## Section A: Q. 1 - Q. 10 Carry ONE mark each.

Q. $1 \quad$ Which one of the following is a simple tissue system in plants?
(A) Epidermis
(B) Parenchyma
(C) Phloem
(D) Xylem
Q. 2 In DNA replication, the Okazaki fragments are joined by
(A) DNA helicase
(B) DNA ligase
(C) DNA polymerase
(D) DNA primase
Q. 3 The most abundant type of RNA in a metabolically active mammalian cell is
(A) mRNA
(B) rRNA
(C) snoRNA
(D) tRNA
Q. 4 Which organelle in a eukaryotic cell is the site of electron transport chain?
(A) Endoplasmic reticulum
(B) Golgi apparatus
(C) Mitochondrion
(D) Peroxisome
Q. $5 \quad$ RNA is a polymer of
(A) glycosides
(B) ribonucleosides
(C) ribonucleotides
(D) riboses
Q. 6 Which one of the following is present in a bacterial cell?
(A) 28 S rRNA
(B) 70S ribosome
(C) Chitinous cell wall
(D) Histones
Q. $7 \quad$ Which color of light excites a natural GFP to emit green fluorescence?
(A) Blue
(B) Green
(C) Infrared
(D) Red
Q. 8 Which one of the following hormones promotes fruit ripening?
(A) Abscisic acid
(B) Auxin
(C) Ethylene
(D) Gibberellin
Q. $9 \quad$ Which one of the following has a catalytic RNA?
(A) Ribonuclease H
(B) Ribozyme
(C) RNA polymerase I
(D) RNA polymerase II
Q. 10 The number of significant figures in a reported measurement of 0.00361 is
(A) 3
(B) 4
(C) 5
(D) 6

## Section A: Q. 11 - Q. 30 Carry TWO marks each.

Q. 11 Match the terminology in Group I with the stimulus in Group II that generates growth response of plants

## Group I

P. Gravitropism
Q. Phototropism
R. Thigmotropism
S. Chemotropism

Group II

1. Light
2. Touch
3. Chemical
4. Gravity
(A) $\mathrm{P}-3, \mathrm{Q}-4, \mathrm{R}-2, \mathrm{~S}-1$
(B) $\mathrm{P}-2, \mathrm{Q}-1, \mathrm{R}-3, \mathrm{~S}-4$
(C) $\mathrm{P}-4, \mathrm{Q}-1, \mathrm{R}-2, \mathrm{~S}-3$
(D) $\mathrm{P}-4, \mathrm{Q}-2, \mathrm{R}-1, \mathrm{~S}-3$
Q. 12 The correct hierarchy of taxa in the Linnaean classification of eukaryotes is
(A) kingdom, class, phylum, order, family, genus
(B) kingdom, order, class, phylum, family, genus
(C) kingdom, phylum, order, family, class, genus
(D) kingdom, phylum, class, order, family, genus
Q. 13 Which one of the following statements about polyploidy is correct?
(A) Autopolyploids are derived from a single species
(B) Autopolyploids are derived from two different species
(C) Allopolyploids are derived from a single species
(D) Allopolyploids are not fertile when mated with each other
Q. 14 Which one of the following hormones is a tyrosine derivative?
(A) Epinephrine
(B) Estradiol
(C) Progesterone
(D) Testosterone
Q. 15 Which one of the following immunoglobulins crosses the human placenta?
(A) $\operatorname{Ig} \mathrm{A}$
(B) $\operatorname{IgE}$
(C) IgG
(D) $\operatorname{IgM}$
Q. 16 Determine the correctness or otherwise of the following Assertion [a] and the Reason [r].

Assertion [a]: The resolving power of a transmission electron microscope is higher than that of the light microscope.

