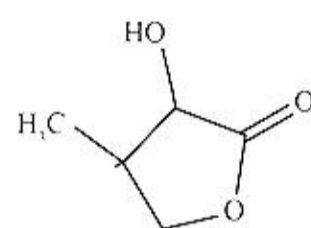
**Question:** Isobutyraldehyde reacts with  $K_2CO_3$  and formaldehyde to give A. A reacts with HCN to give B. Hydrolysis of B gives a stable carboxylic acid C. What is C?

**Options:**

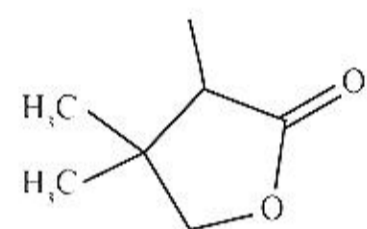
(a)



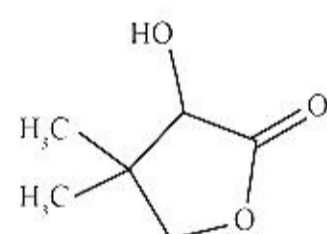
(b)



(c)

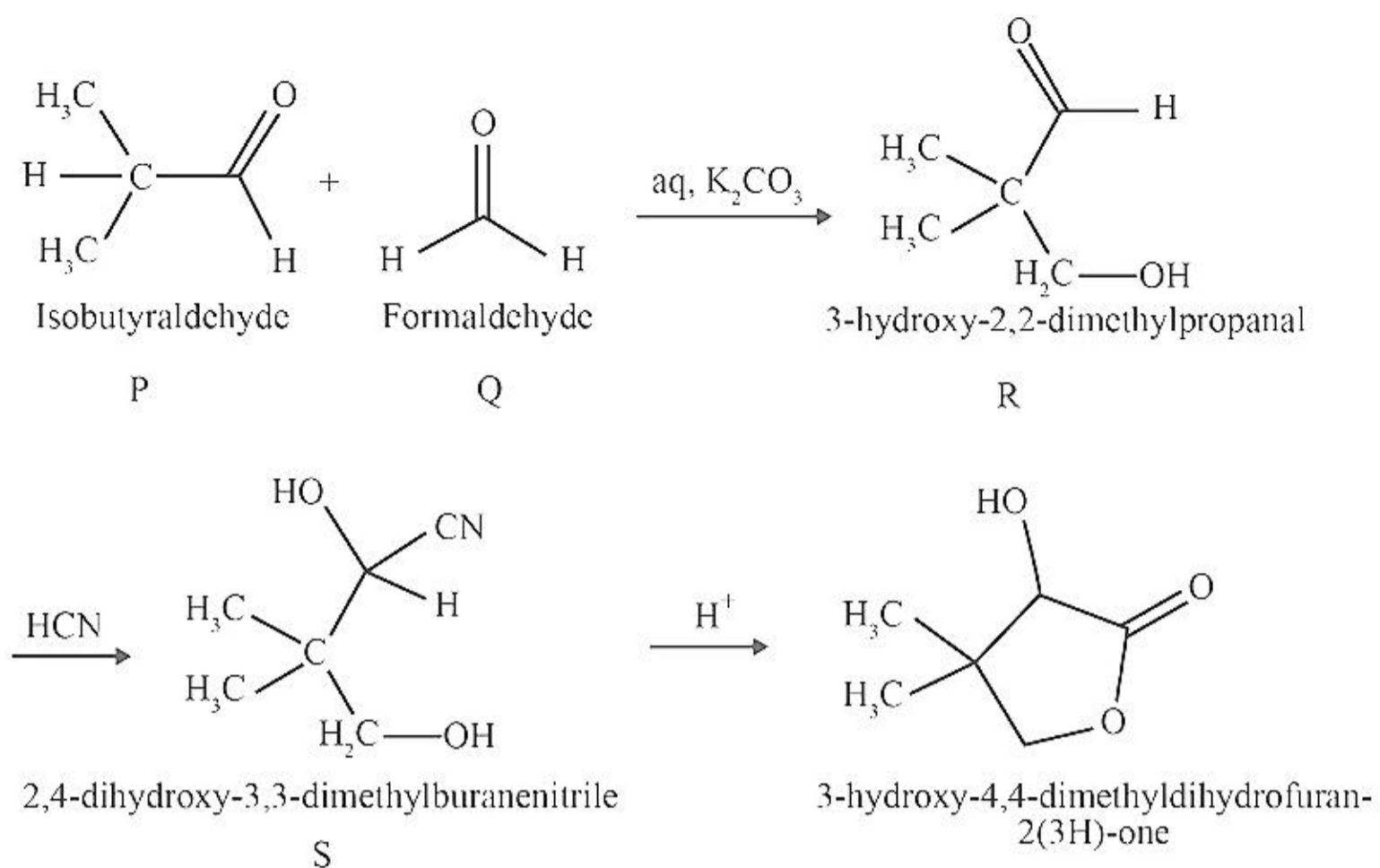


(d)



**Answer:** (d)

**Solution:**



**Question:** The isotopes of Hydrogen differ in the following property

**Options:**

- (a) Electronic configuration
