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## Silverzone Foundation

New Delhi, India

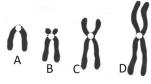
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## **CLASS 9th SAMPLE QUESTIONS**

The Actual Question Paper Contains 50 Questions. The Duration of the Test Paper is 60 Minutes.

Which one of the following is the acrocentric chromosome?



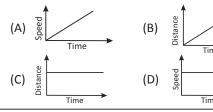
(A) A

(B) B

(C) C

(D) D

2. Which one of the following graphs shows uniform acceleration?



- 3. Choose the isobar from the following.
  - (A)  ${}^{6}C_{12}$  and  ${}^{6}C_{14}$
- (B)  $^{92}U_{235}$  and  $^{92}C_{238}$
- (C) <sup>1</sup>H<sub>1</sub> and <sup>1</sup>C<sub>2</sub>
- (D)  ${}^{18}\text{Ar}_{40}$  and  ${}^{20}\text{Ca}_{40}$
- 4. Humans are classified according to the following hierarchy:

Animalia  $\to$  Chordata  $\to$  Mammalia  $\to$  Primates  $\to$  Hominidae  $\to$  Homo  $\to$  Sapiens

- Which category is represented by Primates in the above hierarchy?
- (A) Order
- (B) Genus
- (C) Kingdom
- (D) Family
- 5. What is the number of water molecules contained in a drop of water weighing 0.12 g?
  - (A)  $2.007 \times 10^{21}$
- (B)  $4.014 \times 10^{21}$
- (C)  $2.007 \times 10^{22}$
- (D)  $4.014 \times 10^{22}$
- 6. Calculate the number of molecules of chloroform (CHCl<sub>2</sub>)weighing 0.0239 g (H =1, C=12, Cl=35.5).
  - (A)  $0.2046 \times 10^{17}$
- (B)  $1.2046 \times 10^{20}$
- (C)  $2.2046 \times 10^{22}$
- (D)  $3.1046 \times 10^{21}$

- 7. Which of the following is not a characteristic of pure substances?
  - (A) A pure substance contains only one type of particles.
  - (B) A pure substance has definite composition.
  - (C) A pure substance may be homogeneous or heterogeneous.
  - (D) A pure substance has definite melting and boiling point.
- 8. Fill in the blanks with the correct combination of terms.

The disease which lasts for <u>P</u> periods of time is called Q disease. <u>R</u> and <u>S</u> are its two examples.

- A. P-very long, Q-acute, R-common cold, S-dysentery
- B. P-very short, Q-acute, R-common cold, S-dysentery
- C. P-very short, Q-acute, R-cancer, S-dysentery
- D. P-very long, Q-chronic, R-common cold, S-dysentery
- 9. Which one of the following shows that cathode rays are negatively charged particles?
  - (A) Cathode rays produce greenish light on striking the wall of discharge tube
  - (B) Cathode rays cast shadows of the objects placed in their path
  - (C) Cathode rays move the blades of a paddle wheel placed in their path
  - (D) Cathode rays are deflected towards the positive plate of an electric field
- 10. Classify the following materials into elements, compounds and mixtures:

Methane, granite, blood, sodium, silver, iron, sugar

(A) Elements: Sodium and silver

Compounds: Methane, Granite, sugar

Mixtures: Iron, blood

(B) Elements: Sodium, silver, iron,

Compounds: Methane, sugar Mixtures: Granite, blood

(C) **Elements:** Sodium, silver, iron, **Compounds:** Methane, blood

Mixtures: Granite, sugar

(D) Elements: Methane, silver, iron, Compounds: Sodium, sugar Mixtures: Granite, blood

ANSWERS											
1. (B)	2. (A)	3. (D)	4. (A)	5. (B)	6. (B)	7. (C)	8. (B)	9. (D)	10. (B)		