Reason [r]: The wavelength of electrons is shorter than that of visible light.
(A) Both [a] and [r] are true and [r] is the correct reason for [a]
(B) Both [a] and [r] are true but [r] is not the correct reason for [a]
(C) Both [a] and [r] are false
(D) [a] is true but [r] is false
Q. 17 Match the morphology in Group I with the corresponding microorganism in Group II

| Group I | Group II |
| :--- | :--- |
| P. Coccus | 1. Treponema |
| Q. Rod | 2. Bacillus |
| R. Comma | 3. Neisseria |
| S. Spiral | 4. Vibrio |

(A) $\mathrm{P}-3, \mathrm{Q}-2, \mathrm{R}-4, \mathrm{~S}-1$
(B) $\mathrm{P}-4, \mathrm{Q}-1, \mathrm{R}-3, \mathrm{~S}-2$
(C) $\mathrm{P}-2, \mathrm{Q}-4, \mathrm{R}-1, \mathrm{~S}-3$
(D) $\mathrm{P}-1, \mathrm{Q}-2, \mathrm{R}-3, \mathrm{~S}-4$
Q. 18 Which one of the following genetic crosses and their results indicates cytoplasmic inheritance?
(A) Wild-type male $\times$ mutant female $\rightarrow 100 \%$ progeny are mutant
(B) Wild-type male $\times$ mutant female $\rightarrow 25 \%$ progeny are wild-type
(C) Mutant male $\times$ wild-type female $\rightarrow 50 \%$ progeny are mutant
(D) Mutant male $\times$ wild-type female $\rightarrow 75 \%$ progeny are wild-type
Q. 19 Which of the following is NOT a characteristic morphological feature of apoptotic cells?
(A) Disassembly of nuclear envelope
(B) DNA fragmentation
(C) Increased cell size
(D) Membrane blebbing
Q. 20 Competition between two populations in an ecosystem is
(A) beneficial (+) to both the populations
(B) deleterious (-) to both the populations
(C) beneficial (+) to one population, but deleterious ( - ) to the other population
(D) beneficial $(+)$ to one population, but no effect $(0)$ on the other population
Q. 21 Adenine constitutes 0.16 mole fraction in a given single-stranded DNA. What is the mole fraction of uracil in the resultant RNA, if this entire DNA fragment is transcribed?
(A) 0.16
(B) 0.32
(C) 0.34
(D) 0.68
Q. 22 Which one of the following is NOT used as a component in subunit vaccines?
(A) Capsular polysaccharide
(B) Inactivated exotoxin
(C) Inactivated virus
(D) Viral glycoprotein
Q. 23 Metabolic acidosis is associated with decreased plasma level of
(A) bicarbonate
(B) lactate
(C) oxygen
(D) urea
Q. 24 Genes in two species that are derived from the same ancestral gene in their most recent common ancestor are called
(A) analogs
(B) heterologs
(C) orthologs
(D) paralogs
Q. 25 An object is placed 15 cm in front of a convex mirror, which has a radius of curvature 30 cm . Which one of the following is true of the image formed?
(A) Real and inverted
(B) Real and upright
(C) Virtual and inverted
(D) Virtual and upright
Q. 26 If a variable $Z$ shows a standard normal distribution, then the percent probability that

$$
0 \leq z^{2} \leq 1
$$

is $\qquad$ (rounded off to the nearest integer).
(A) 34
(B) 68
(C) 95
(D) 99
Q. 27 In chick embryo, the ectoderm generates
(A) alveolar cells
(B) germ cells
(C) neurons
(D) red blood cells
Q. 28 The boiling points of Iodomethane, Dibromomethane, Bromomethane, Chloromethane follow the order
(A) Bromomethane $>$ Dibromomethane $>$ Iodomethane $>$ Chloromethane
(B) Bromomethane $>$ Iodomethane $>$ Chloromethane $>$ Dibromomethane
(C) Dibromomethane $>$ Iodomethane $>$ Bromomethane $>$ Chloromethane
(D) Iodomethane $>$ Bromomethane $>$ Chloromethane $>$ Dibromomethane
Q. 29 Chromosome duplication during the cell cycle occurs in
(A) $\mathrm{G}_{1}$ phase
(B) $\mathrm{G}_{2}$ phase
(C) M phase
(D) S phase
Q. 30 Ionic character of the covalent bonds in the compounds $\mathrm{Cl}_{2}, \mathrm{HCl}, \mathrm{NaCl}, \mathrm{NaF}$ follows the order
(A) $\mathrm{Cl}_{2}>\mathrm{NaCl}>\mathrm{HCl}>\mathrm{NaF}$
(B) $\mathrm{HCl}>\mathrm{Cl}_{2}>\mathrm{NaF}>\mathrm{NaCl}$
(C) $\mathrm{HCl}>\mathrm{NaCl}>\mathrm{NaF}>\mathrm{Cl}_{2}$
(D) $\mathrm{NaF}>\mathrm{NaCl}>\mathrm{HCl}>\mathrm{Cl}_{2}$