- (b) No of protons
- (c) Atomic number
- (d) Atomic mass

**Answer:** (d)

**Solution:** The three isotopes of hydrogen differ in mass numbers which are 1, 2 and 3 respectively known as protium, deuterium and tritium.

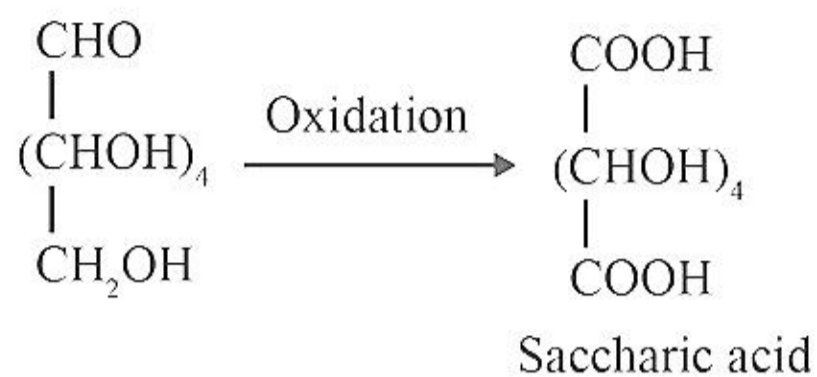
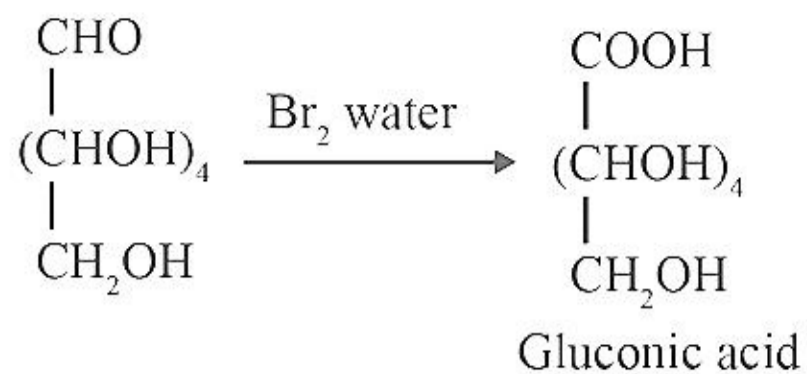
**Question:** X reacts with  $\text{Br}_2/\text{H}_2\text{O}$  to give gluconic acid and reacts with  $\text{HNO}_3$  to give saccharic acid. Name X

**Options:**

- (a) Maltose
- (b) Starch
- (c) Fructose
- (d) Glucose

**Answer:** (d)

**Solution:**



**Question:** Comparison of KE for wavelengths  $\lambda$  and  $3\lambda$ , keeping work function constant

**Options:**

(a)  $K.E_2 = 9 K.E_1$

(b)  $K.E_2 = \frac{1}{9} K.E_1$

(c)  $K.E_2 = 3 K.E_1$

(d)  $K.E_2 = \frac{1}{3} K.E_1$

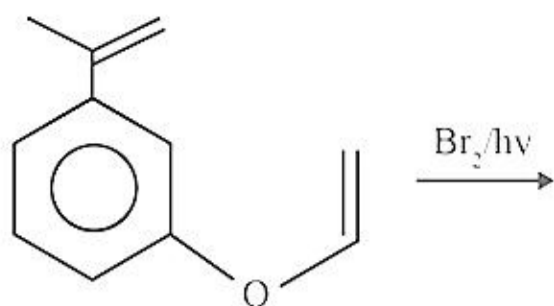
**Answer:** (b)

**Solution:**  $\lambda = \frac{h}{\sqrt{2m K.E}}$

$$\frac{\lambda}{3\lambda} = \frac{\frac{h}{\sqrt{2m K.E_1}}}{\frac{h}{\sqrt{2m K.E_2}}}$$

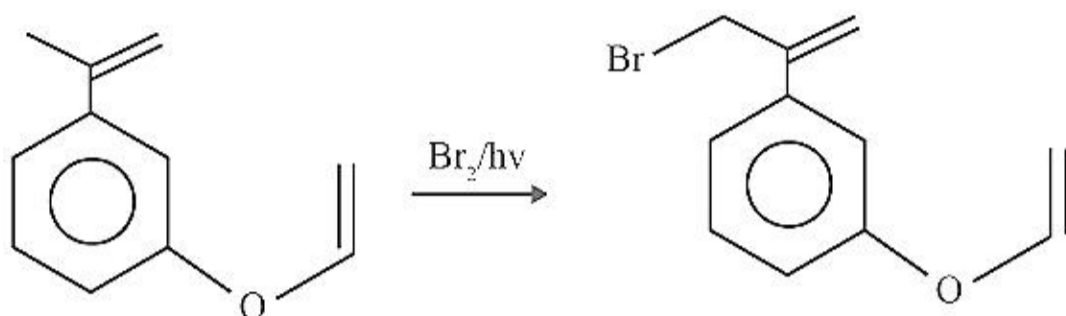
$$K.E_2 = \frac{1}{9} K.E_1$$

**Question:** Product will have how many Br.....



**Answer:** 1.00

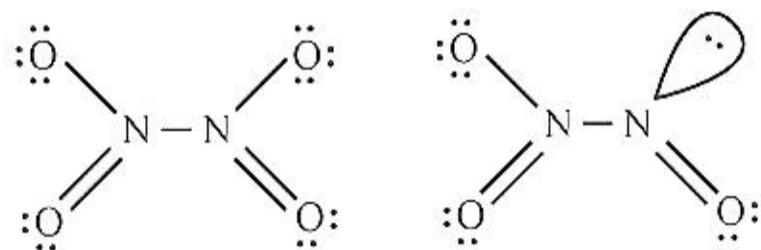
**Solution:**



**Question:** How many of the following contain N – N bond  $\text{N}_2\text{O}$ ,  $\text{N}_2\text{O}_3$ ,  $\text{N}_2\text{O}_4$ ,  $\text{N}_2\text{O}_5$ ?

**Answer:** 2.00

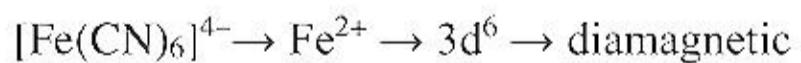
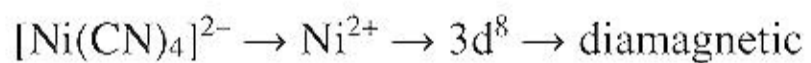
**Solution:**  $\text{N}_2\text{O}_4$  and  $\text{N}_2\text{O}_3$  has one N – N bond as shown below



**Question:** Consider the following complexes  $[\text{Fe}(\text{CN})_6]^{3-}$ ,  $[\text{Ni}(\text{CN})_4]^{2-}$  and  $[\text{Fe}(\text{CN})_6]^{4-}$ . How many complex(es) is/are paramagnetic?