## Section B: Q. 31 - Q. 40 Carry TWO marks each.

Q. 31 Which of the following is/are lateral meristems?
(A) Cork cambium
(B) Procambium
(C) Protoderm
(D) Vascular cambium
Q. 32 Which of the following statement(s) about Golden Rice is/are correct?
(A) Consumption of it increases vitamin A levels
(B) Consumption of it increases vitamin D levels
(C) Consumption of it increases vitamin K levels
(D) It is a transgenic crop containing $\beta$-carotene
Q. 33 Which of the following statement(s) about eukaryotic DNA topoisomerase is/are correct?
(A) Topoisomerase I creates transient single-strand breaks
(B) Topoisomerase I creates transient double-strand breaks
(C) Topoisomerase II creates transient single-strand breaks
(D) Topoisomerase II creates transient double-strand breaks
Q. 34 Which of the following method(s) is/are used to estimate protein concentration?
(A) Anthrone
(B) Biuret
(C) Bradford
(D) Lowry
Q. 35 Which of the following is/are example(s) of a lotic ecosystem?
(A) Lake
(B) Pond
(C) River
(D) Stream
Q. 36 Which of the following statement(s) about the effect of genetic drift is/are correct?
(A) It can cause changes in the frequency of alleles at random
(B) It is a mechanism of evolution
(C) It can lead to loss of genetic variation within small populations
(D) It is significant in large populations
Q. 37 Which of the following technique(s) can be used to determine the threedimensional structure of an organic compound?
(A) Mass spectrometry
(B) NMR spectroscopy
(C) UV-visible spectroscopy
(D) X-ray crystallography
Q. 38 Which of the following entity(ies) is/are found inside the intact nucleus of eukaryotic cells?
(A) Centrosome
(B) Lysosome
(C) Nucleolus
(D) Nucleosome
Q. 39 Which of the following is/are trace element(s)?
(A) Mn
(B) P
(C) S
(D) Zn
Q. 40 Which of the following is/are true about Retrovirus?
(A) It contains double-stranded RNA genome
(B) It can cause cancer
(C) It contains reverse transcriptase
(D) It contains double-stranded DNA genome

## Section C: Q. 41 - Q. 50 Carry ONE mark each.

Q. 41 A wooden plant accumulates $10 \mathrm{mg} \mathrm{kg}^{-1}$ of ${ }^{14} \mathrm{C}$ during its life span. A fossil of this plant was discovered and contains $2.5 \mathrm{mg} \mathrm{kg}^{-1}$ of ${ }^{14} \mathrm{C}$. The age of this fossil at the time of discovery is $\qquad$ years (rounded off to the nearest integer).
(Use 5730 years as half-life of ${ }^{14} \mathrm{C}$ )
Q. 42 A cylinder contains 50 L of an ideal gas at a pressure of 50 atm . Assuming that the temperature remains unchanged, the volume of the gas at 1 atm is $\qquad$ $L$ (rounded off to the nearest integer).
Q. 43 One molecule of the protein myoglobin contains one atom of iron. A myoglobin sample was found to contain $0.34 \%$ iron. The molecular weight of myoglobin is $\qquad$ $g \mathrm{~mol}^{-1}$ (rounded off to the nearest integer).
(Use $55.9 \mathrm{~g} \mathrm{~mol}^{-1}$ as atomic mass of iron)
Q. 44 The wavelength of visible light for the green color is 600 nm . The energy of photons of this color is $\qquad$ eV (rounded off to one decimal place).
(Planck's constant $=6.63 \times 10^{-34} \mathrm{Js}, 1 \mathrm{eV}=1.6 \times 10^{-19} \mathrm{~J}$, speed of light $=3 \times 10^{8} \mathrm{~ms}^{-1}$ )
Q. 45 A ball dropped from a bridge hits the surface of the water in 3 s . The height of the bridge, ignoring air resistance, is $\qquad$ $m$ (rounded off to one decimal place).
(Use $g=9.8 \mathrm{~ms}^{-2}$ )
Q. 46 For a given square, if the area of its incircle is $100 \mathrm{~cm}^{2}$, then the area of its circumcircle is $\qquad$ $\mathrm{cm}^{2}$ (rounded off to the nearest integer).
Q. 47 The number of peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum of methoxymethane $\left(\mathrm{CH}_{3} \mathrm{OCH}_{3}\right)$ is $\qquad$ .
Q. 48 The amount of agarose required to prepare 250 mL of $0.8 \%$ agarose gel is $\qquad$ grams (rounded off to the nearest integer).
Q. 49 Three genes $x, y$, and $z$ are located on a chromosome in a linear order. If the recombination frequencies between $x$ and $y$ is 0.15 , and between $y$ and $z$ is 0.10 , then the expected frequency of double crossovers is $\qquad$ (rounded off to three decimal places).
Q. $50 \quad$ A bacterial cell suspension contains $2 \times 10^{5}$ cells mL ${ }^{-1}$. The volume of this suspension required to obtain $1.4 \times 10^{6}$ cells is $\qquad$ $m L$ (rounded off to the nearest integer).