**Answer:** 1.00

**Solution:**



## JEE-Main-28-06-2022-Shift-2 (Memory Based)

### MATHEMATICS

**Question:** If  $n$  arithmetic means are inserted between  $a$  and 100 then ratio of first AM and  $n^{\text{th}}$  AM is 1:7 and  $a+n=33$ . Find  $n$ .

**Options:**

(a) 21

(b) 22

(c) 23

(d) 24

**Answer:** (c)

**Solution:**

$$d = \frac{100 - a}{n + 1}$$

$$A_1 = a + \frac{100 - a}{n + 1}$$

$$A_n = a + n \left( \frac{100 - a}{n + 1} \right)$$

$$\frac{A_1}{A_n} = \frac{an + 100}{a + 100n} = \frac{1}{7}$$

$$\Rightarrow 7an + 700 = a + 100n$$

$$a + n = 33$$

$$\Rightarrow a = 33 - n$$

$$\Rightarrow 7(33 - n)n + 700 = 33 - n + 100n$$

$$\Rightarrow 231n - 7n^2 + 700 = 33 + 99n$$

$$\Rightarrow 7n^2 - 132n - 667 = 0$$

$$\Rightarrow n = 23, \frac{-29}{7}$$

$$23 \times \alpha = \frac{-667}{7}$$

$$\Rightarrow n = 23$$

**Question:** Find the area enclosed by  $x$ -axis &  $y = 3 - |x + 1| - \left| x - \frac{1}{2} \right|$ .

**Options:**

(a)  $\frac{27}{8}$

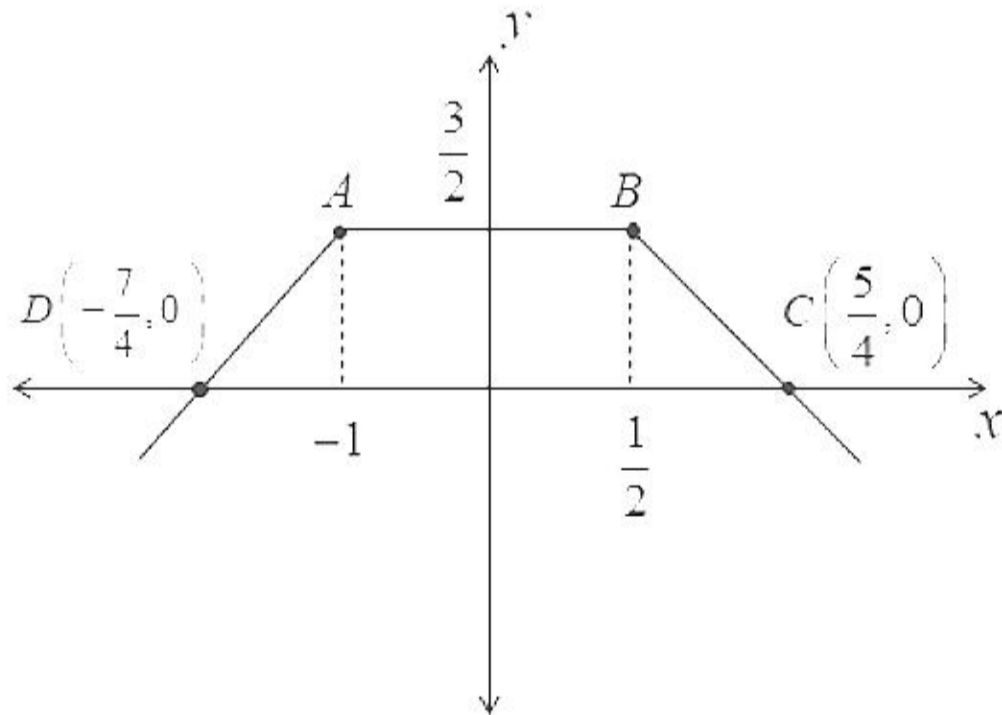
- (b)  $\frac{23}{8}$   
 (c)  $\frac{25}{8}$   
 (d)  $\frac{27}{4}$

**Answer: (d)**

**Solution:**

$$\begin{cases} 3+x+1+x-\frac{1}{2} & ; \quad x < -1 \\ 3-x-1+x-\frac{1}{2} & ; \quad -1 \leq x < \frac{1}{2} \\ 3-x-1-x+\frac{1}{2} & ; \quad x \geq \frac{1}{2} \end{cases}$$

$$\begin{cases} 2x+\frac{7}{2} & ; \quad x < -1 \\ \frac{3}{2} & ; \quad -1 \leq x < \frac{1}{2} \\ -2x+\frac{5}{2} & ; \quad x \geq \frac{1}{2} \end{cases}$$



$$\begin{aligned} \text{Required area } (ABCD) &= \frac{1}{2} \times \left( 3 + \frac{3}{2} \right) \frac{3}{2} \\ &= \frac{27}{8} \end{aligned}$$

**Question:**  $f(x) + f(x+k) = n$ ,  $I_1 = \int_0^{4nk} f(x) dx$ ,  $I_2 = \int_{-k}^{3k} f(x+k) dx$

**Options:**

(a)  $I_1 = 2I_2$

(b)  $I_1 = nI_2$

(c)

(d)

**Answer: (b)**

**Solution:**

$$f(x) + f(x+k) = n$$

$$x \rightarrow x+k$$

$$f(x+k) = f(x+2k) = n$$

$$\Rightarrow f(x) = f(x+2k)$$

Period =  $2k$

$$I_1 = \int_0^{4nk} f(x) dx = 2n \int_0^{2k} f(x) dx$$

$$I_2 = \int_{-k}^{3k} f(x+k) dx = \int_0^{4k} f(t) dt$$

$$= 2 \int_0^{2k} f(x) dx$$

$$I_1 = nI_2$$

**Question:** There are 30 candies to be distributed among 4 children  $C_1, C_2, C_3, C_4$  such that  $C_2$  gets at least 4 and at most 7 candies and  $C_3$  gets at least 2 and at most 6 candies. The number of ways to distribute it.

**Answer:**  ${}^{27}C_3 - {}^{23}C_3 - {}^{22}C_3 + {}^{18}C_3$

**Solution:**

$$x_1 + x_2 + x_3 + x_4 = 30$$

$$4 \leq x_2 \leq 7$$

$$t_2 = x_2 - 4$$

$$2 \leq x_3 \leq 6$$

$$t_3 = x_3 - 2$$

$$\Rightarrow x_1 + t_2 + t_3 + x_4 = 24$$

$${}^{24+4-1}C_{4-1} = {}^{27}C_3$$

$$x_1 + x_2 + x_3 + x_4 = 30$$

$$8 \leq x_2, 2 \leq x_3$$

$$x_1 + t_2 + t_3 + x_4 = 20$$



$${}^{20+4-1}C_{4-1} = {}^{23}C_3$$

$$x_1 + x_2 + x_3 + x_4 = 30$$

$$4 \leq x_2, 7 \leq x_3$$

$$x_1 + t_2 + t_3 + x_4 = 19$$

$${}^{19+4-1}C_{4-1} = {}^{22}C_3$$

$$x_1 + x_2 + x_3 + x_4 = 30$$

$$8 \leq x_2, 7 \leq x_3$$

$$x_1 + t_1 + t_2 + x_4 = 15$$

$${}^{15+4-1}C_{4-1} = {}^{18}C_3$$

$$\text{Required answer} = {}^{27}C_3 - {}^{23}C_3 - {}^{22}C_3 + {}^{18}C_3$$

**Question:**  $f(x)$  is a quadratic polynomial. If  $f(-2) + f(3) = 0$  and one root of equation is -1. Find the sum of roots.

**Answer:**  $\frac{11}{3}$

**Solution:**

$$f(x) = ax^2 + bx + c$$

$$f(-1) = 0 \Rightarrow a - b + c \quad \dots(1)$$

$$f(-2) + f(3) = 0$$

$$\Rightarrow (4a - 2b + c) + (9a + 3b + c) = 0$$

$$13a + b + 2c = 0 \quad \dots(2)$$

$$\text{Eq. (2)} - 2(\text{Eq. 1})$$

$$\Rightarrow 11a + 3b = 0$$

$$\Rightarrow b = \frac{-11a}{3}$$

$$\text{Sum of roots} = \frac{-b}{a} = \frac{11}{3}$$

**Question:**  $\lim_{n \rightarrow \infty} 6 \tan \left( \sum_{r=1}^n \tan^{-1} \left( \frac{1}{r^2 + 3r + 3} \right) \right)$