## Section C: Q. 51 - Q. 60 Carry TWO marks each.

Q. 51 The data provided in the table were obtained from the following reaction, carried out at 273 K .

$$
A+B \rightarrow C
$$

| Initial concentration <br> of $[A] \mathrm{mol} \mathrm{L}^{-1}$ | Initial concentration <br> of $[B] \mathrm{mol} L^{-1}$ | Initial rate of formation <br> of $[C] \mathrm{mol} \mathrm{L}^{-1} \mathrm{~s}^{-1}$ |
| :---: | :---: | :---: |
| 0.2 | 0.2 | 0.3 |
| 0.4 | 0.2 | 0.6 |
| 0.4 | 0.4 | 2.4 |

The order of the reaction with respect to $A$ is $\qquad$ .
Q. 52 Ammonia is synthesized in the Haber process in the following reaction.

$$
\mathrm{N}_{2}(\mathrm{~g})+3 \mathrm{H}_{2}(\mathrm{~g}) \rightarrow 2 \mathrm{NH}_{3}(g)
$$

The temperature above which the reaction becomes spontaneous is $\qquad$ $K$ (rounded off to one decimal place).
$\left(\Delta H^{0}=-92.2 k J, \quad \Delta S^{0}=-199 J K^{-1}\right)$
Q. 53 In the given molecule,

the number of chiral centers is $\qquad$ .
Q. $54 \quad$ Two resistors $2 \Omega$ and $4 \Omega$ are combined in parallel. If this combination is connected to a battery of 16 V , the maximum current that can be drawn from the battery is $\qquad$ $A$ (rounded off to the nearest integer).
Q. 55 A box of mass 20 kg is pulled at constant speed across a floor by a rope.

The rope makes an angle of $45^{\circ}$ with the horizontal. Assuming that friction is negligible, the work done in pulling the box by a distance of 20 m is $\qquad$ $J$ (rounded off to the nearest integer).
(Use $g=9.8 \mathrm{~ms}^{-2}$ )
Q. 56 Consider an enzyme that follows simple Michaelis-Menten kinetics, and has a $K_{M}$ of $5 \mu M$. The initial velocity of the reaction will be $10 \%$ of the maximum velocity at a substrate concentration of $\qquad$ $\mu M$ (rounded off to two decimal places).
Q. 57 The value of $\lim _{x \rightarrow 3} \frac{x^{2}-9}{x^{2}-4 x+3}$ is $\qquad$ (rounded off to the nearest integer).
Q. 58 A population of 1000 plants are in Hardy-Weinberg equilibrium. Two alleles $R$ and $r$ determine a particular trait in this population. If the number of plants with $R R$ genotype is $640, R r$ genotype is 320 and $r r$ genotype is 40, the frequency of $r$ allele (in percentage) in this population is $\qquad$ (rounded off to the nearest integer).
Q. 59 If a fair coin is tossed two times, the probability that the first or the second toss will be heads is $\qquad$ (rounded off to two decimal places).
Q. 60 The restriction map of a circular plasmid is shown below, along with the indicated distances between the restriction sites.


The plasmid was completely digested with EcoRI, and XhoI. The products were analysed by agarose gel electrophoresis followed by ethidium bromide staining. The number of bands that will be visible in the gel when exposed to UV light is $\qquad$ .