**Answer: 3.00**

**Solution:**

$$\text{Given, } \lim_{n \rightarrow \infty} 6 \tan \left( \sum_{r=1}^n \tan^{-1} \left( \frac{1}{r^2 + 3r + 3} \right) \right)$$

$$\lim_{n \rightarrow \infty} 6 \tan \left( \sum_{r=1}^n \tan^{-1} \left[ \frac{(r+2) - (r+1)}{1 + (r+2)(r+1)} \right] \right)$$

$$\begin{aligned}
&= \lim_{n \rightarrow \infty} 6 \tan \left( \sum_{r=1}^n \tan^{-1}(r+2) - \tan^{-1}(r+1) \right) \\
&= \lim_{n \rightarrow \infty} 6 \tan \left[ \tan^{-1}(n+2) - \tan^{-1}(2) \right] \\
&= \lim_{n \rightarrow \infty} 6 \tan \left[ \tan^{-1} \left( \frac{n}{1+2(n+2)} \right) \right] \\
&= \lim_{n \rightarrow \infty} \frac{6n}{2n+5} = 3
\end{aligned}$$

**Question:** If  $\lim_{x \rightarrow 1} \left( \frac{\sin(3x^2 - 4x + 1) - 4x + 1}{2x^3 - 7x^2 + ax + b} \right) = -2$  then  $a - b = ?$

**Answer: 11.00**

**Solution:**

$$\begin{aligned}
\lim_{x \rightarrow 1} \left[ \frac{\sin(3x^2 - 4x + 1) - x^2 + 1}{2x^3 - 7x^2 + ax + b} \right] &= -2 \\
\Rightarrow a + b &= 5 \\
\lim_{x \rightarrow 1} \frac{\cos(3x^2 - 4x + 1)(6x - 4) - 2x}{6x^2 - 14x + a} &= -2 \\
\Rightarrow a = 8, b = -3 \\
\therefore a - b &= 11
\end{aligned}$$

**Question:** For a parabola directrix:  $3x - 4y = 21$ , vertex  $(2, -1)$ . Find length of Latus Rectum.

**Answer:**  $\frac{32}{5}$

**Solution:**

Directrix:  $3x - 4y = 21$ ; vertex  $(2, -1)$

$$\begin{aligned}
a &= \left| \frac{3(2) - 4(-1) - 21}{5} \right| = \left| \frac{6 + 4 - 21}{5} \right| = \left| \frac{11}{5} \right| \\
\therefore \text{LLR} &= 4a = \frac{44}{5}
\end{aligned}$$

**Question:**  $\cot \alpha = 1$ ;  $\alpha \in \left( \pi, \frac{3\pi}{2} \right)$ ,  $\sec \beta = \frac{-5}{3}$ ,  $\beta \in \left( \frac{\pi}{2}, \pi \right)$ . Find  $\tan(\alpha + \beta)$ .

**Answer:**  $\frac{-1}{7}$

**Solution:**

$$\tan(\alpha + \beta) = \frac{\tan \alpha + \tan \beta}{1 - \tan \alpha \tan \beta} = \frac{1 - \frac{4}{3}}{1 + \frac{4}{3}} = \frac{-1}{7}$$

**Question:**  $A = \{a, b, c, d\}$ ,  $B = \{1, 2, 3, 4, 5\}$ ,  $f: A \rightarrow B$  is one-one. Find the probability that  $f(a) + 2f(b) - f(c) = f(d)$ .

**Answer:**  $\frac{1}{20}$

**Solution:**

$$f(b) = \frac{f(c) + f(d) - f(a)}{2}$$

$a$	$b$	$c$	$d$
1	3	2	5
5	1	3	4
4	2	3	5
5	1	4	3
1	3	5	2
4	2	5	3

Favourable cases = 6

Total cases =  $5 \times 4 \times 3 \times 2$

$\therefore$  Required probability =  $\frac{6}{5!} = \frac{1}{20}$

**Question:** Find the coefficient of term independent of  $x$  in  $(1 - x^2 + 3x^3) \left( \frac{5}{2}x^3 - \frac{1}{5x^2} \right)^{11}$ .

**Answer:**  $\frac{33}{200}$

**Solution:**

General term of  $\left( \frac{5}{2}x^3 - \frac{1}{5x^2} \right)^{11}$  is:

$$T_{r+1} = {}^{11}C_r \left( \frac{5}{2}x^3 \right)^{11-r} \left( -\frac{1}{5x^2} \right)^r$$

$$= {}^{11}C_r (-1)^r \cdot \frac{5^{11-2r}}{2^{11-r}} \cdot x^{33-5r}$$

Coefficient of term independent of  $x$  = Coefficient of  $x^0$  in  $\left( \frac{5}{2}x^3 - \frac{1}{5x^2} \right)^{11}$

$$\begin{aligned}
 & \text{-Coefficient of } x^{-2} \text{ in } \left(\frac{5}{2}x^3 - \frac{1}{5x^2}\right)^{11} + 3 \times \text{Coefficient of } x^{-3} \text{ in } \left(\frac{5}{2}x^3 - \frac{1}{5x^2}\right)^{11} \\
 & = 0 - {}^{11}C_7 (-1)^7 \cdot \frac{5^{-3}}{2^4} + 0 = \frac{330}{5^3 \cdot 2^4} = \frac{33}{200}
 \end{aligned}$$

**Question:** If vertex of parabola is (2, -1) and equation of its directrix is  $4x - 3y = 21$ , then the length of latus rectum is

**Answer: 8.00**

**Solution:**

Length of Latus rectum = 4 (perpendicular distance of (2, -1) from  $4x - 3y - 21 = 0$ )

$$= 4 \frac{|8 + 3 - 21|}{5} = 8$$

**Question:** Find the equation of plane passing through the points (2, -1, 0) and perpendicular to planes  $2x - 3y + z = 0$  and  $2x - y - 3z = 0$ .

**Answer: 6.00**

**Solution:**

Let DRs be  $a, b, c$

$$\text{Now } 2a - 3b + c = 0$$

$$2a - b - 3c = 0$$

$$\frac{a}{5} = \frac{b}{4} = \frac{c}{2}$$

Equation of required plane is

$$5(x-2) + 4(y+1) + 2(z-0) = 0$$

$$\Rightarrow 5x + 4y + 2z = 6$$

**Question:** In an infinite GP,  $a = n^2, r = \frac{1}{(n+1)^2}, \frac{1}{26} + \sum_{n=0}^{50} \left( S_n - \frac{2}{n+1} - n - 1 \right) = ?$ , where  $S_n$  is

sum of given GP.

**Answer: 41652.00**

**Solution:**

$$a = n^2, r = \frac{1}{(n+1)^2}; s_n = \frac{a}{1-r} = \frac{n(n+1)^2}{(n+2)}$$

$$\Rightarrow \frac{1}{26} + \sum_{n=0}^{50} \left[ \frac{n(n+1)^2}{(n+2)} + \left( \frac{2}{n+1} \right) - (n+1) \right]$$

$$\Rightarrow \frac{1}{26} + \sum_{n=0}^{50} \left[ (n^2 - n - 1) + \left( \frac{n}{n+2} \right) + \left( \frac{2}{n+1} \right) \right]$$

$$\begin{aligned} &\Rightarrow \frac{1}{26} + \sum_{n=0}^{50} \left[ n^2 - n - 2 \left( \frac{1}{n+2} - \frac{1}{n+1} \right) \right] \\ &\Rightarrow \frac{1}{26} + \left[ \frac{50 \cdot 51 \cdot 101}{6} - \frac{50 \cdot 51}{2} - 2 \sum_{n=0}^{50} \left( \frac{1}{n+2} - \frac{1}{n+1} \right) \right] \\ &\Rightarrow \frac{1}{26} + \left[ 41650 - 2 \left( \frac{1}{52} - 1 \right) \right] = 41652 \end{aligned}$$